

TDLS of methane and its applications to study of methane emissions from natural structures to the atmosphere

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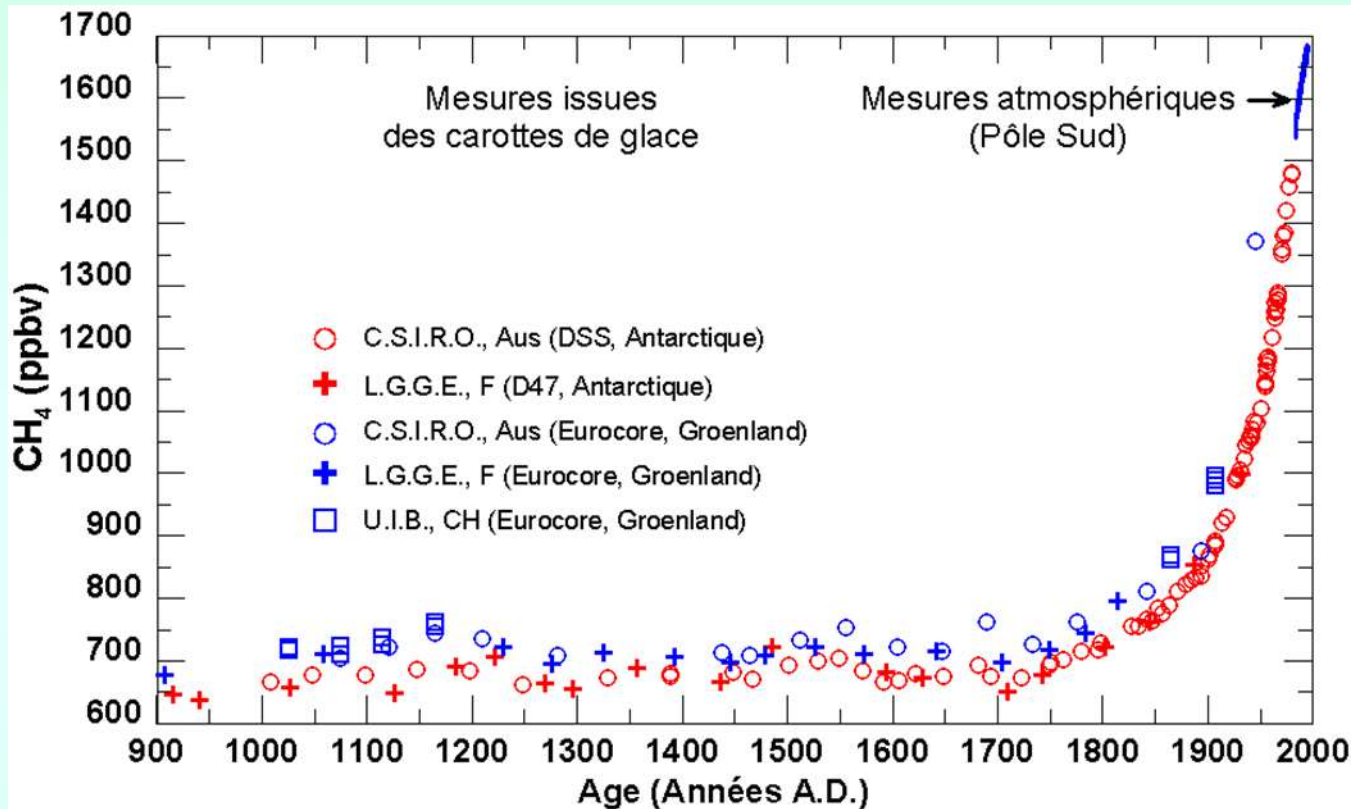
yupon@iao.ru

- 1. Motivation**
- 2. PA TDLS spectrometers**
- 3. Experimental study of CH₄ spectra**
 - line positions and intensities**
 - line broadening, shifting and interference**
- 4. TDLS applications to study of methane emissions
(hydrocarbons, coal, biological systems)**
- 5. Summary**

1.Motivation

- 1. Spectroscopic databases progress**
- 2. Validation of spectral line shape models**
- 3. Retrieval of methane profiles in Earth atmosphere from satellite data**
- 4. Local and remote detection of methane emissions with TDL techniques**

Changes of atmospheric methane total amount from 900 to 2000 year.



CH₄ increase
0,6 → 1,7 ppm

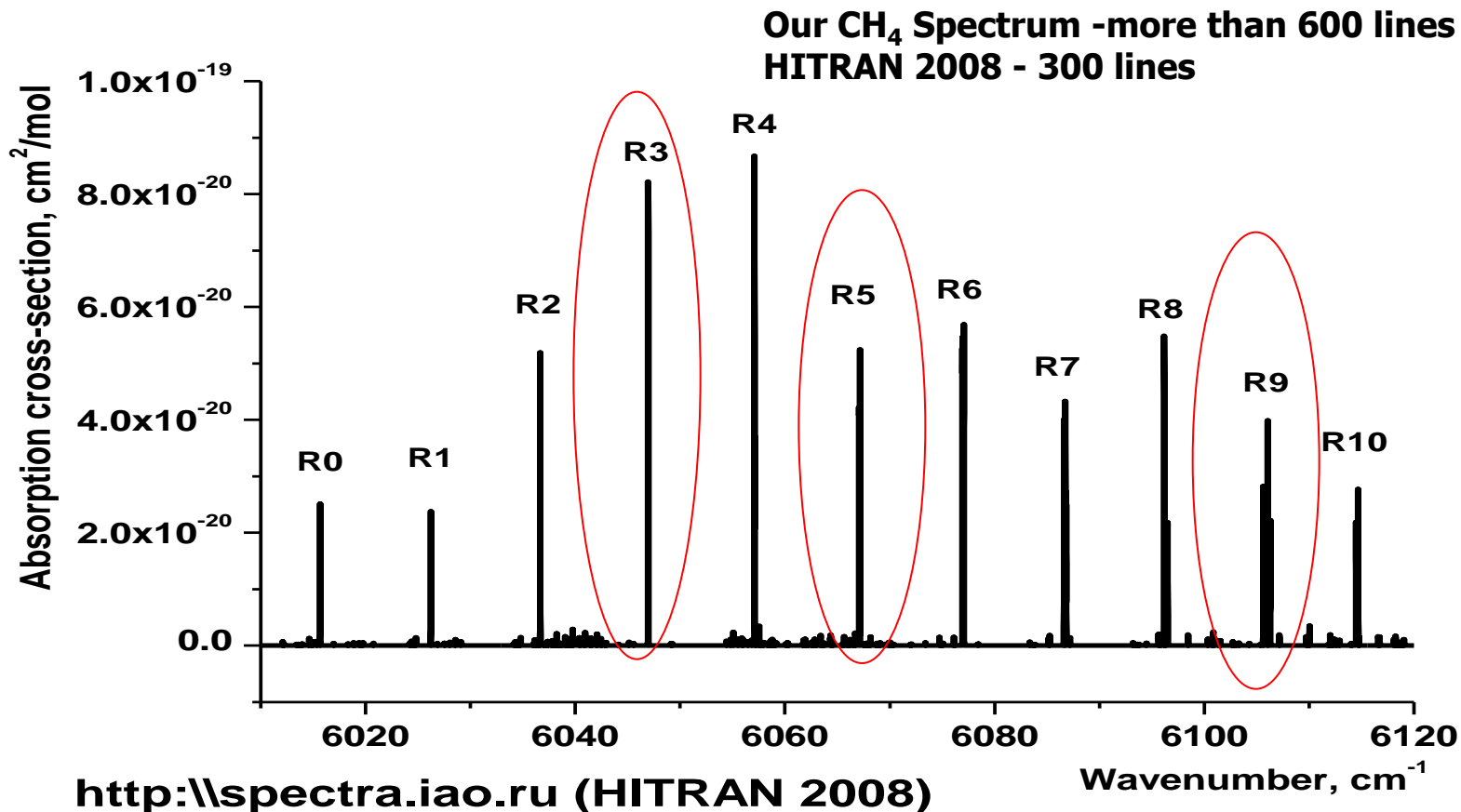
Change in radiative heating, W/m²

Atmospheric gas	Atmosphere	Surface
CH ₄ doubling	0,43	0,21
CO ₂ doubling	1,14	1,56

Contribution of CH₄
doubling = 38% of
CO₂ **doubling**

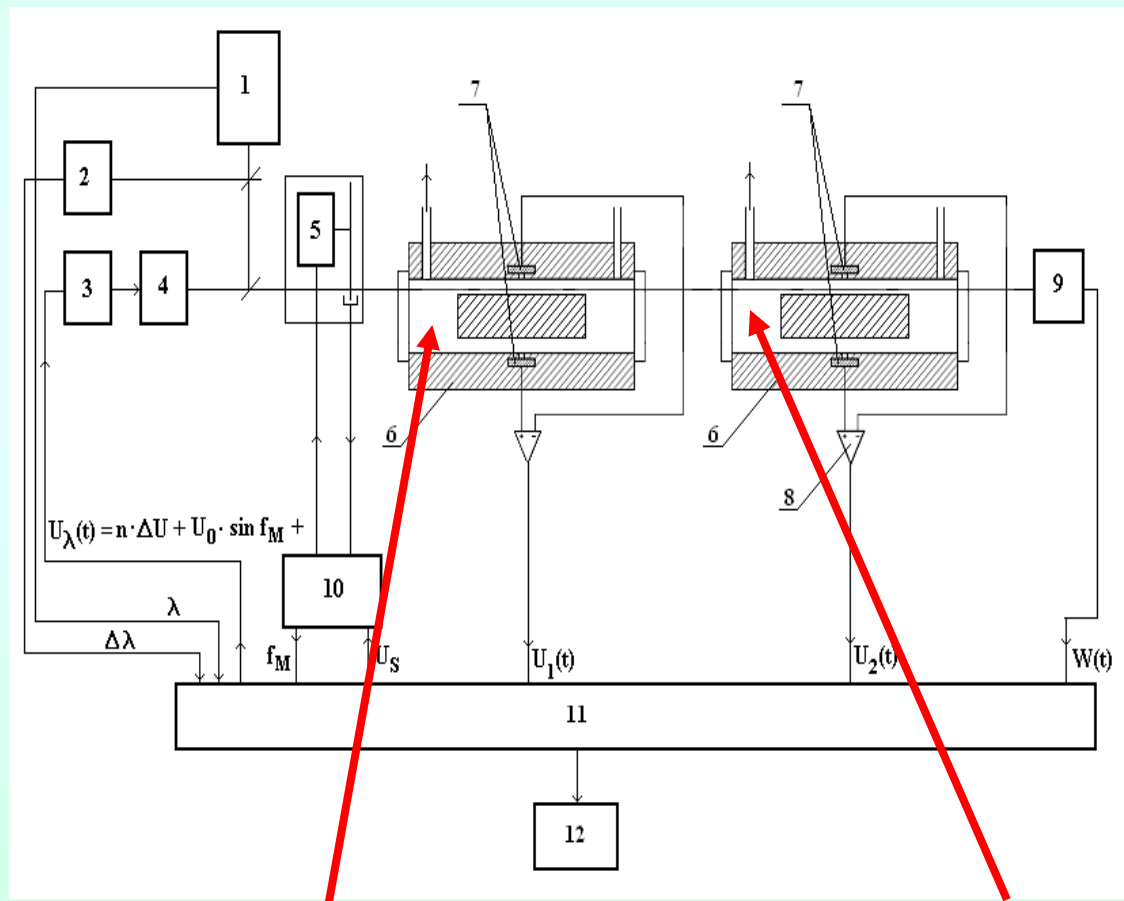
PA TDL spectroscopy of CH₄ spectral line manifolds near 1.65 μm

Kapitanov V.A., Ponomarev Yu.N., Tyryshkin I.S. and Rostov A.P.
// *Spectrochimica Acta Part A*. 2007. V. 66, N 4-5, P. 811-818.



Two-channel diode laser OAD spectrometer

Kapitanov V.A., Ponomarev Yu.N., Tyryshkin I.S and Rostov A.P.:
Spectrochimica Acta Part A, 66A, 4-5, 811-818 (2007)



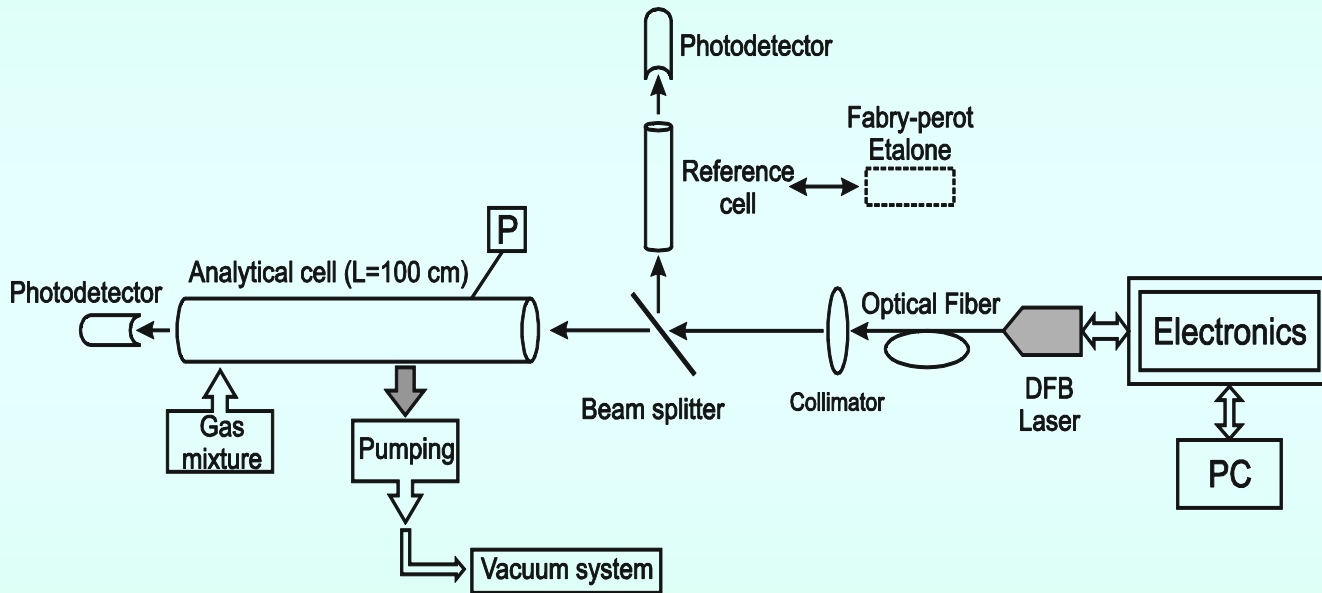
laser:
 $\Delta\nu - 6060-6250\text{cm}^{-1}$
 $d\nu - 2,5-3 \text{ cm}^{-1}$
W - 3-7 mBT
OAD:
 $\Delta = (U_m^2)^{1/2}/R =$
 $4 \cdot 10^{-9} \text{ cm}^{-1}\text{BT}$

CH₄, low pressure

Mixture, high pressure

Block-diagram of the two channel DL spectrometer

*K.Y. Osipov, A.E. Protasevich, V.A. Kapitanov, Y.Y. Ponurovskii:
Appl.Phys. B. 2012. V. 106, No 3. P. 725-732*



laser: $\Delta\nu$ - 6066-6068 cm^{-1} , W - 15 mBT

Fitting procedure

1. Spectrum by spectrum fitting
2. Multispectrum fitting

Multispectrum fitting procedure applied for all spectra simultaneously.

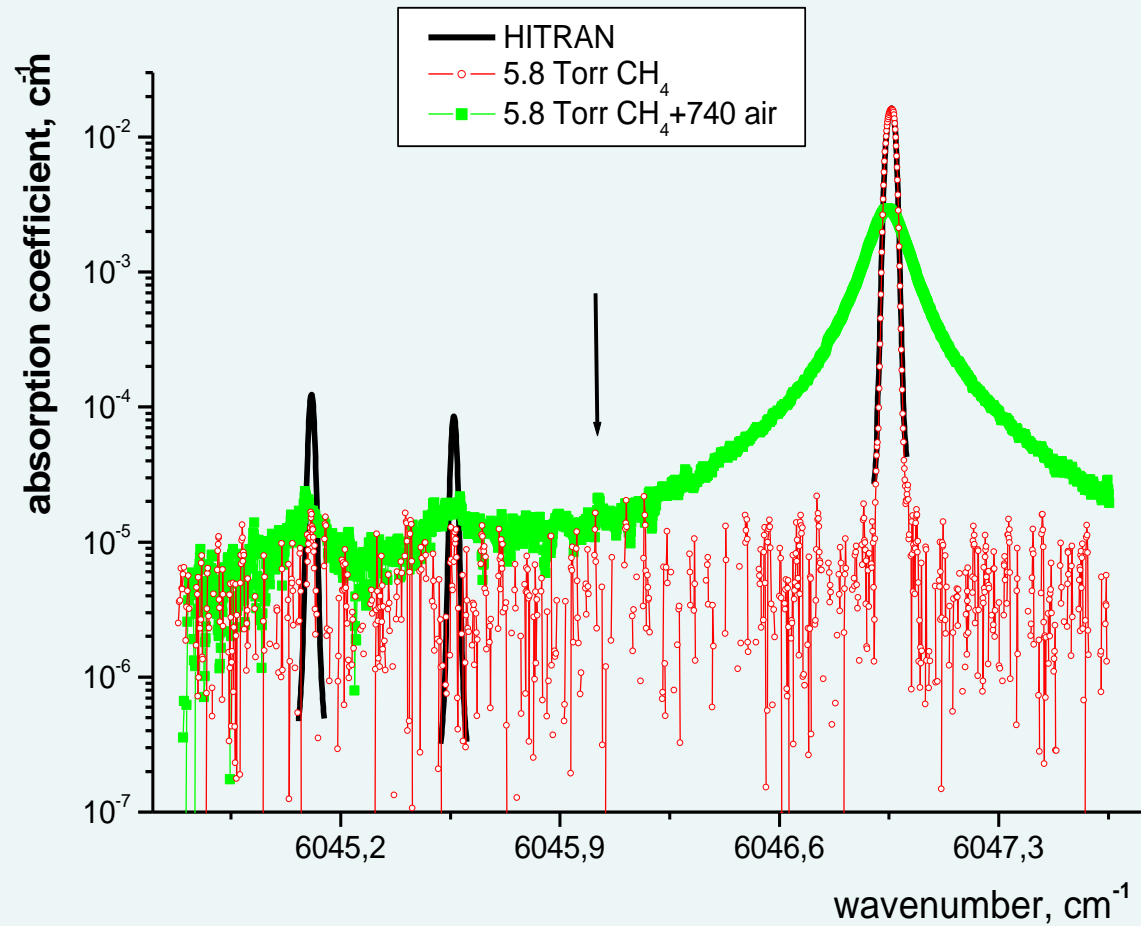
[Pine A. S. Line mixing sum rules for the analysis of multiplet spectra // JQSRT 1997. Vol. 57. No 2. pp. 145-155.]

$$K(\omega) = \frac{1}{\sqrt{\pi}} \sum_m \frac{\xi_m \operatorname{Re} P(x'_m, y_m, \zeta_m) + \eta_m \operatorname{Im} P(x'_m, y_m, \zeta_m)}{\sigma_m}$$

$$\sigma_m = \omega_m^0 \sqrt{\frac{2k_B T}{Mc^2}} \quad \text{- Doppler halfwidth of line numbered } m \text{ at } e^{-1} \text{ of line maximum, } \text{cm}^{-1}$$

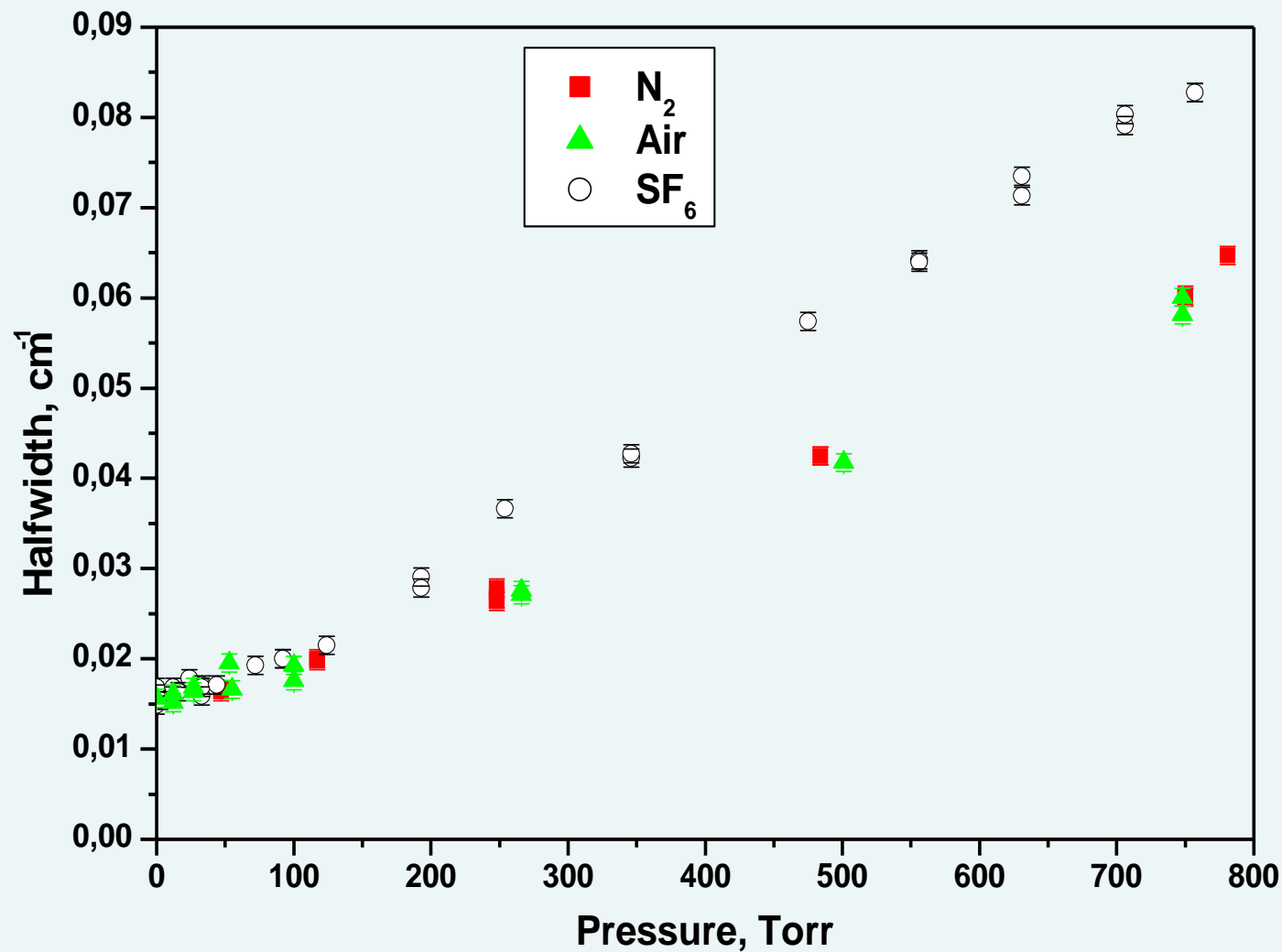
$$\omega_m^0 \quad \text{- line center at zero pressure, } \text{cm}^{-1}$$

Triplet R(3) of $2\nu_3$ methane absorption band, broadened by air pressure

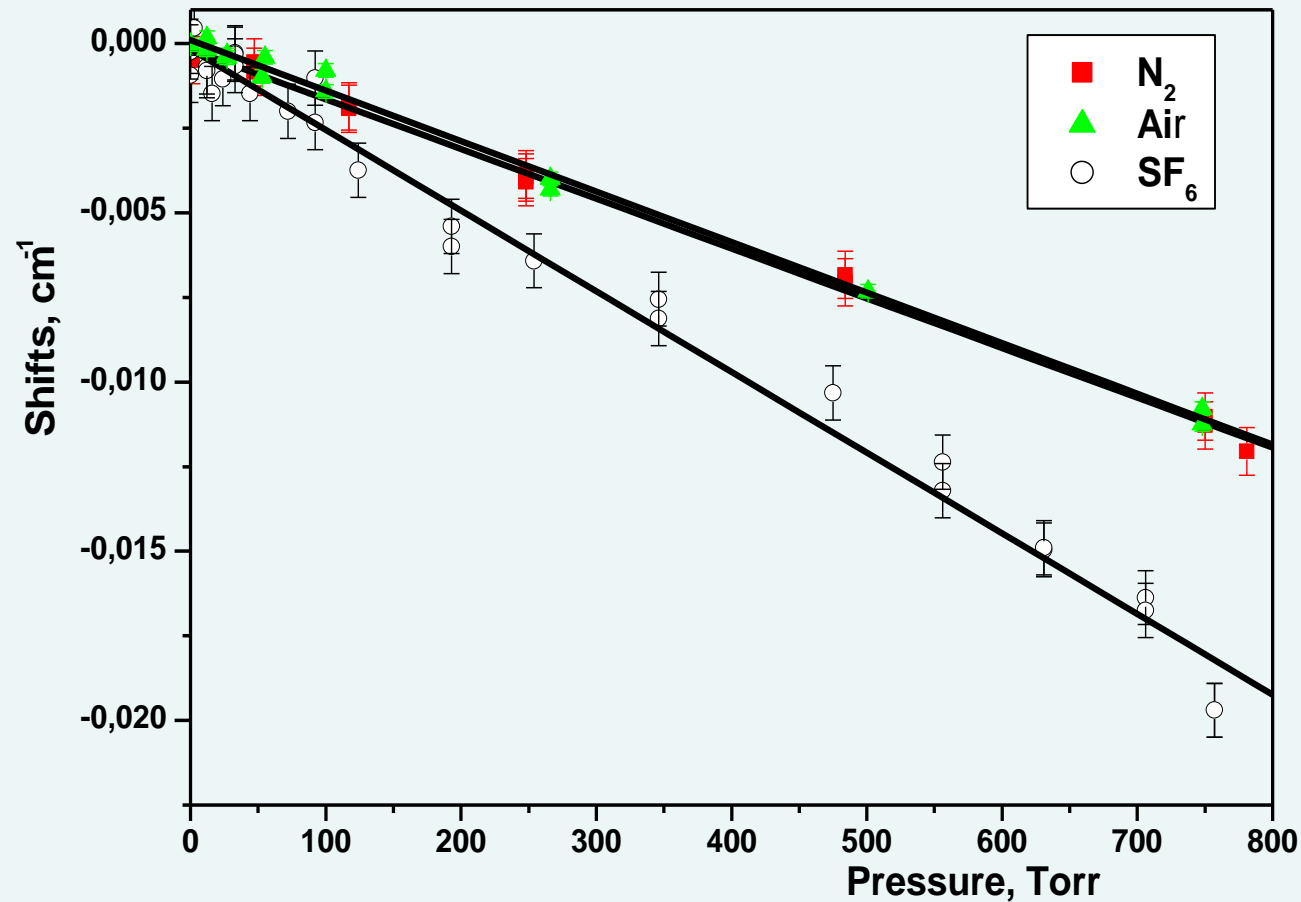


V. Zeninari, B. Parvitte D. Courtois V.A. Kapitanov, Yu.N. Ponomarev:
Appl. Phys. B 72, 953–959 (2001)

Halfwidth of the triplet R(3) vs CH₄ - N₂, -air and - SF₆ mixture pressure



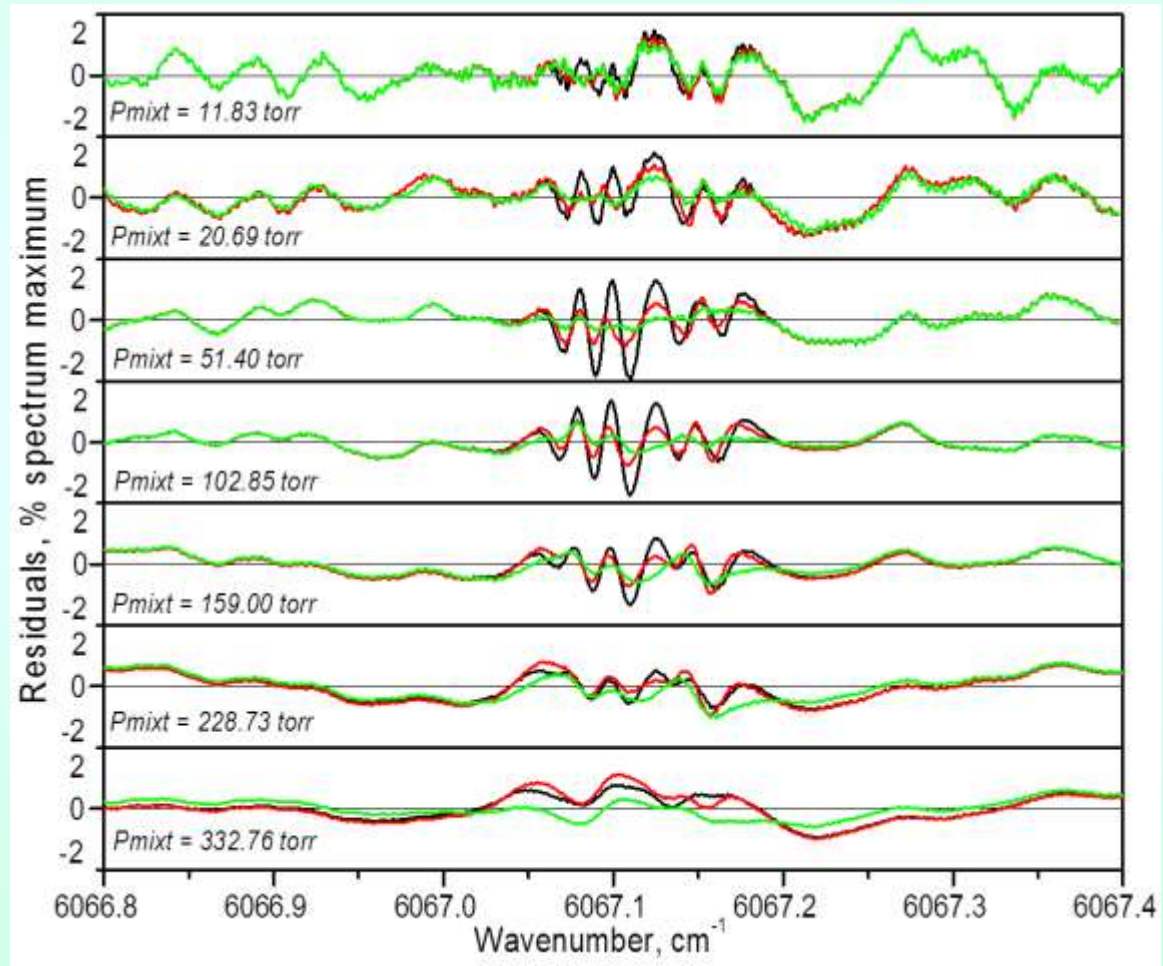
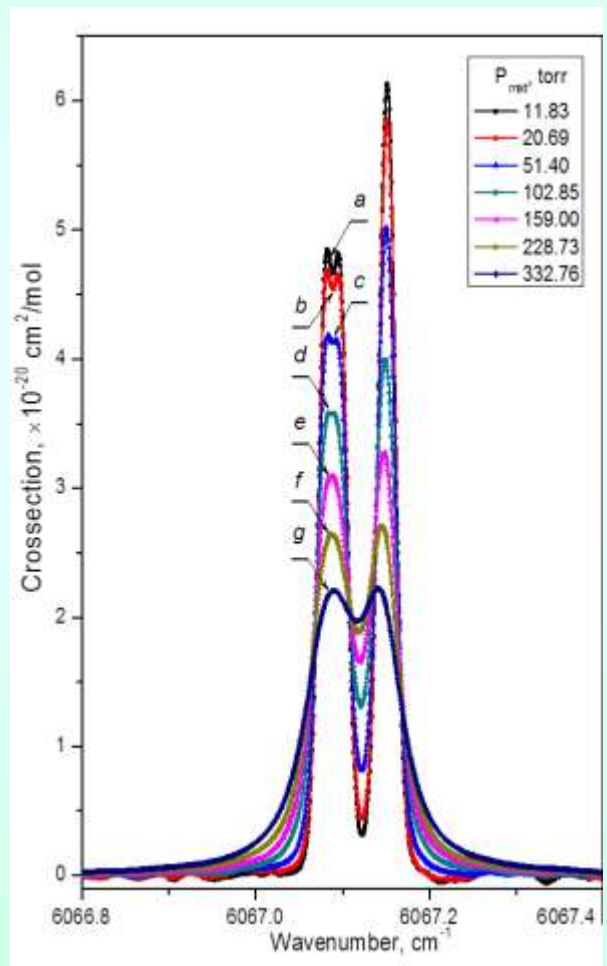
Mass center shift of the triplet R(3) vs CH₄ - N₂, -air and – SF₆ mixture pressure



Kapitanov V.A., Ponomarev Yu.N., Tyryshkin I.S., Bykov A.D., Savel'ev V.N.:
Atmos.Oceanic Optics, 2008, Vol.21, No.7, 493-499

Methane R5 spectra broadened by N₂ pressure

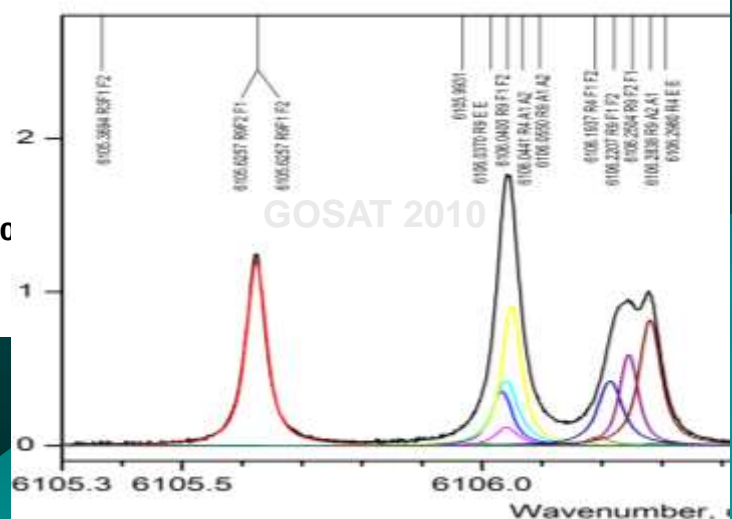
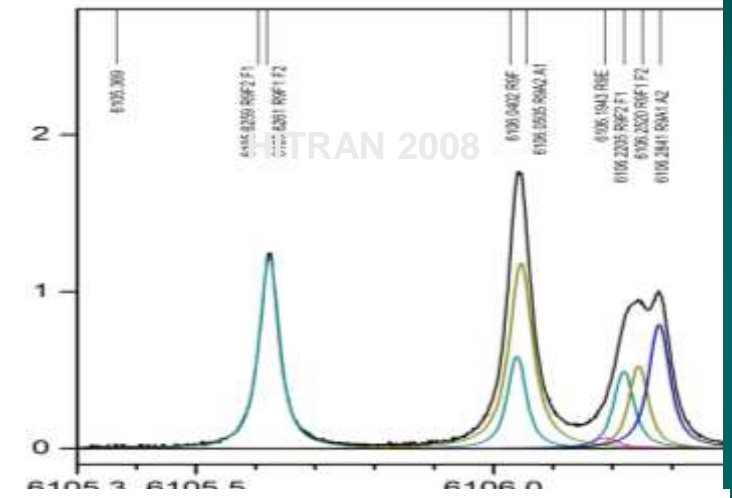
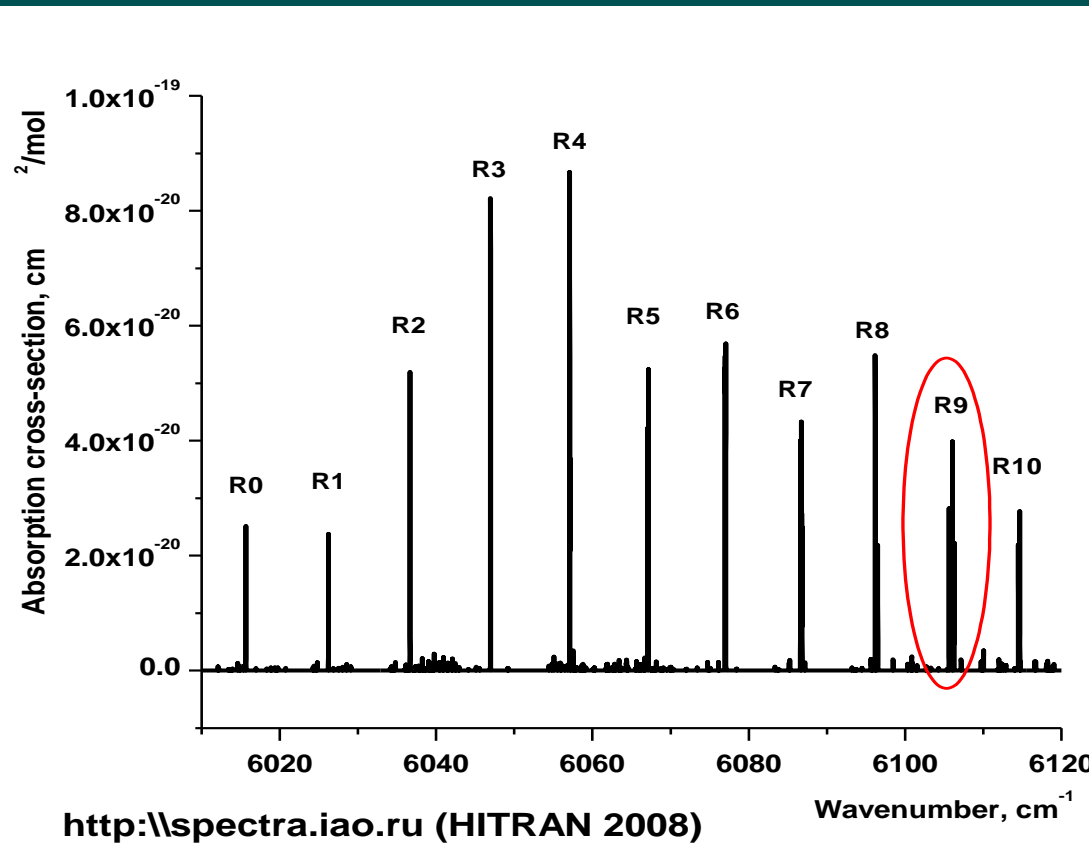
Residuals: black – Voigt, red – RS, green – Pine profiles



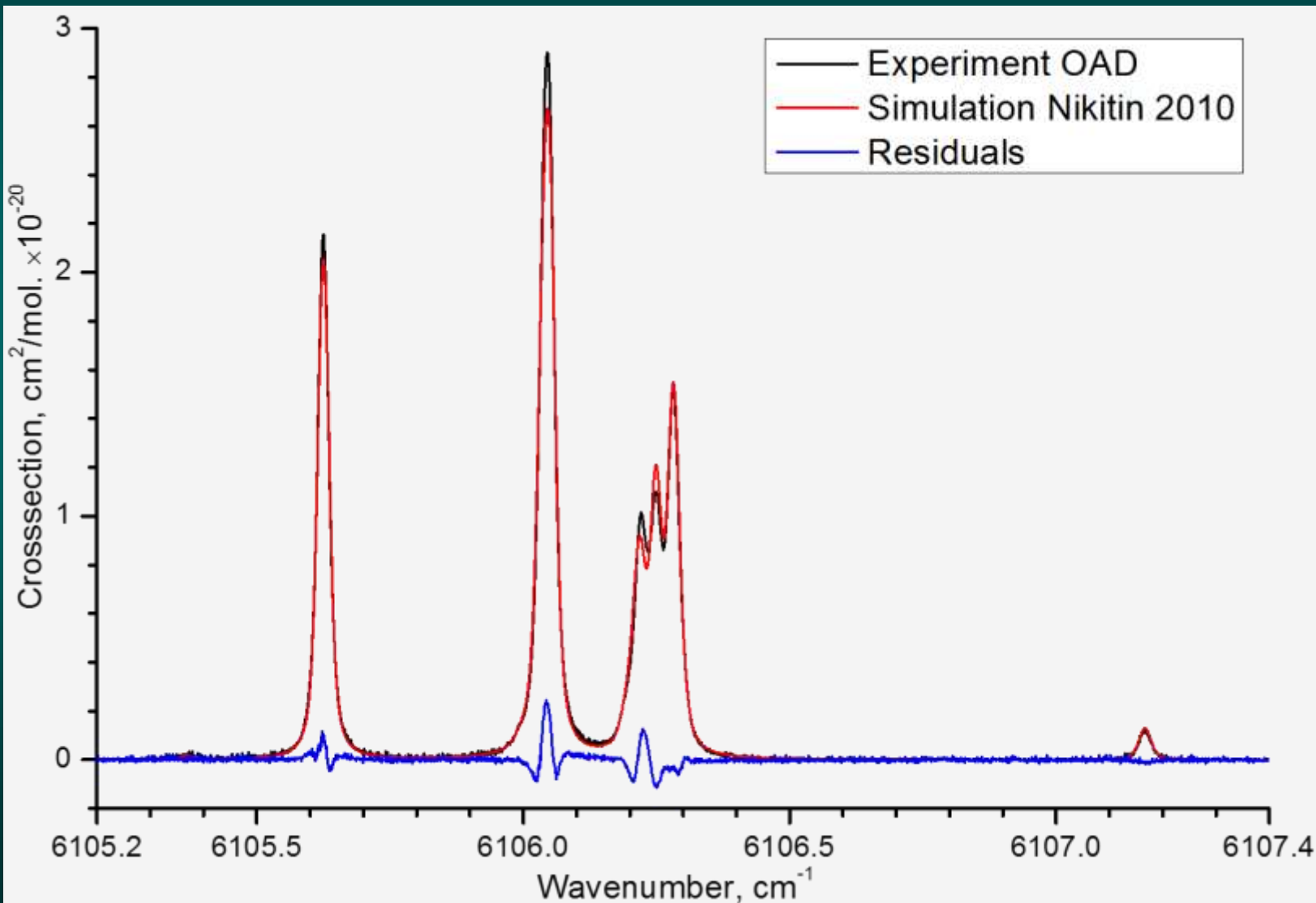
#	This work (Pine model [2])	This work (Rautian-Sobelman model [8])	This work (Voigt profile)	GOSAT data [7]	Frankenberg data [15]
Line center position, ω^0 , cm^{-1}					
1	6067.0818703(72)	6067.0818708(77)	6067.0817706(73)	6067.08186	6067.0816
2	6067.0999781(72)	6067.0999762(77)	6067.0998809(73)	6067.09982	6067.0997
3	6067.148138(35)	6067.148114(36)	6067.147999(33)	6067.14848	6067.1485
4	6067.157028(26)	6067.157032(26)	6067.156913(22)	6067.15554	6067.1570
Line intensity, S , $\text{cm/mol} \times 10^{-22}$					
1	9.2089(57)	9.2296(58)	9.1506(53)	8.874	8.803(12)
2	9.1754(56)	9.1854(58)	9.2490(56)	9.020	8.440(11)
3	6.465(45)	6.468(45)	6.322(40)	5.842	6.316(10)
4	8.603(45)	8.644(45)	8.779(39)	8.750	7.844(10)
Broadening coefficient, γ (N_2), $\text{cm}^{-1}/\text{atm}$					
1	0.05706(27)	0.05676(11)	0.057334(81)	0.067	0.0597(2)
2	0.06258(30)	0.06748(14)	0.064283(97)	0.067	0.0630(4)
3	0.05809(40)	0.06149(26)	0.06006(23)	0.061	0.0518(6)
4	0.06149(33)	0.05414(14)	0.05581(15)	0.058	0.0597(3)
Shifting coefficient, δ (N_2), $\text{cm}^{-1}/\text{atm}$					
1	-0.00620(29)	-0.006133(84)	-0.007040(82)	-0.0085	-0.0075(1)
2	-0.01315(31)	-0.002897(99)	-0.003108(93)	-0.0065	-0.0056(1)
3	-0.009577(59)	-0.01890(26)	-0.00885(26)	-0.0165	-0.0148(1)
4	-0.01307(34)	-0.01543(18)	-0.02080(17)	-0.0125	-0.0158(2)
Dike narrowing parameter, β (N_2), $\text{cm}^{-1}/\text{atm}$					
1	0.00980(50)	0.00174(40)	–	–	–
2	0.01430(56)	0.02142(57)	–	–	–
3	0.02442(14)	0.0310(12)	–	–	–
4	0.0202(10)	0.01719(77)	–	–	–
Line mixing parameter, ζ (N_2), atm^{-1}					
1	-0.511(Depend)	–	–	–	–
2	0.874(49)	–	–	–	–
3	0	–	–	–	–
4	-0.385(68)	–	–	–	–

Methane spectrum in the 6010-6200 cm^{-1}

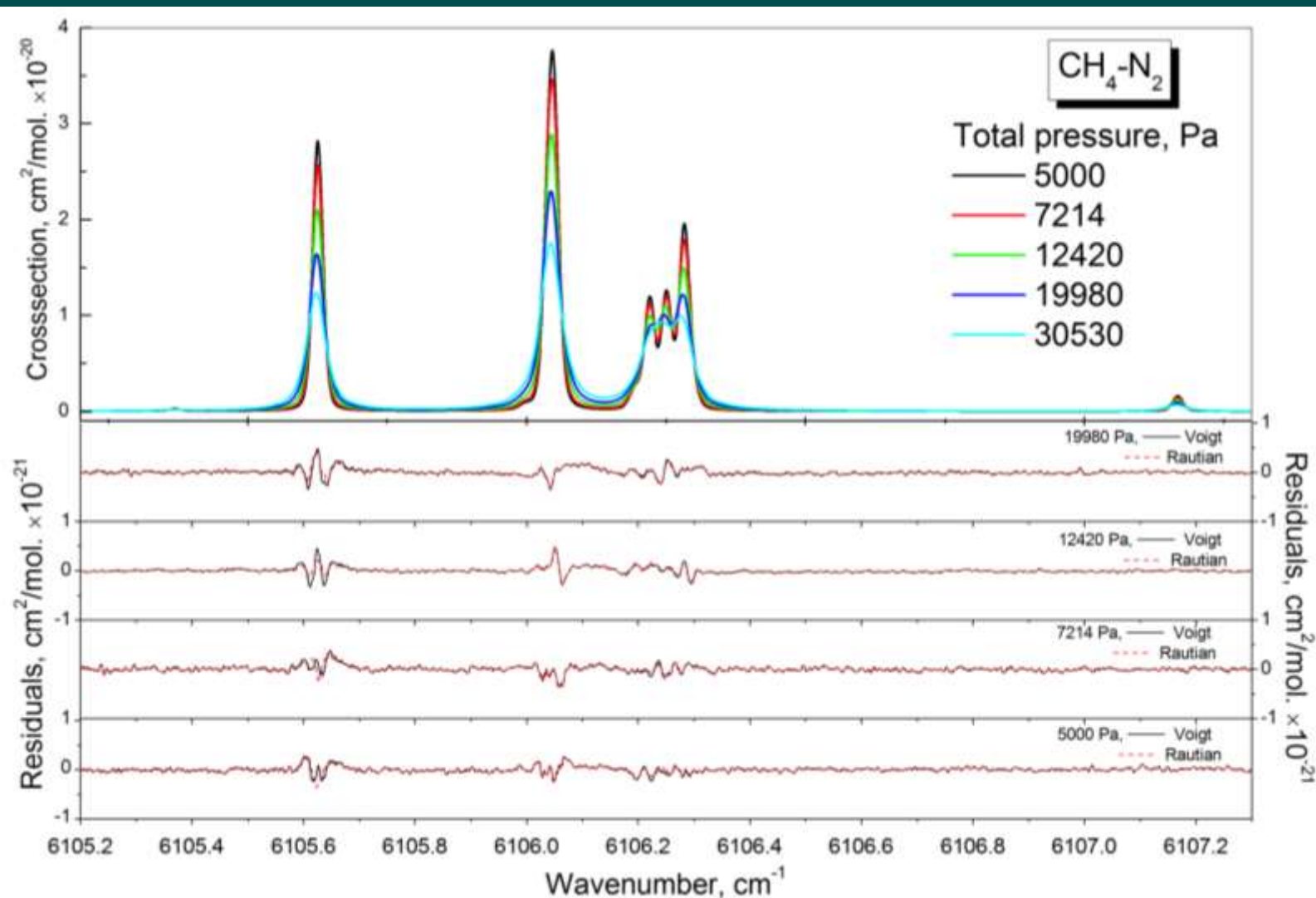
CH_4 absorption spectra are registered by two-channels diode laser OA spectrometer (IAO SB RAS, Tomsk) [Kapitanov V.A., Ponomarev Yu.N., Tyryshkin I.S. and Rostov A.P. // *Spectrochimica Acta Part A*. 2007. V. 66, N 4-5, P. 811-818.]



OAD spectrum+GOSAT simulatoin



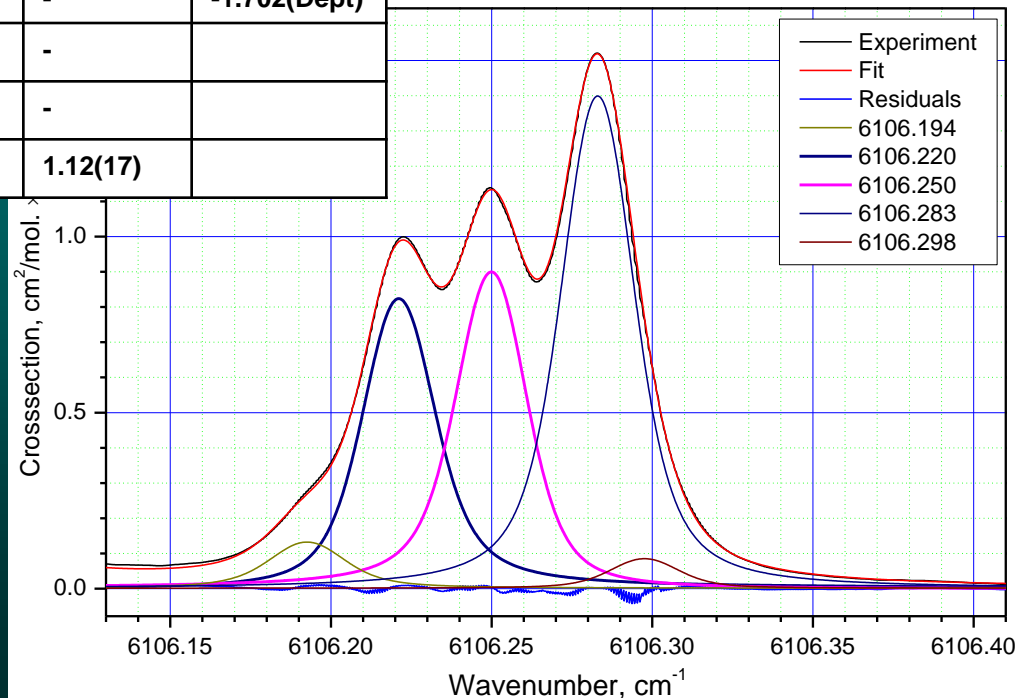
Simultaneous fitting of experimental spectra, registered by OA spectrometer at $T=296^{\circ}\text{C}$ and different partial pressures of $\text{CH}_4:\text{N}_2$ (1:14) mixture. Model spectrum consists of 11 lines with $S_i > 6E-24 \text{ cm/mol}$



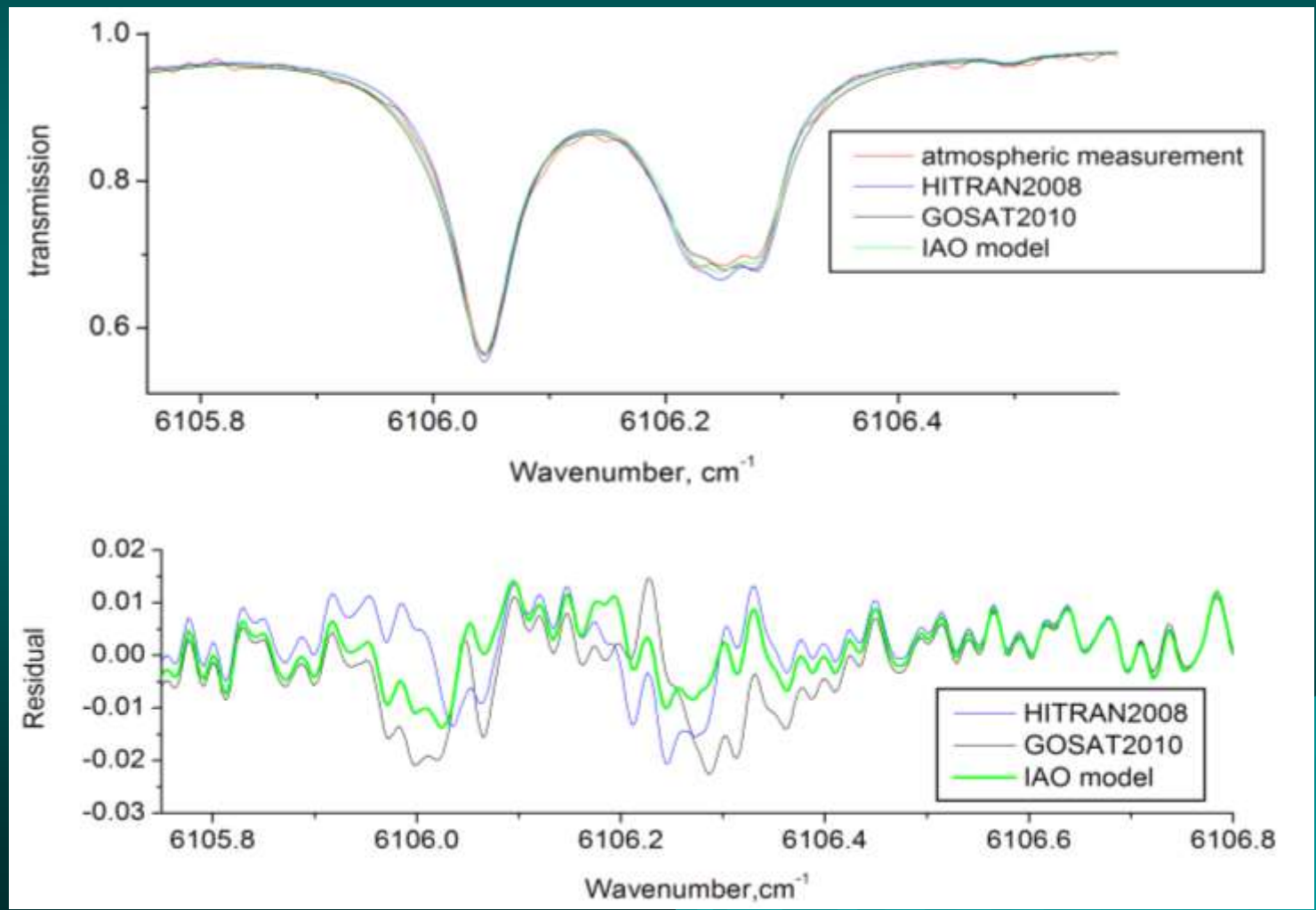
ν_0 , cm^{-1}	S , $\text{cm}^2/\text{mol} \times 10^{-22}$	γ_L , Ne, $\text{cm}^{-1}/\text{atm} \times 10^{-2}$	δ , Ne, $\text{cm}^{-1}/\text{atm} \times 10^{-2}$	β , $\text{cm}^{-1}/\text{atm} \times 10^{-2}$	η , $\text{cm}^2/\text{atm} \times 10^{-3}$
6105.36875(27)	0.0750(13)	5.34(25)	-0.00(27)	-	
6105.6258930(27)	7.0457(12)	3.3746(25)	-0.1415(21)	1.0824(94)	
6105.99560(13)	0.1079(11)	3.5(Fixed)	-0.5 (Fixed)	-	
6106.039566(16)	4.438(10)	3.8459(85)	0.6985(97)	-	
6106.050820(10)	6.652(10)	4.1899(58)	-0.5090(69)	-	
6106.194805(51)	0.4934(17)	4.210(43)	-1.225(52)	-	
6106.2205803(86)	2.8876(17)	3.6599(99)	0.087(16)	-	1.702(95)
6106.2519971(75)	3.0585(16)	3.4237(85)	-0.930(15)	-	-1.702(Dept)
6106.2839776(49)	4.9943(19)	3.7921(64)	-0.5230(62)	-	
6106.300823(95)	0.2716(Fixed)	3.051(71)	-1.770(99)	-	
6107.168209(44)	0.4797(14)	4.004(44)	-0.492(38)	1.12(17)	

Line parameters of R9 multiplet of $2\nu_3$ band, retrieved from the spectra, measured by OA spectrometer. Broadening gas is Ne. Line mixing is taken into account.

Line mixing model (Pine A. S. JQSRT 1997. Vol. 57. № 2. pp. 145-155.)



Comparison of atmospheric spectra, measured by ground-based Fourier spectrometer (FTS) and spectra, modeled with HITRAN2008, GOSAT CH₄ line list and our CH₄ line parameters, retrieved from OA spectrometer measurements (IAO model) (FTS July 1st, 2010. Solar zenith angle is 52.9°, Kourvka Ekaterinburg, Russia http://www.remotesensing.ru/fts_sta.html)

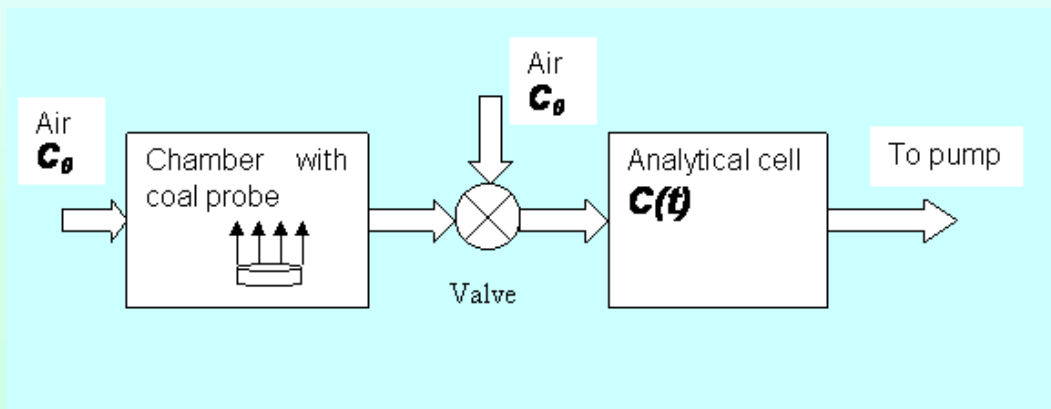


Methane line mixing

- The neglecting of the CH₄ line-mixing effects in the modeled spectra in the 1,6-1,7 μm can give an error in the methane column amount determination from the solar spectra, measured by ground—based Fourier spectrometer, up to 0.05 ppm for large solar zenith angles (H. Tran et al, JQSRT,11, 2010)
- Taking CH₄ line mixing into account in the spectral region of 7-10 μm allows to decrease the difference between calculated emitted radiance and values measured in the Jupiter atmosphere from 0,5 W/m²sr (10%) to 0,02 W/m²sr (0,3%). [*H. Tran et al. JQSRT. 2006. V.101*]
- Neglect of CH₄ line mixing in the inversion model of CH₄ concentration retrieval from the atmospheric limb transmission spectra, measured by FTIR spectrometer in the 3,4 μm spectral region, leads to an error of 7% in the retrieved methane concentration. [*D. Mondelain et al. JMS. 2007. V.244*]

Our preliminary studies found an effect of CH₄ line mixing in the measured spectra of methane in the 1,6-1,7 μm (R5 and R9 manifolds) at the pressures 0-1 atm. [*K. Yu. Osipov, V.A. Kapitanov, Yu.N. Ponomarev, Protasevich A.E. HRMS, Dijon 2011*]

TDLS applications to study of methane emissions (hydrocarbons, coal, biological systems)



Is there an increased methane flux at the lake surface in the seep areas and how does the methane concentration in the air vary spatially?

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Institute of Atmospheric Optics,
SB Russian Academy of Sciences, Tomsk***

v.VERESHAGIN

air bleeding position



2003

2004
Chamber

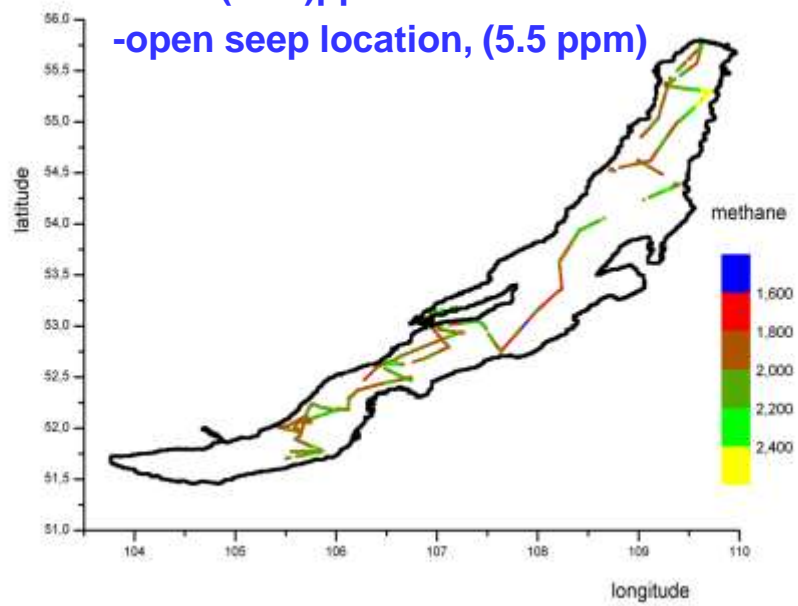
v.Verechagin



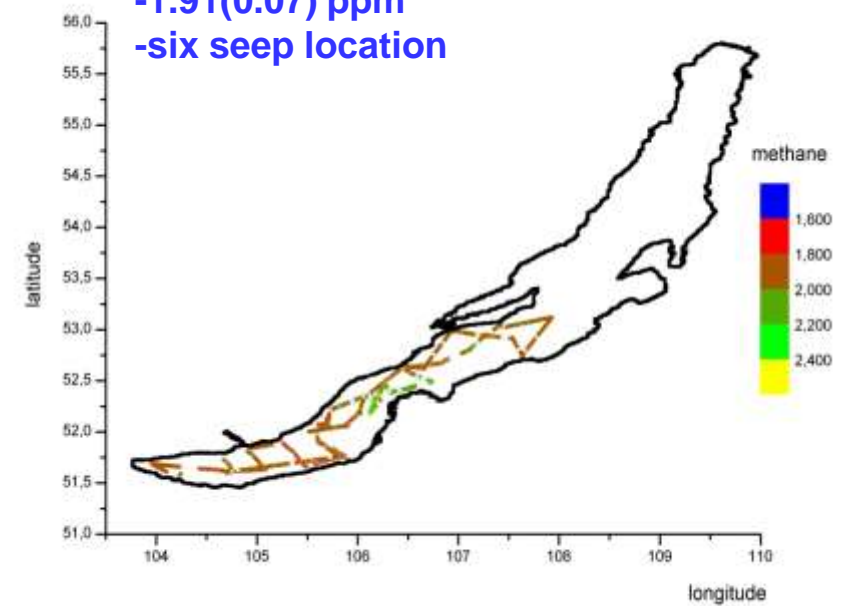
8-17 AUGUST 2003

16-24 JUNE 2004

-2.001(0.16)ppm
-open seep location, (5.5 ppm)



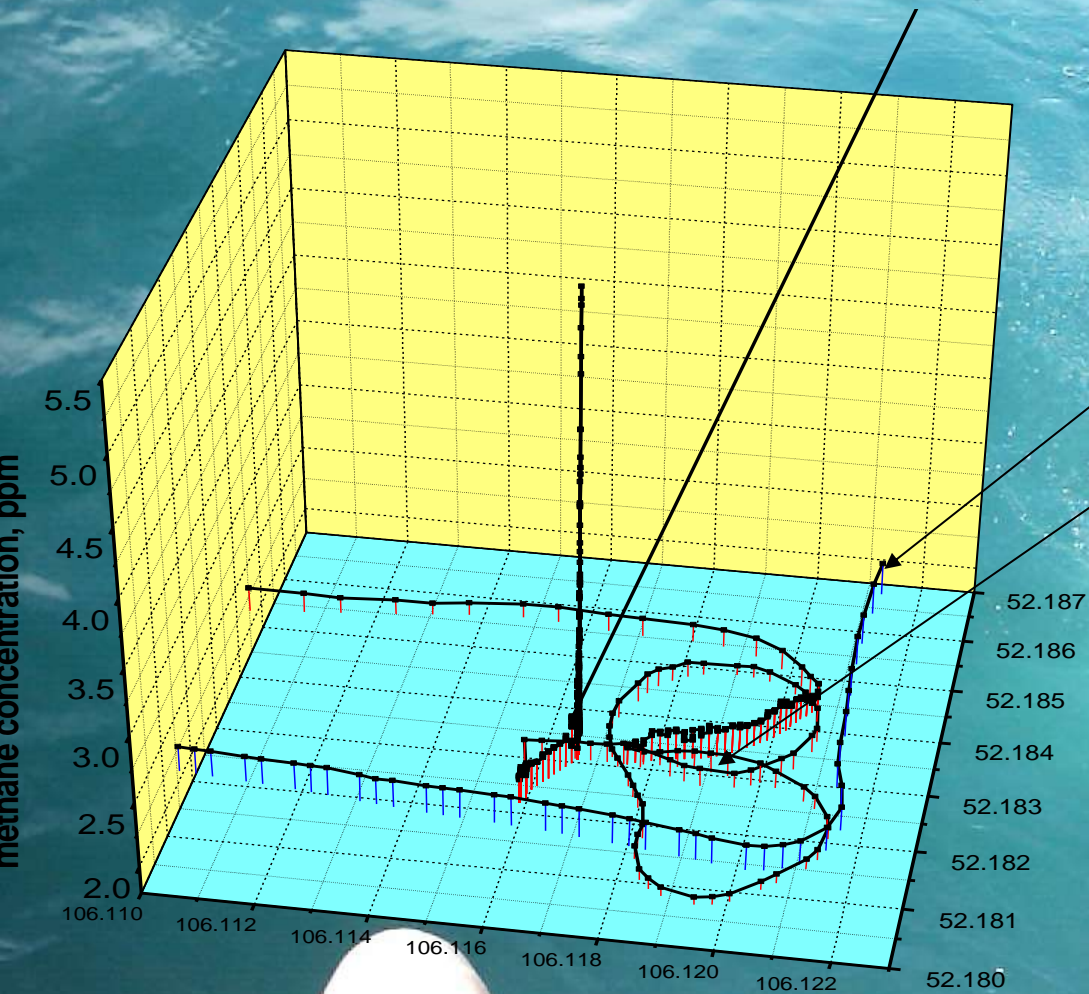
-1.91(0.07) ppm
-six seep location



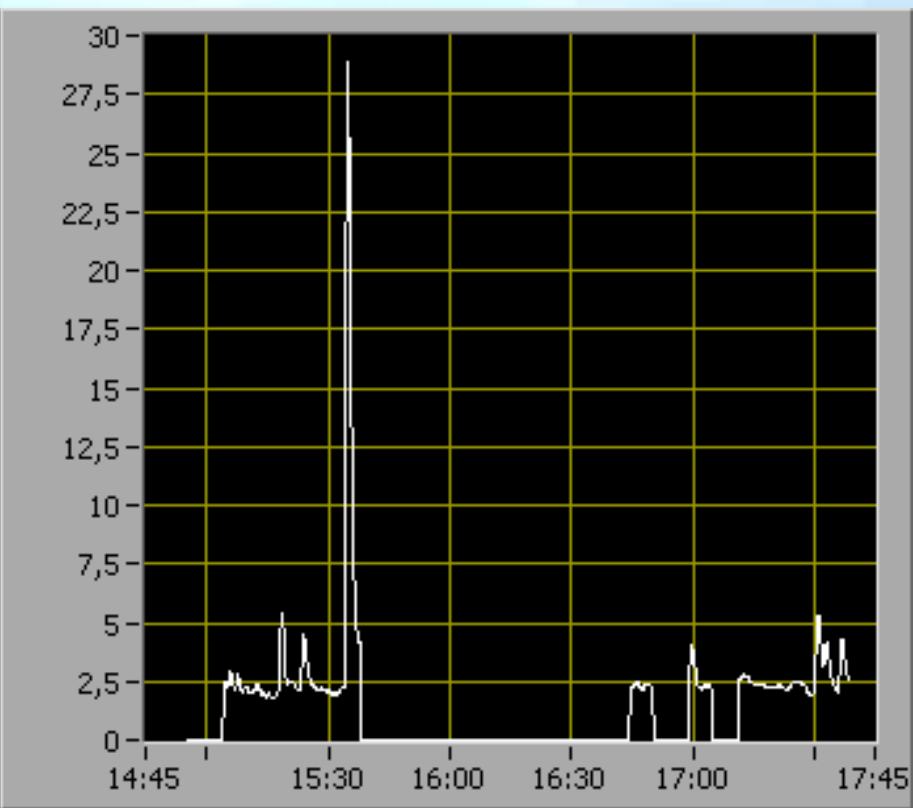
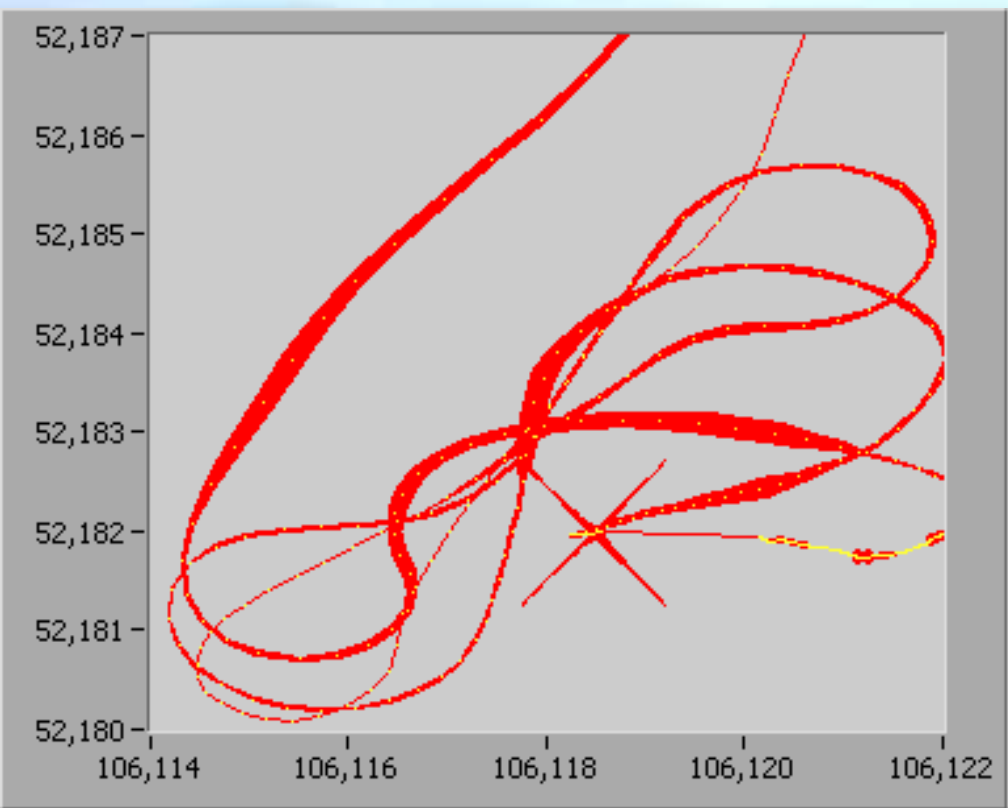
METHANE SEEP LOCATION

open seep near Selenga

11.08.03
15.08.03

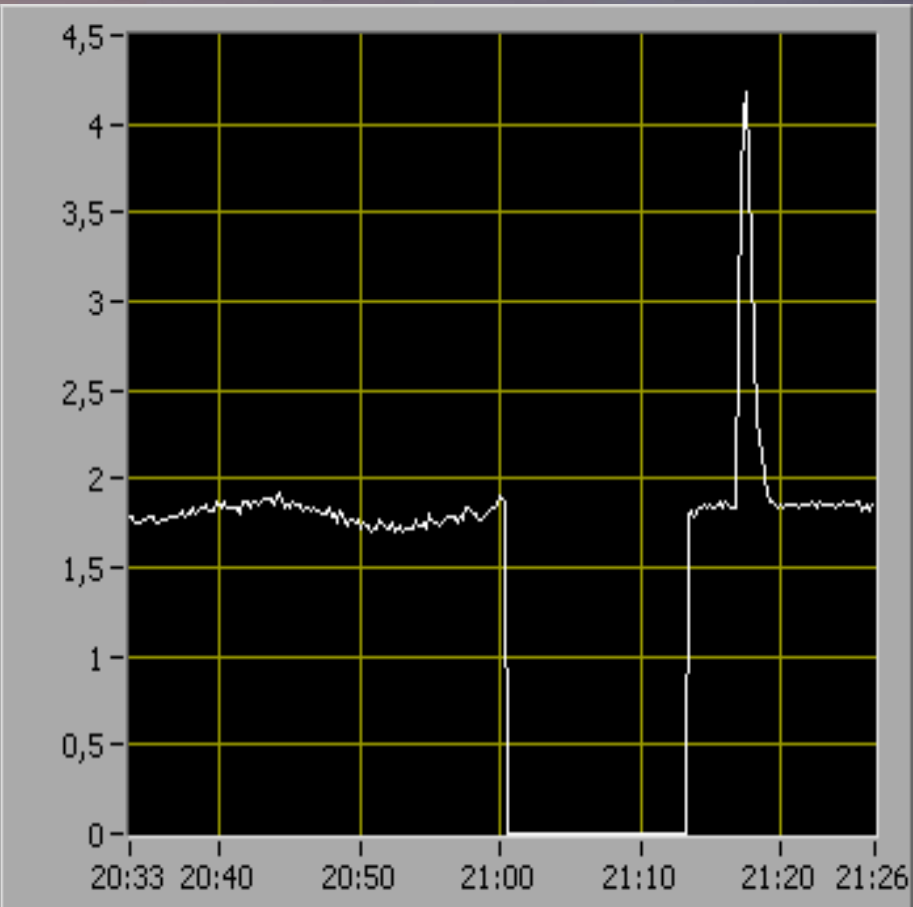
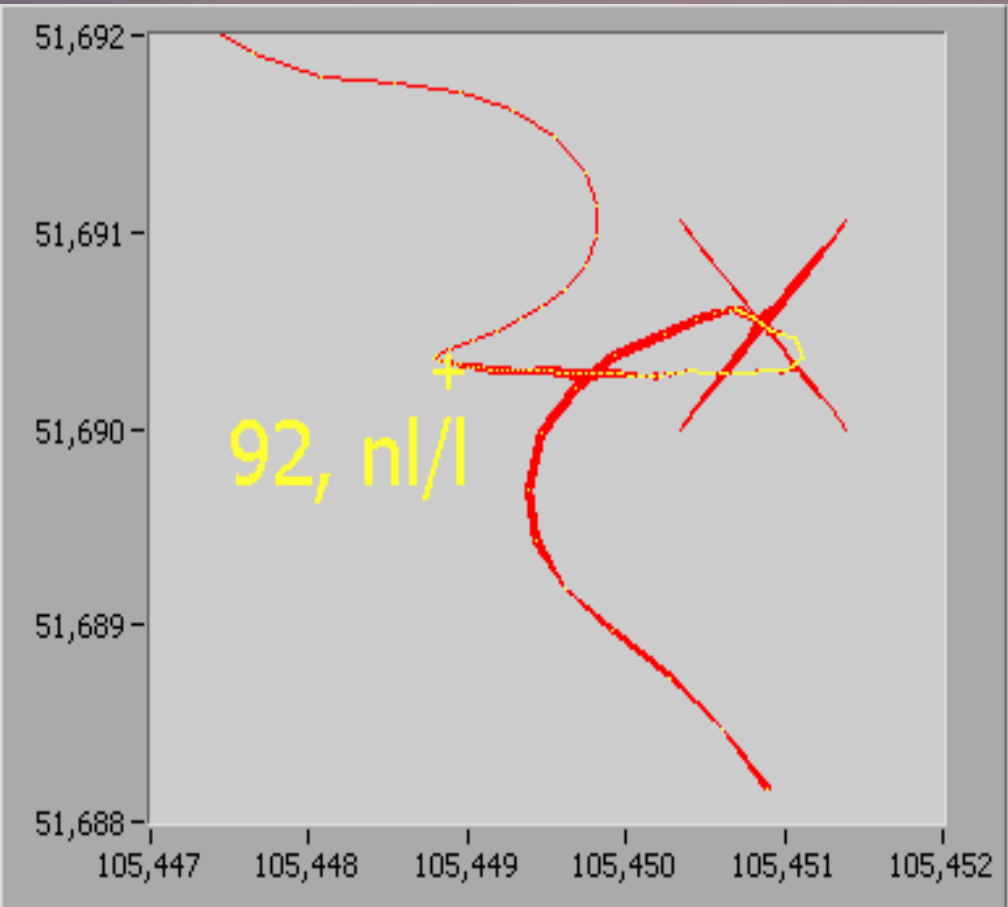


OPEN SEEP (Selenga's entry), 22 June 2004



DEEP-SEA SEEP

18.08.04 June, near Mishicha, depth ~ 1000 m



TEAM WORK (IOA+POI)

- *Intercomparison*

Air sample for gas chromatograph was bled directly from air tube near detector

21.06.04

GC five air samples
Methane detector



- 1.89 (0.056) ppm

- **1.94 (0.054) ppm**

22.06.04

GC two air samples
Methane detector

- 1.95 (0.056) ppm

- **2.02 (0.09) ppm**

23.06.04

GC one air sample
Methane detector

- 8.98 (0.27) ppm

- **13.7 (0.09) ppm**

- Direct measurements and evaluation of water surface **methane fluxes** in South and Middle Baikal were carried out.
- High methane content in water or air within seep area is strongly located with scale ~ 100 m
- Singularity of air methane content like “seep” was detected near Mishicha, (depth > 1000 m)

Summary

- **Progress of TDL provides the permanent perfection of all types of TDL spectroscopy (absorption spectrophotometry, CRDS, PAS, etc.), leads to sensitivity and tuning range increase.**
- **Due to high sensitivity and high resolution of TDLS methods the detail analysis of molecular spectra in gases requires:**
 - **Application of multispectrum fitting instead spectrum by spectrum procedure;**
 - **Perfection of line shape models;**
 - **Taking into account even very weak neighbor spectral lines**
- **TDLS techniques could be efficiently applied to basic study of spectra structure and line shape parameters under pressure and temperature varying as well as for application to gas analysis and environmental monitoring.**

Acknowledgements

- **The authors are very thankful TDLS division of IGP RAS and personally Prof. A.Nadezhdenskii and Dr. Ya.Ponurovskii, K. Osipov and A. Protasevitch (IAO SB RAS) for collaboration and help.**
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