



Mid-infrared semiconductor laser based trace gas technologies: recent advances and applications

F. K. Tittel¹, R. Lewicki^{1,4}, M. Jahjah¹, P. Stefanski^{1,4}, J. Tarka^{1,4}, L. Gong², R. Griffin², S. So⁴, D. Thomazy⁴, W. Jiang¹, J. Zhang^{1,6}, P. Lane⁵, R. Talbot⁵

¹ Department of Electrical and Computer Engineering, Rice University, 6100 Main St., Houston, TX 77005, USA

² Department of Civil and Environmental Engineering, Rice University, 6100 Main St., Houston, TX 77005, USA

³ Laser and Fiber Electronics Group, Wroclaw University of Technology, Wybrzeze Wyspianskiego 27, Wroclaw, Poland

⁴ Sentinel Photonics, Monmouth Junction, NJ 08852, USA

⁵ University of Houston, Department of Earth & Atmospheric Sciences, Houston, TX 77204, USA

⁶ Northeast Forestry University, Department of Electromechanical Engineering, Harbin, China

OUTLINE

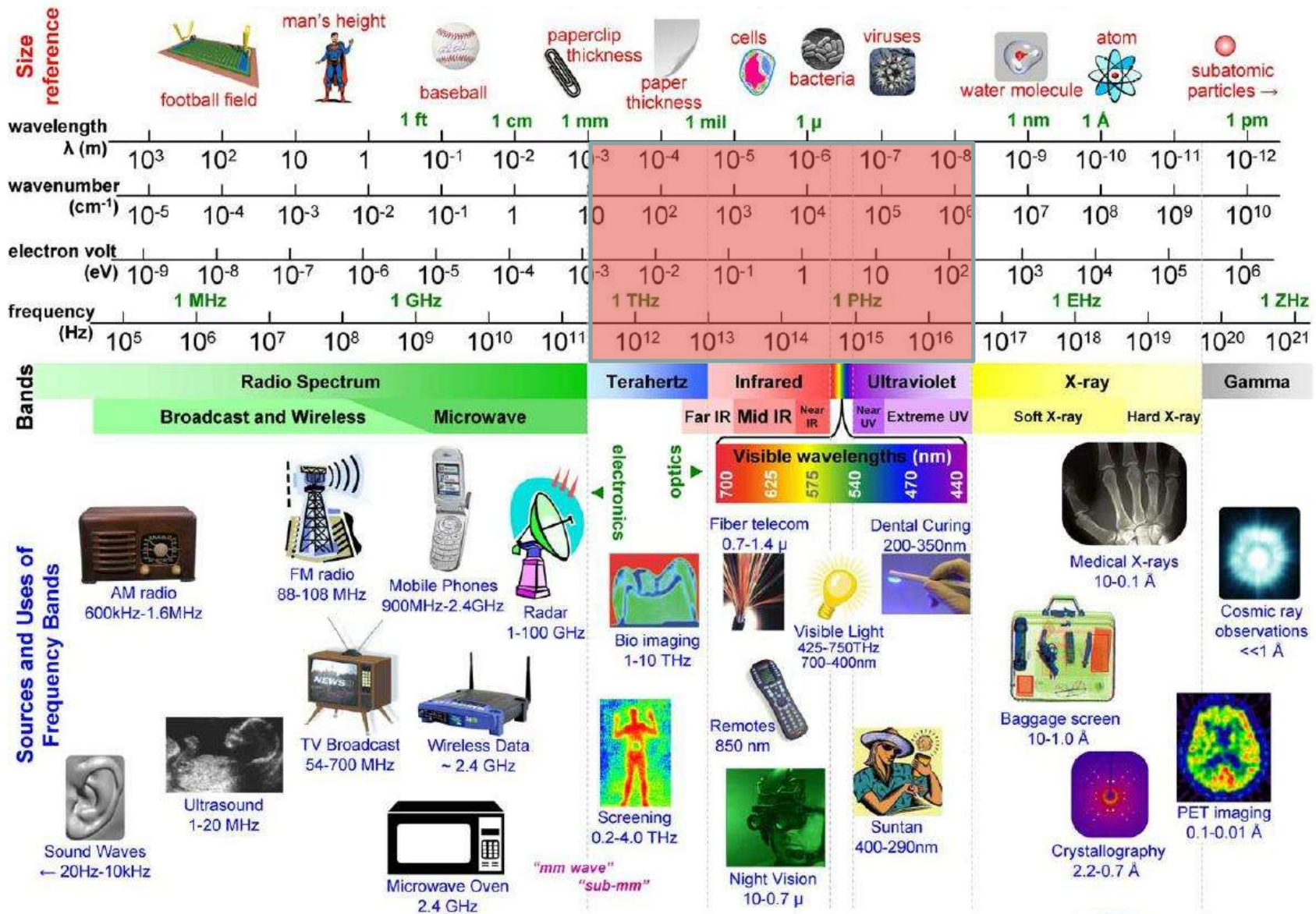
TDLS 2013

**Moscow,
Russia**

June 19, 2013

- New Laser Based Trace Gas Sensor Technology
 - Novel Multipass Absorption Cell & Electronics
 - Quartz Enhanced Photoacoustic Spectroscopy
- Examples of Mid-Infrared Sensor Architectures
 - C₂H₆, NH₃, NO, CO, SO₂, CH₄ and N₂O
 - Future Directions of Laser Based Gas Sensor Technology and Conclusions

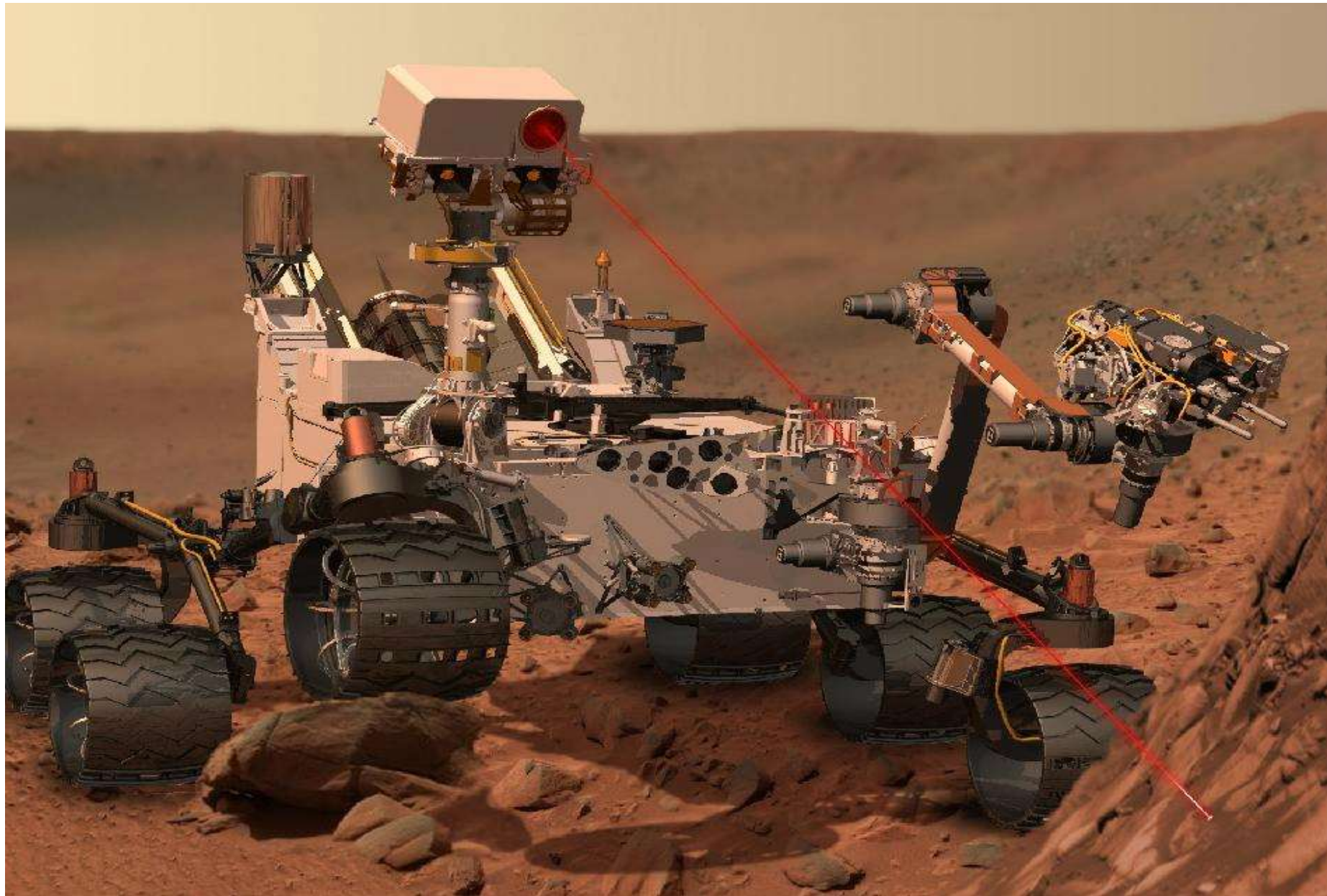
Mid-IR and THz Spectroscopic Phenomena



Wide Range of Trace Gas Sensing Applications

- **Urban and Industrial Emission Measurements**
 - Industrial Plants
 - Combustion Sources and Processes (e.g. fire detection)
 - Automobile, Truck, Aircraft and Marine Emissions
- **Rural Emission Measurements**
 - Agriculture & Forestry, Livestock
- **Environmental Monitoring**
 - Atmospheric Chemistry (e.g isotopologues, climate modeling,...)
 - Volcanic Emissions
- **Chemical Analysis and Industrial Process Control**
 - Petrochemical, Semiconductor, Pharmaceutical, Metals Processing, Food & Beverage Industries; Nuclear Technology & Safeguards
- **Spacecraft and Planetary Surface Monitoring**
 - Crew Health Maintenance & Life Support
- **Applications in Medical Diagnostics and the Life Sciences**
- **Technologies for Law Enforcement, Defense and Security**
- **Fundamental Science and Photochemistry**

“Curiosity” landed on Mars on August 6, 2012



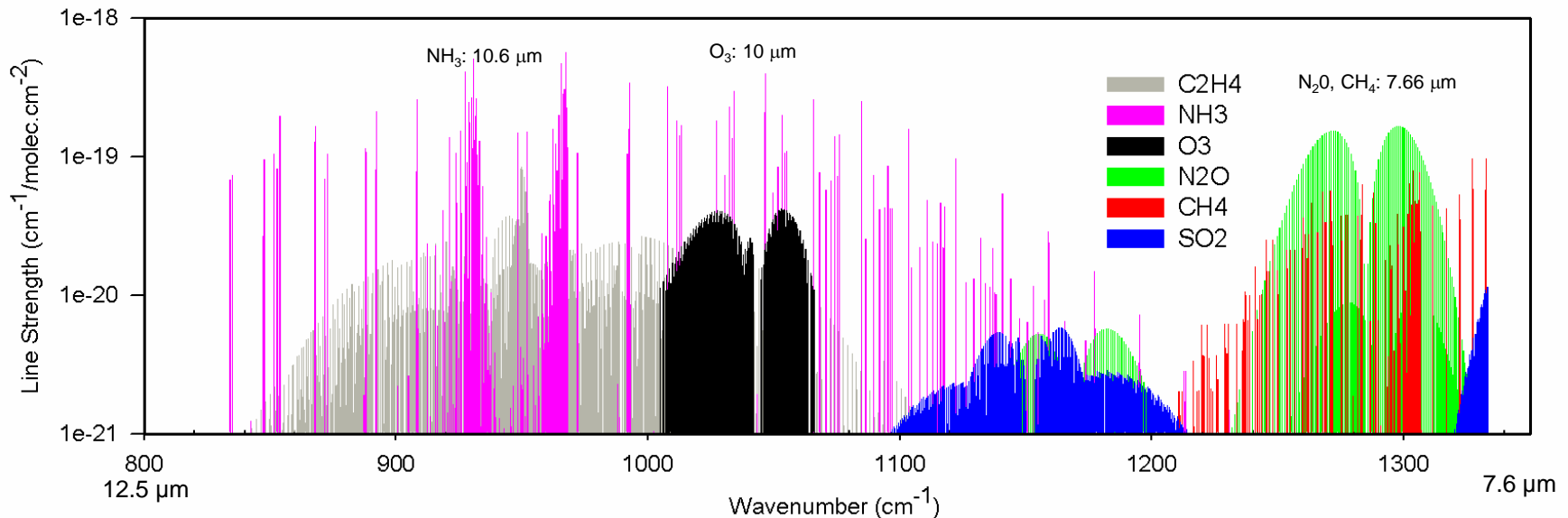
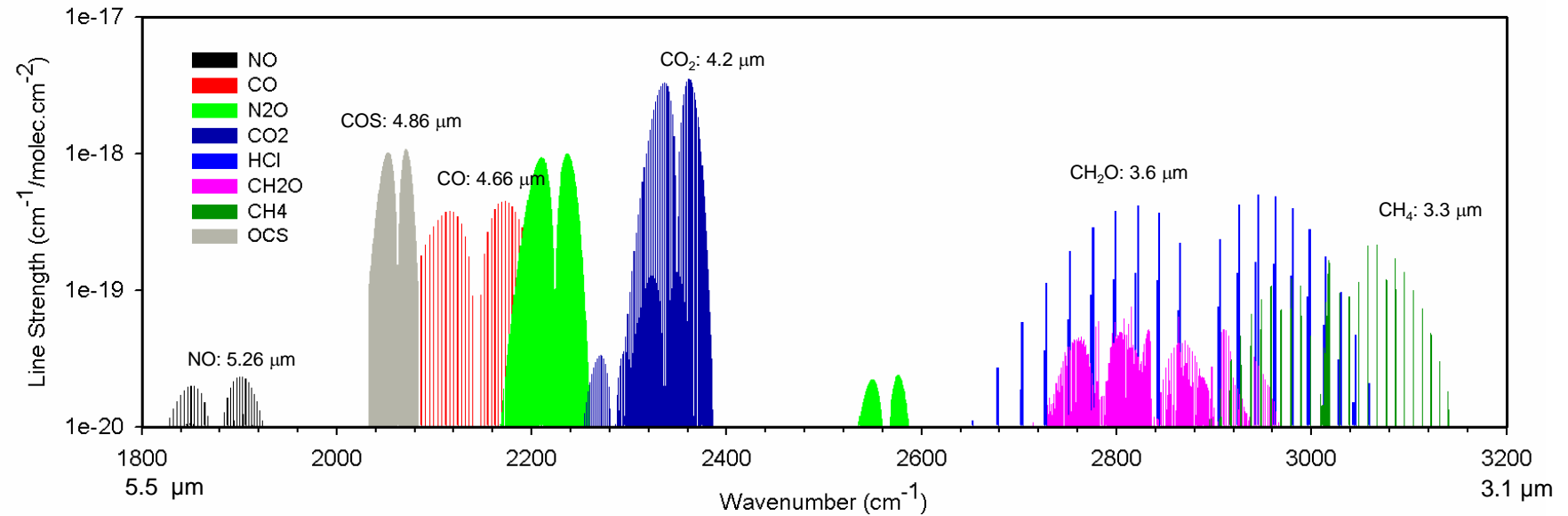
Laser based Trace Gas Sensing Techniques

- **Optimum Molecular Absorbing Transition**
 - Overtone or Combination Bands (NIR)
 - Fundamental Absorption Bands (Mid-IR)
- **Long Optical Pathlength**
 - Multipass Absorption Cell (White, Herriot, Chernin, Sentinel Photonics)
 - Cavity Enhanced and Cavity Ringdown Spectroscopy
 - Open Path Monitoring (with retro-reflector): Standoff and Remote Detection
 - Fiberoptic Evanescent Wave Spectroscopy
- **Spectroscopic Detection Schemes**
 - Frequency or Wavelength Modulation
 - Balanced Detection
 - Zero-air Subtraction
 - Photoacoustic & Quartz Enhanced Photoacoustic Spectroscopy (QEPAS)

Other spectroscopic methods

- Faraday Rotation Spectroscopy (limited to paramagnetic chemical species)
- Differential Optical Dispersion Spectroscopy (DODiS)
- Noise Immune Cavity Enhanced-Optical Heterodyne Molecular Spectroscopy (NICE-OHMS)
- Frequency Comb Spectroscopy
- Laser Induced Breakdown Spectroscopy (LIBS)

HITRAN Simulated Mid-Infrared Molecular Absorption Spectra



Mid-IR Source Requirements for Laser Spectroscopy

<u>REQUIREMENTS</u>	<u>IR LASER SOURCE</u>
Sensitivity (% to ppt)	Optimum Wavelength, Power
Selectivity (Spectral Resolution)	Stable Single Mode Operation and Narrow Linewidth
Multi-gas Components, Multiple Absorption Lines and Broadband Absorbers	Mode Hop-free Wavelength Tunability
Directionality or Cavity Mode Matching	Beam Quality
Rapid Data Acquisition	Fast Time Response
Room Temperature Operation	High wall plug efficiency, no cryogenics or cooling water
Field deployable in harsh environments	Compact & Robust

Key Characteristics of Mid-IR QCL & ICL Sources – May 2013

- **Band – structure engineered devices**

Emission wavelength is determined by layer thickness – MBE or MOCVD; Type I QCLs operate in the 3 to 24 μm spectral region; Type II and GaSb based ICLs can cover the 3 to 6 μm spectral range.

- Compact, reliable, stable, long lifetime, and commercial availability
- Fabry-Perot (FP), single mode (DFB) and multi-wavelength devices

- **Wide spectral tuning ranges in the mid-IR**

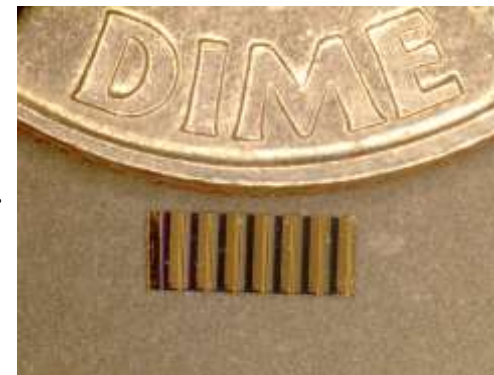
- 1.5 cm^{-1} using injection current control for DFB devices
- 10-20 cm^{-1} using temperature control for DFB devices
- $\sim 100 \text{cm}^{-1}$ using current and temperature control for QCL DFB Array
- $\sim 525 \text{cm}^{-1}$ (22% of c.w.) using an external grating element and FP chips with heterogeneous cascade active region design; also QCL DFB Array

- **Narrow spectral linewidths**

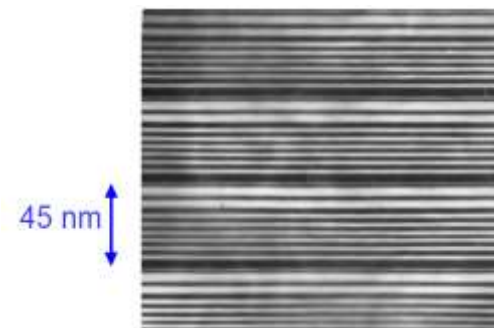
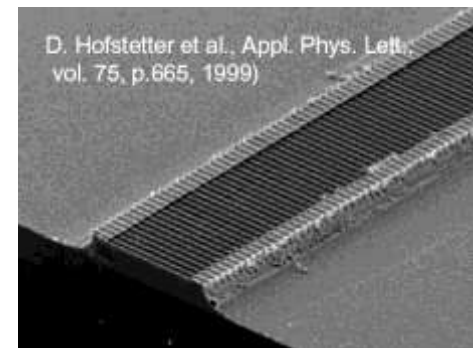
- CW: 0.1 - 3 MHz & $< 10 \text{kHz}$ with frequency stabilization (0.0004cm^{-1})
- Pulsed: $\sim 300 \text{MHz}$

- **High pulsed and CW powers of QCLs at TEC/RT temperatures**

- Room temperature pulsed power of $> 30 \text{W}$ with 27% wall plug efficiency and CW powers of $\sim 5 \text{W}$ with 21% wall plug efficiency
- $> 1 \text{W}$, TEC CW DFB @ 4.6 μm
- $> 600 \text{mW}$ (CW FP) @ RT; wall plug efficiency of $\sim 17 \%$ at 4.6 μm ;



4 mm

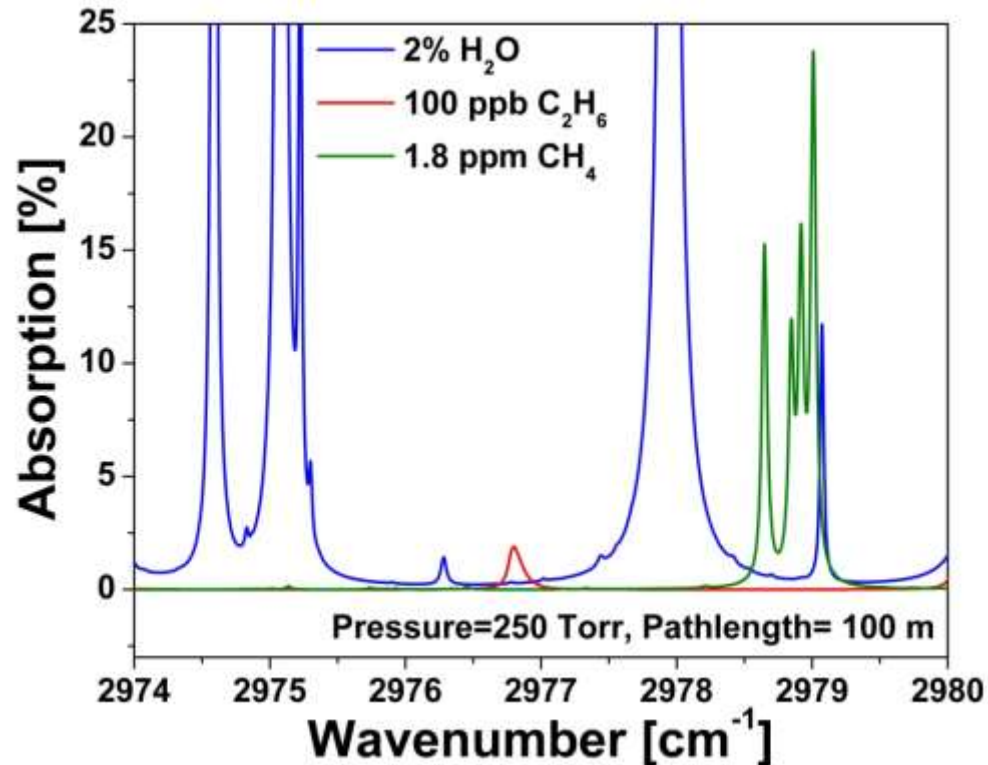


Improvements and New Capabilities of QCLs and ICLs

- Optimum wavelength (> 3 to $< 20 \mu\text{m}$) and power ($> 10 \text{ mW}$ to $< 1 \text{ W}$) at room temperature ($> 15^\circ\text{C}$ and $< 30^\circ\text{C}$) with state-of-the-art fabrication/processing methods based on MBE and MOCVD, good wall plug efficiency and lifetime ($> 20,000$ hours) for detection sensitivities from % to pptv with low electrical power budget
- Stable single TEM_{00} transverse and axial mode, CW and pulsed operation of mid-infrared laser sources (narrow linewidth of $\sim 300 \text{ MHz}$ to $< 10\text{kHz}$)
- Mode hop-free ultra-broad wavelength tunability for detection of broad band absorbers and multiple absorption lines based on external cavity or mid-infrared semiconductor arrays
- Good beam quality for directionality and/or cavity mode matching. Implementation of innovative collimation concepts.
- Rapid data acquisition based on fast time response
- Compact, robust, readily commercially available and affordable in order to be field deployable in harsh operating environments (temperature, pressure, etc...)

Motivation for Mid-infrared C_2H_6 Detection

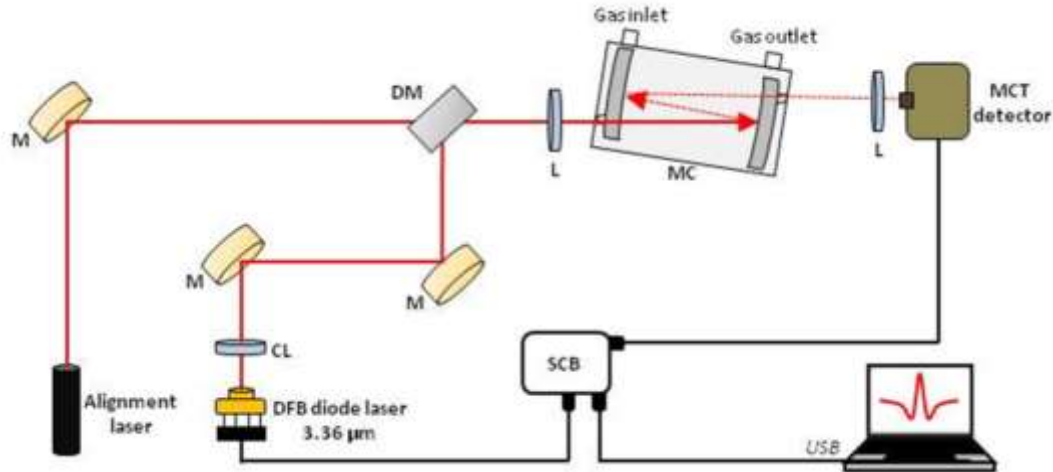
- Atmospheric chemistry and climate
 - Fossil fuel and biofuel consumption,
 - biomass burning,
 - vegetation/soil,
 - natural gas loss
- Oil and gas prospecting
- Application in medical breath analysis (a non-invasive method to identify and monitor different diseases):
 - asthma,
 - schizophrenia,
 - Lung cancer,
 - lung cancer,
 - vitamin E deficiency.



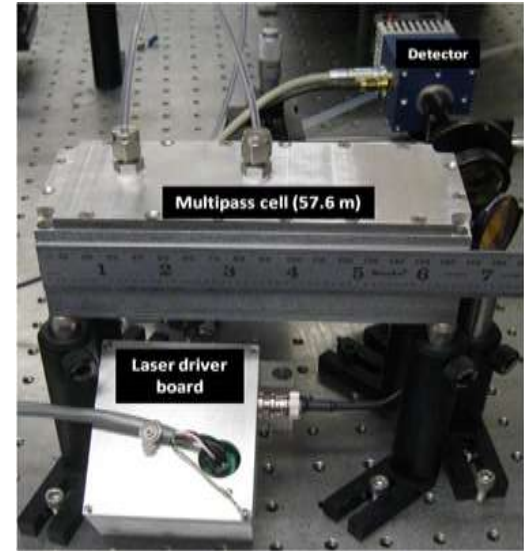
HITRAN absorption spectra of C_2H_6 , CH_4 , and H_2O



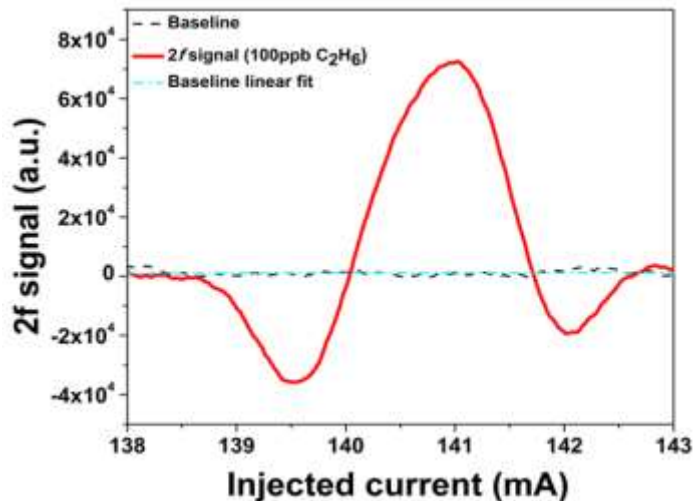
C_2H_6 Detection with a 3.36 μm CW DFB LD using a Novel Compact Multipass Absorption Cell and Control Electronics



Schematic of a C_2H_6 gas sensor using a Nanoplus 3.36 μm DFB laser diode as an excitation source. M – mirror, CL – collimating lens, DM – dichroic mirror, MC – multipass cell, L – lens, SCB – sensor control board.



Innovative long path, small volume multipass gas cell: 57.6 m with 459 passes



2f WMS signal for a C_2H_6 line at 2976.8 cm^{-1} at a pressure of 200 Torr

Minimum detectable C_2H_6 concentration is:
~ 130 pptv (1σ ; 1 s time resolution)

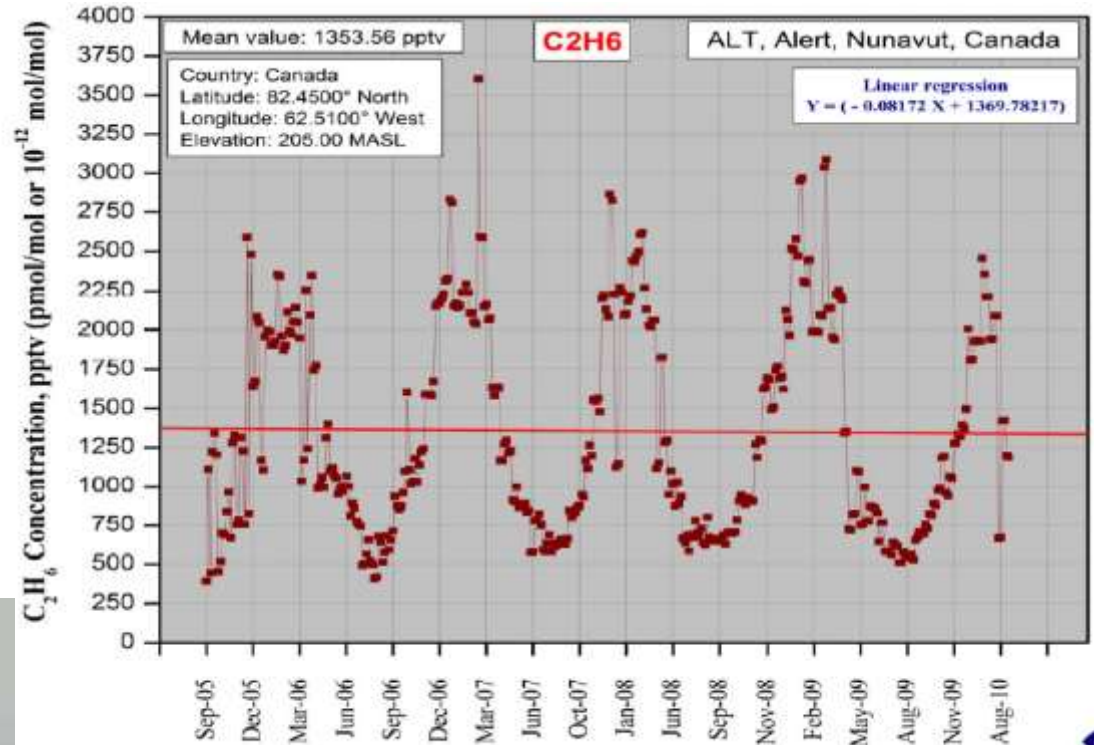


MC dimensions: 17 x 6.5 x 5.5 (cm)
Distance between the MGC mirrors: 13 cm

NOAA Monitoring & Sampling Location: Alert, Nunavut, Canada



ALT, Ethane Concentration Measurements



General View on the Facility

Latitude: 82.4508° North

Longitude: 62.5056° West

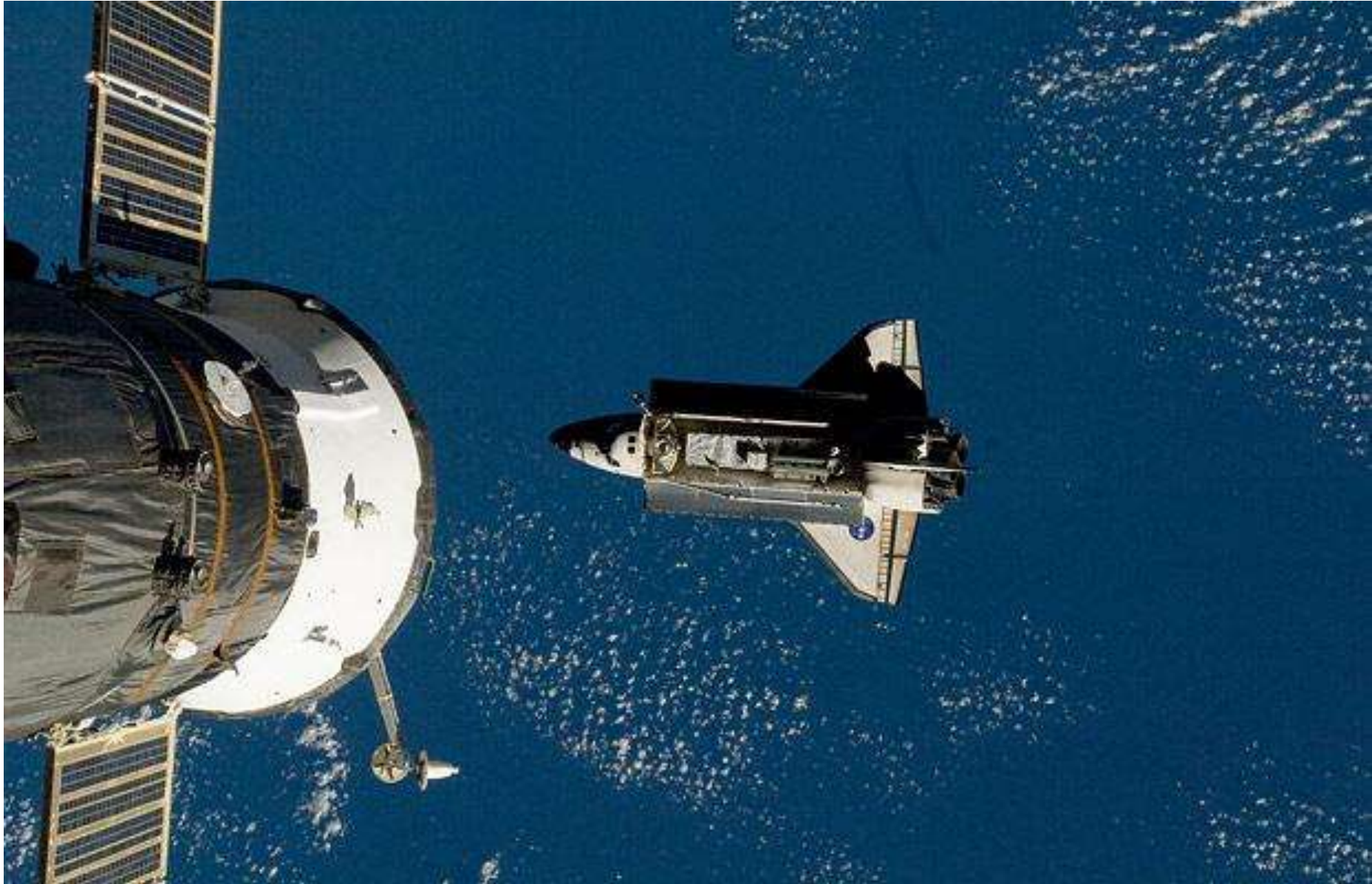
Elevation: 200.00 m



Motivation for NH₃ Detection

- Atmospheric chemistry
- Pollution gas monitoring
- Monitoring NH₃ concentrations in the exhaust stream of NO_x removal systems based on selective catalytic reduction (SCR) techniques
- Spacecraft related trace gas monitoring
- Semiconductor process monitoring & control
- Monitoring of industrial refrigeration facilities
- Monitoring of gas separation processes
- Medical diagnostics (kidney & liver diseases)
- Detection of ammonium-nitrate explosives

Ammonia Leaks from ISS May 2013



Conventional PAS

Cell is **OPTIONAL!**

V-effective volume

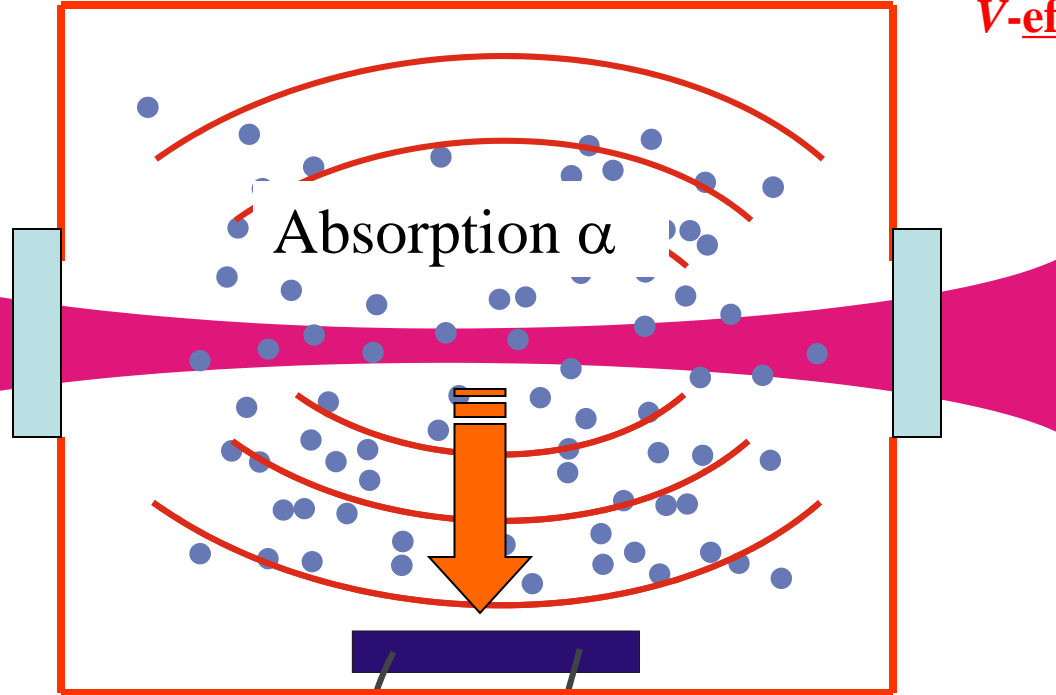
Laser beam,
power P



Modulated
(P or λ) at f
or $f/2$

$$S \sim \frac{Q \alpha P}{f V}$$

$$NNEA = \frac{\alpha_{\min} P}{\sqrt{\Delta f}} \left[\frac{\text{cm}^{-1} \times \text{W}}{\sqrt{\text{Hz}}} \right]$$

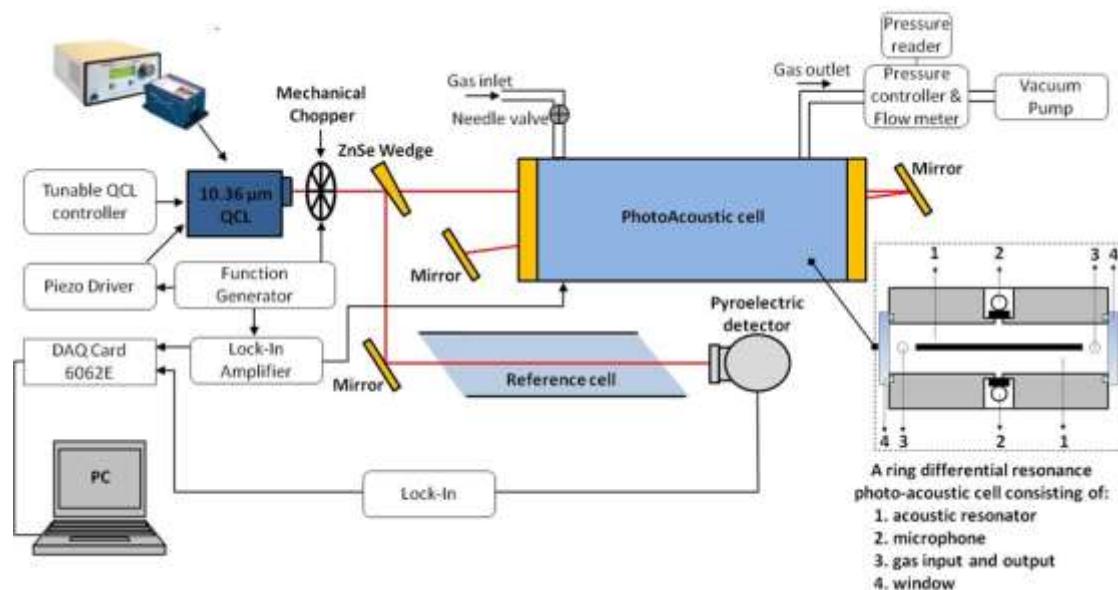


**Sensitive
microphone**



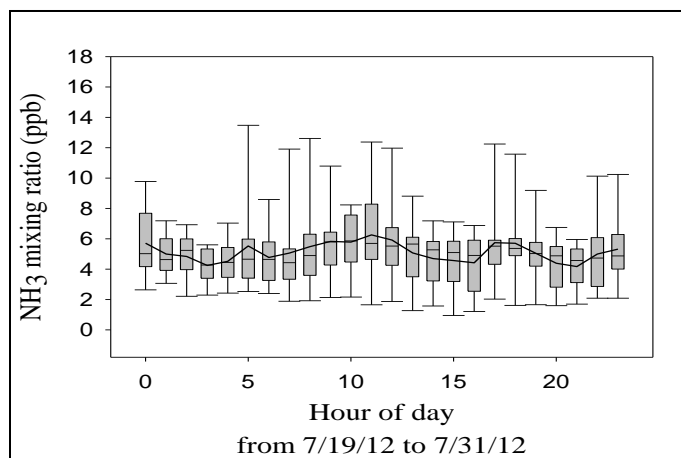
RICE

Atmospheric NH₃ Measurements using an EC-QCL PAS Sensor

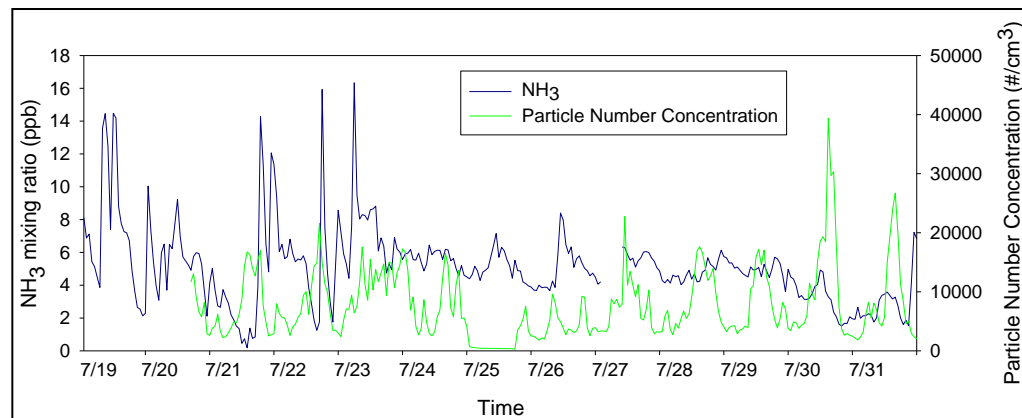


NH₃ sensor deployed at the UH Moody Tower rooftop monitoring site.

Schematic of a Daylight Solutions 10.36 μm CW TEC EC-QCL based PAS NH₃ Sensor.

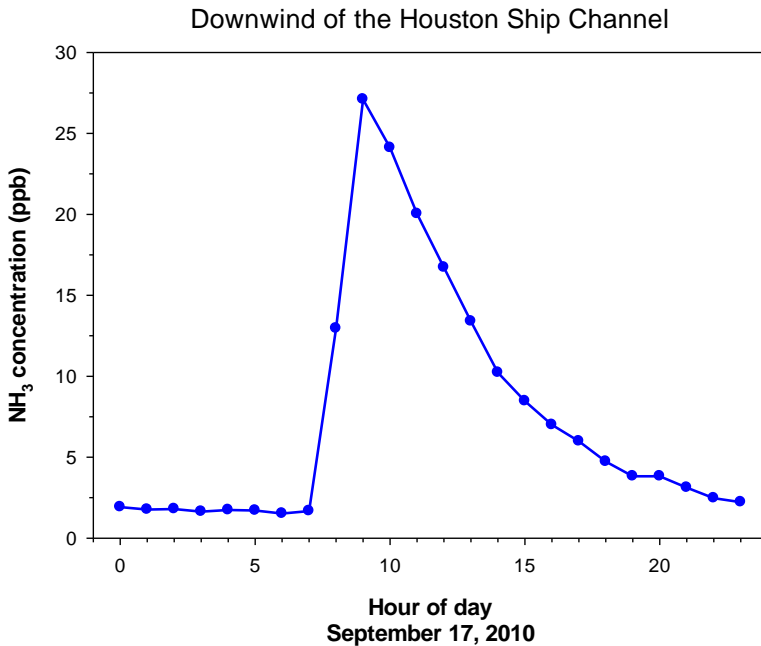
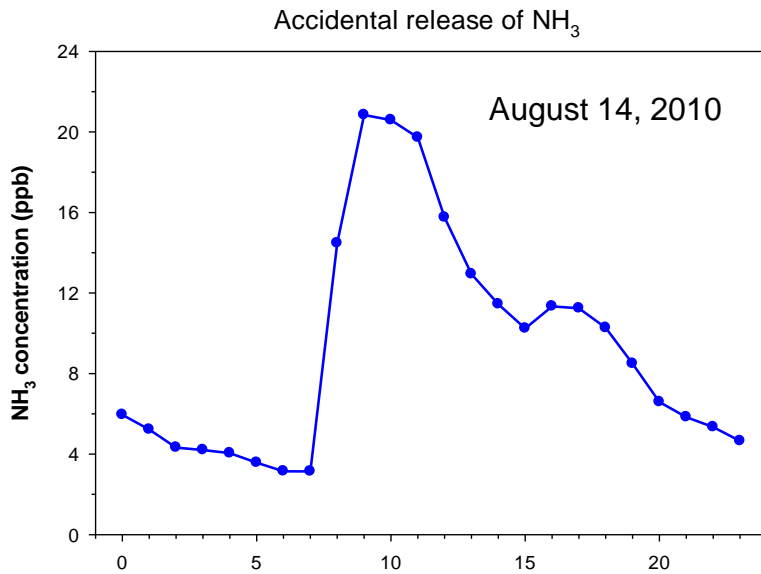


Diurnal profile of atmospheric NH₃ levels in Houston, TX.



Comparison between NH₃ and particle number concentration time series from July 19 to July 31 2012.

NH₃ Detection due to a Fire resulting from a Truck Collision



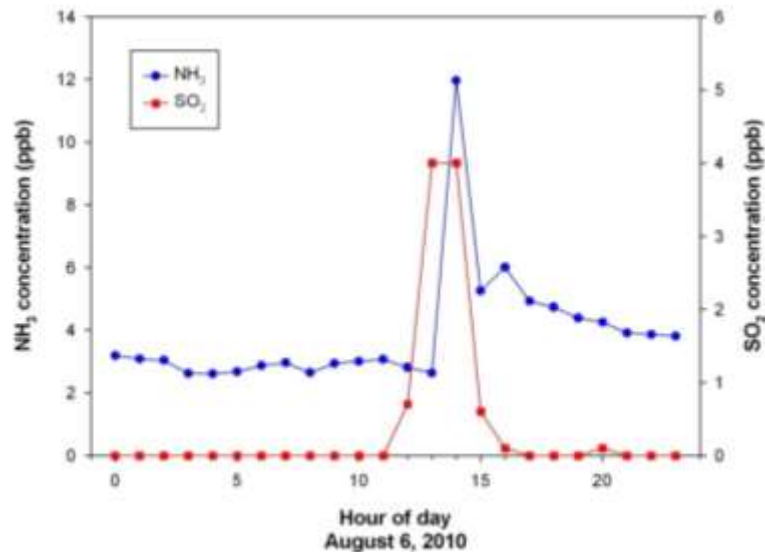
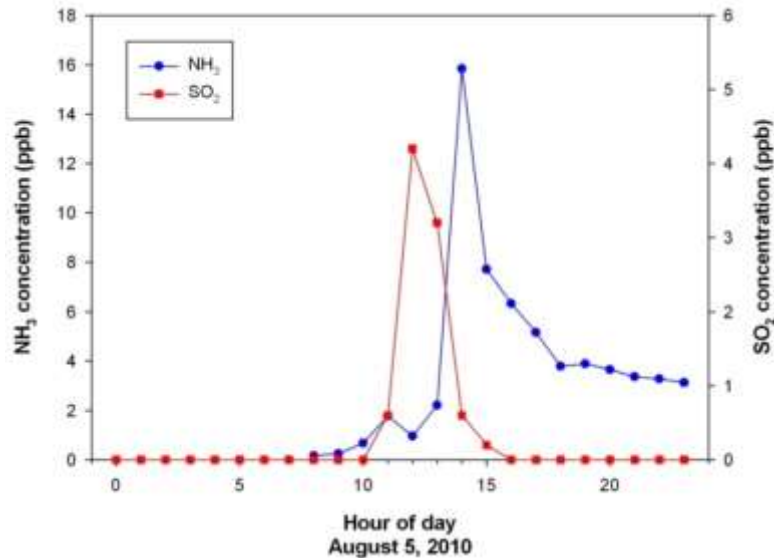
A chemical incident occurred at ~ 6 a.m. after two trucks collided on I-59. Both trucks caught fire. [www.chron.com]



photo: Public Domain / RJN2

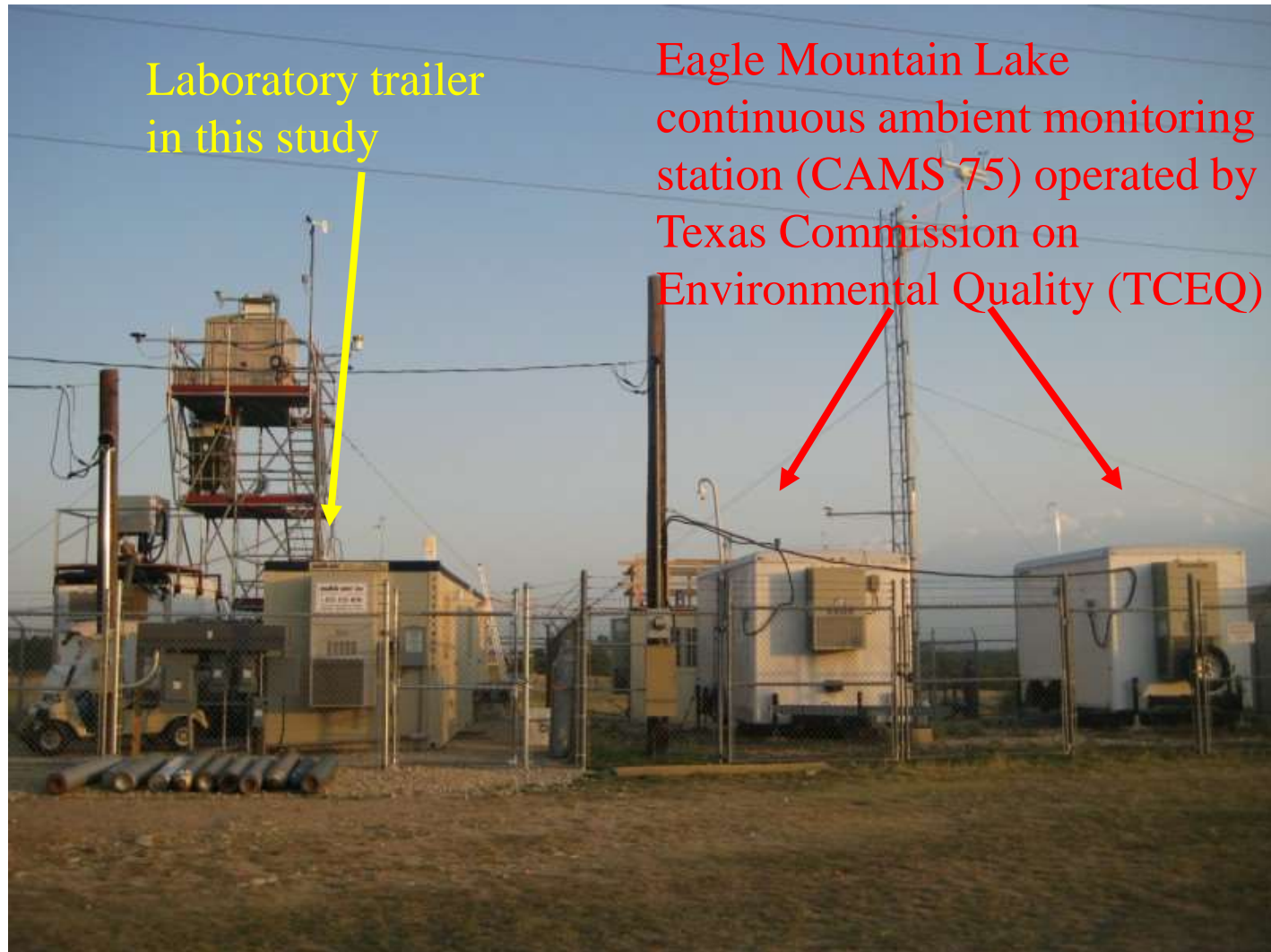
Estimated hourly NH₃ emission from the Houston Ship Channel area is about 0.25 ton. Mellqvist et al., (2007) Final Report, HARC Project H-53.

Sporadic increased NH_3 Concentration Levels related to Emissions by the Parish Electric Power Plant, TX



The Parish electric power plant is located near the Brazos River in Fort Bend County, Texas (~27 miles SW from downtown Houston)

Fort-Worth, Dallas(TX) CAMS 75 & TCEQ monitoring site

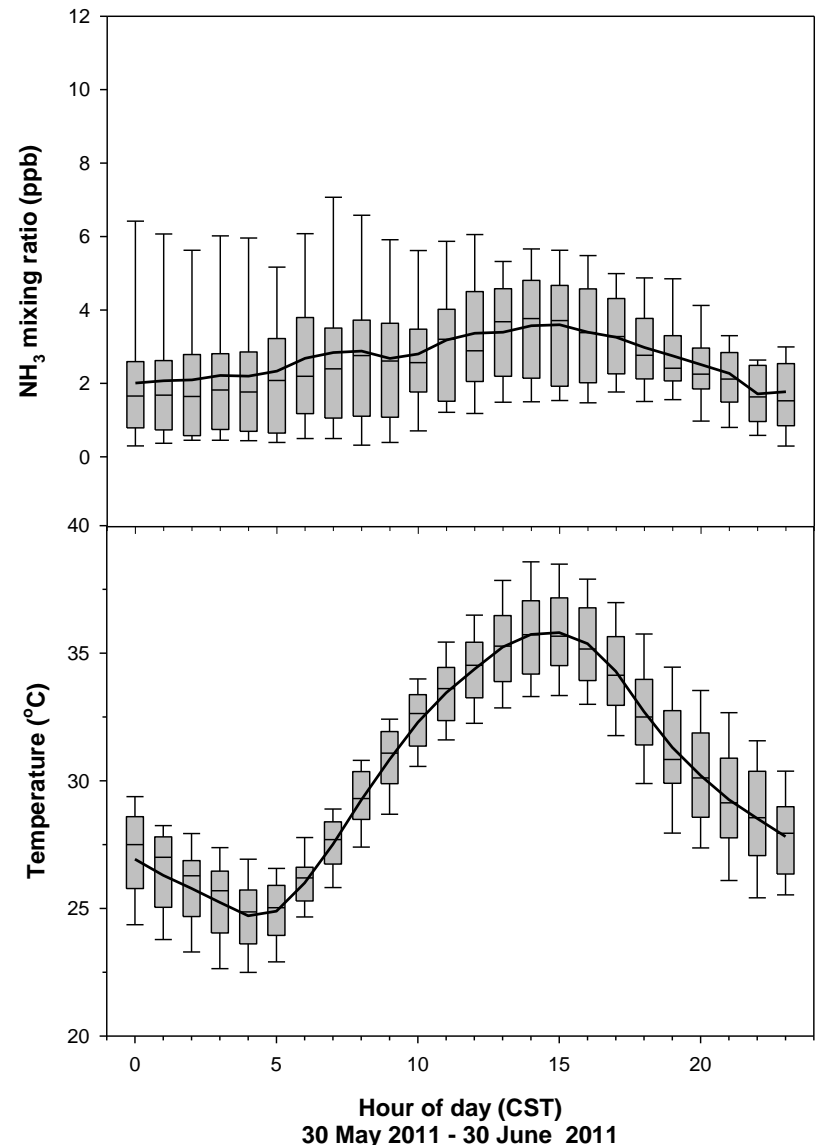


Instrumentation available at CAMS 75 & TCEQ monitoring site

Species/parameter	Measurement technique
NH ₃	Daylight Solutions External Cavity Quantum Cascade Laser (Photo-acoustic Spectroscopy)
CO	Thermo Electron Corp. 48C Trace Level CO Analyzer (Gas Filter Correlation)
SO ₂	Thermo Electron Corp. 43C Trace Level SO ₂ Analyzer (Pulsed Fluorescence)
NO _x	Thermo Electron Corp. 42C Trace Level NO-NO ₂ -NO _x Analyzer (Chemiluminescence)
NO _y	Thermo Electron Corp. 42C-Y NO _y Analyzer (Molybdenum Converter)
HNO ₃	Mist Chamber coupled to Ion Chromatography (Dionex, Model CD20-1)
HCl	Mist Chamber coupled to Ion Chromatography (Dionex, Model CD20-1)
VOC _s	IONICON Analytik Proton Transfer Reaction Mass Spectrometer and TCEQ Automated Gas Chromatograph
PBL height	Vaisala Ceilometer CL31 with updated firmware to work with Vaisala Boundary Layer View software
Temperature	Campbell Scientific HMP45C Platinum Resistance Thermometer
Wind speed	Campbell Scientific 05103 R. M. Young Wind Monitor
Wind direction	Campbell Scientific 05103 R. M. Young Wind Monitor

NH₃ source attribution & temperature variations

- Emission events from specified point sources (i.e., industrial facilities)
- Estimated NH₃ emissions from cows (1.3 tons/day)
- Estimated NH₃ emissions from soil and vegetation (0.15 tons/day)
- EPA PMF (biogenic: 74.1%; light duty vehicles: 12.1%; natural gas/industry: 9.4%; and heavy duty vehicles: 4.4%)
- Livestock might account for approximately 66.4% of total NH₃ emissions
- Increased contribution from industry (→ 18.9%)



From Conventional PAS to QEPAS

$Q \gg 1000$

Cell is **OPTIONAL!**

V-effective volume

Laser beam,
power P

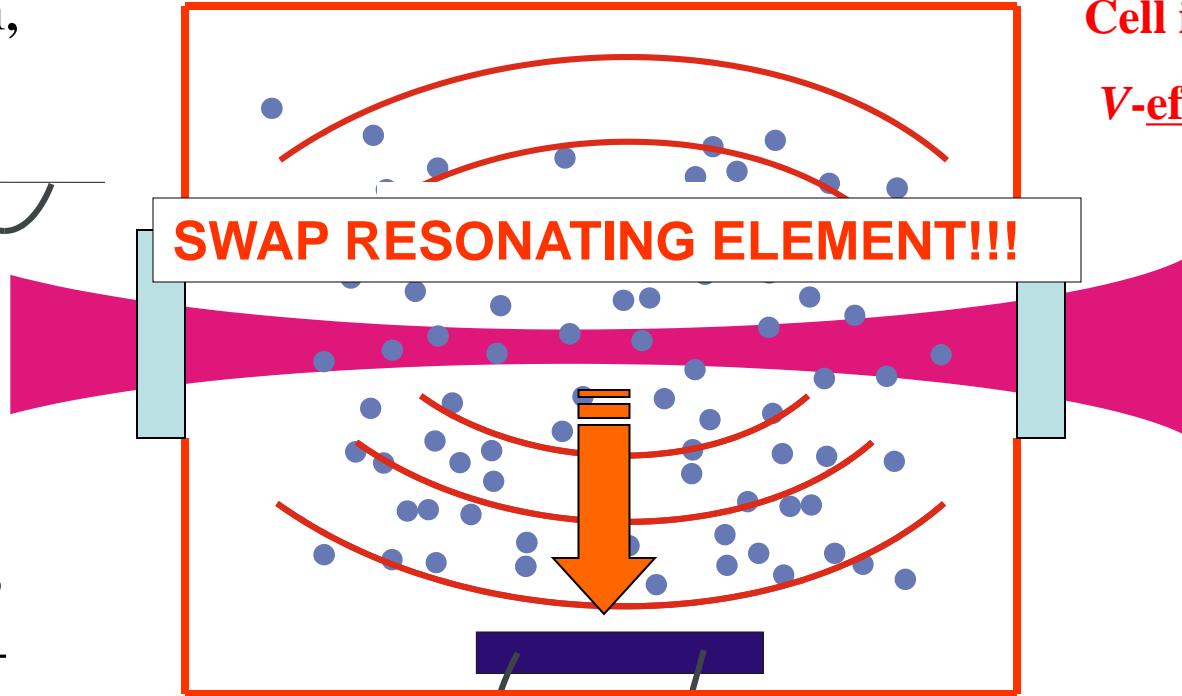


Modulated
(P or λ) at f
or $f/2$

$$S \sim \frac{Q \alpha P}{f V}$$

$$NNEA = \frac{\alpha_{\min} P}{\sqrt{\Delta f}} \left[\frac{\text{cm}^{-1} \times \text{W}}{\sqrt{\text{Hz}}} \right]$$

SWAP RESONATING ELEMENT!!!



**Piezoelectric
crystal**

**Resonant at f
quality factor Q**



RICE

Quartz Tuning Fork as a Resonant Microphone for QEPAS

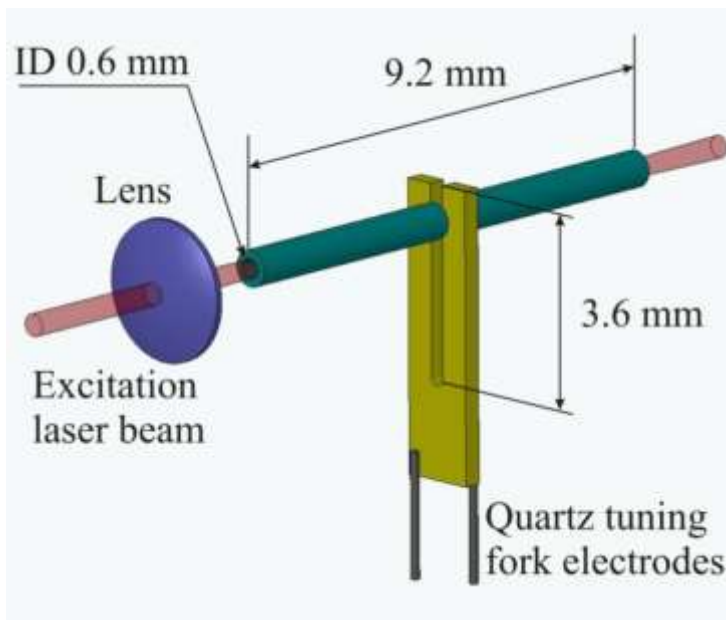


Unique properties

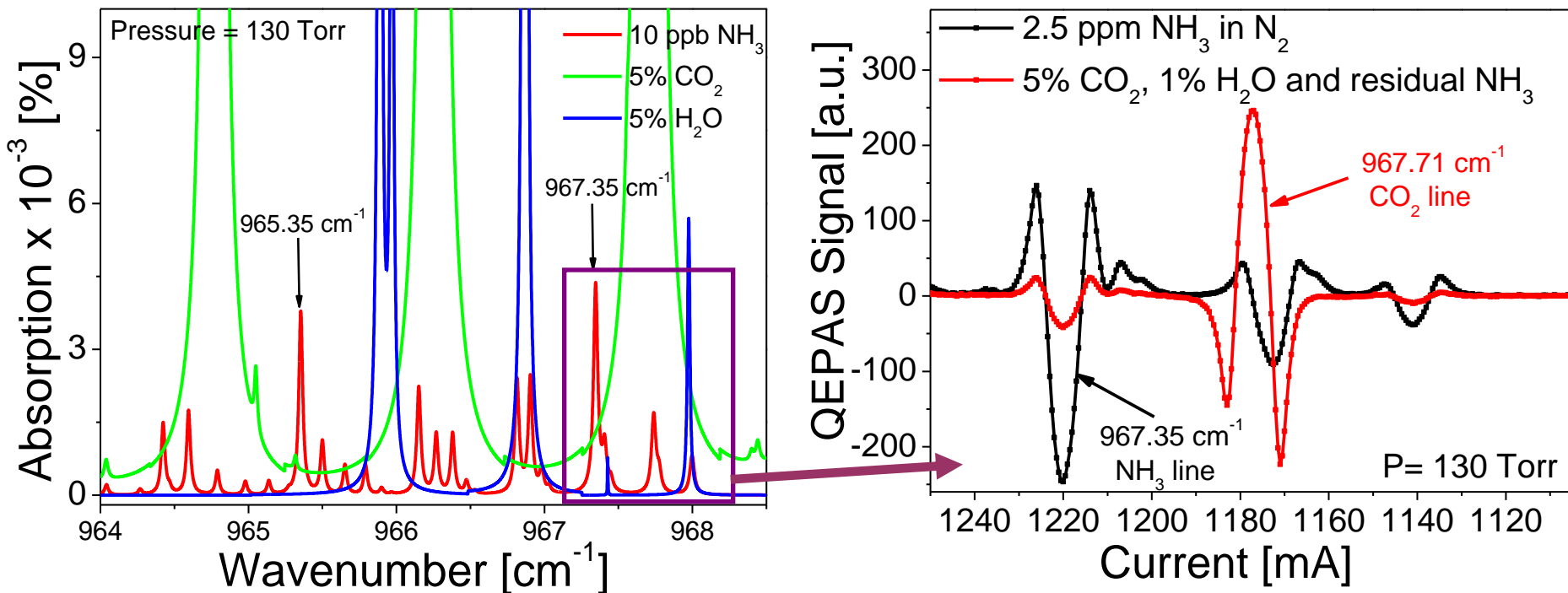
- Extremely low internal losses:
 - $Q \sim 10\,000$ at 1 atm
 - $Q \sim 100\,000$ in vacuum
- Acoustic quadrupole geometry
 - Low sensitivity to external sound
- Large dynamic range ($\sim 10^6$) – linear from thermal noise to breakdown deformation
 - 300K noise: $x \sim 10^{-11}$ cm
 - Breakdown: $x \sim 10^{-2}$ cm
- Wide temperature range: from 1.6K to ~ 700 K

Acoustic Micro-resonator (mR) tubes

- Optimum inner diameter: 0.6 mm; mR-QTF gap is 25-50 μm
- Optimum mR tubes must be ~ 4.4 mm long ($\sim \lambda/4 < l < \lambda/2$ for sound at 32.8 kHz)
- SNR of QTF with mR tubes: $\times 30$ (depending on gas composition and pressure)



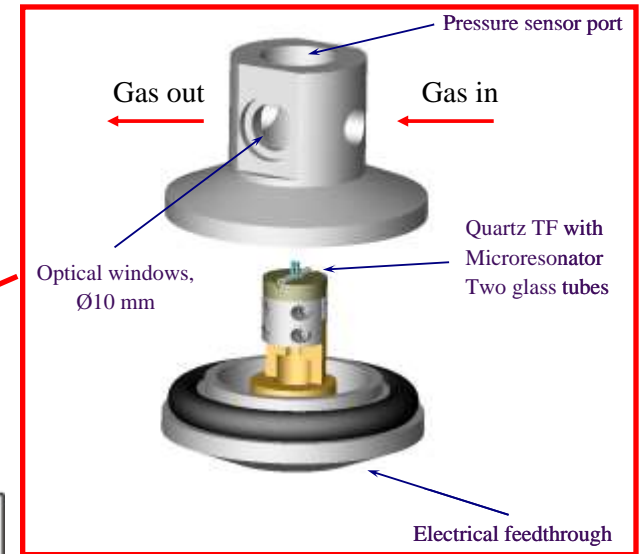
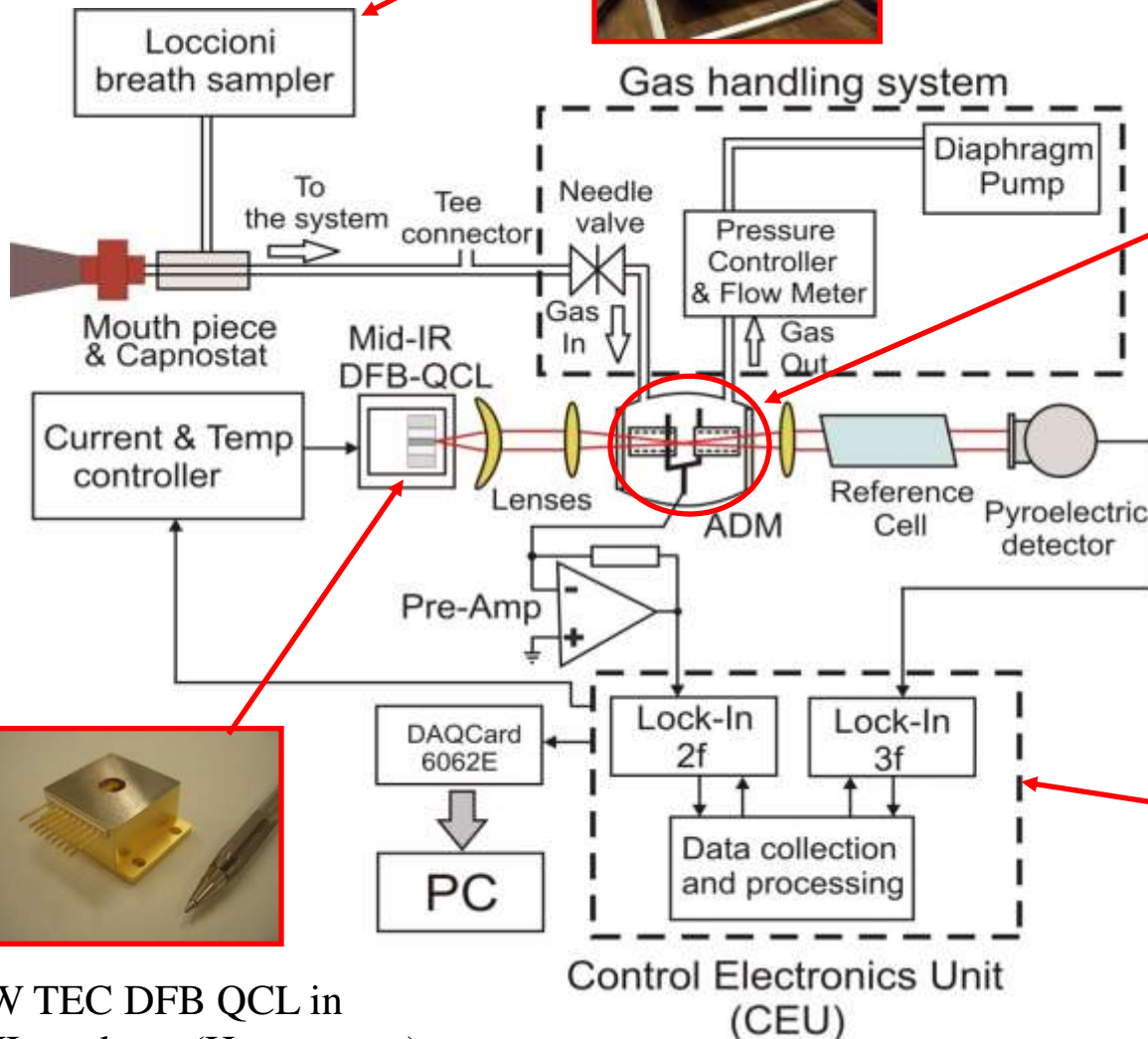
Optimum NH_3 Line Selection for a $10.34\text{ }\mu\text{m}$ CW TEC DFB QCL



Simulated HITRAN high resolution spectra @ 130 Torr indicating two NH_3 absorption lines of interest

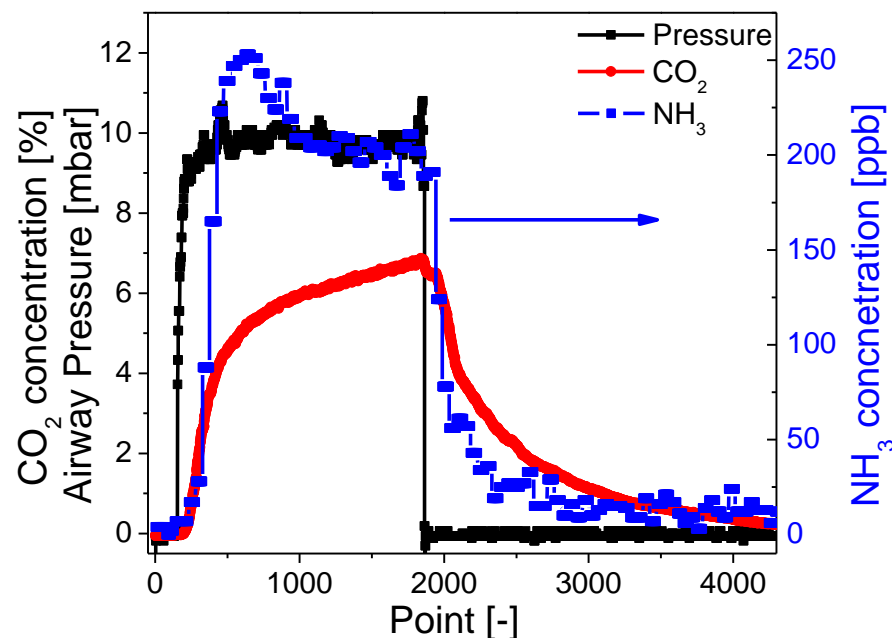
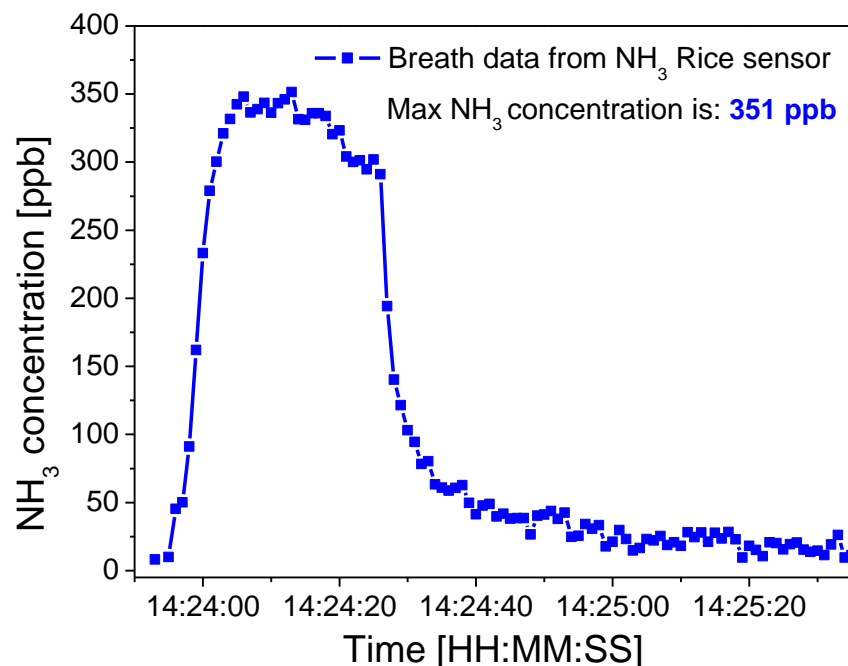
No overlap between NH_3 and CO_2 absorption lines was observed for the selected **967.35 cm^{-1}** NH_3 absorption line in the ν_2 R band.

QEPAS based NH_3 Gas Sensor Architecture



CW TEC DFB QCL in
HHL package (Hamamatsu)

Real-time exhaled human NH_3 Breath Measurements



Airway pressure (black), CO₂ (red), and NH₃ (blue) profiles of a single breath exhalation lasting 40sec.

Successful testing of a 2nd generation breath ammonia monitor installed in a clinical environment. (Johns Hopkins, Baltimore, MD and St. Luke's Hospital, Bethlehem, PA)

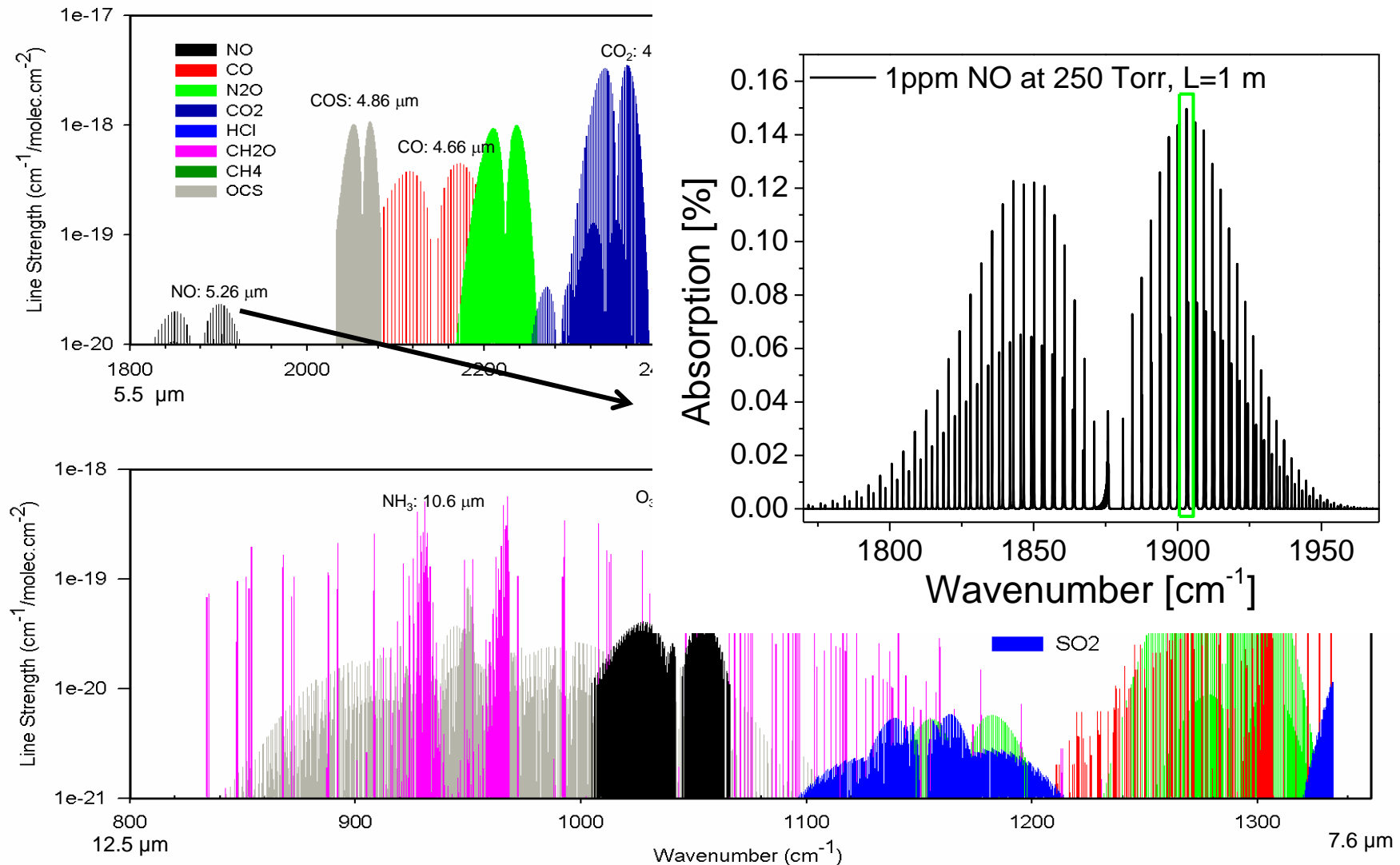
Minimum detectable concentration of NH_3 is:
~ 6 ppbv at 967.35 cm⁻¹ (1 σ ; 1 s time resolution)



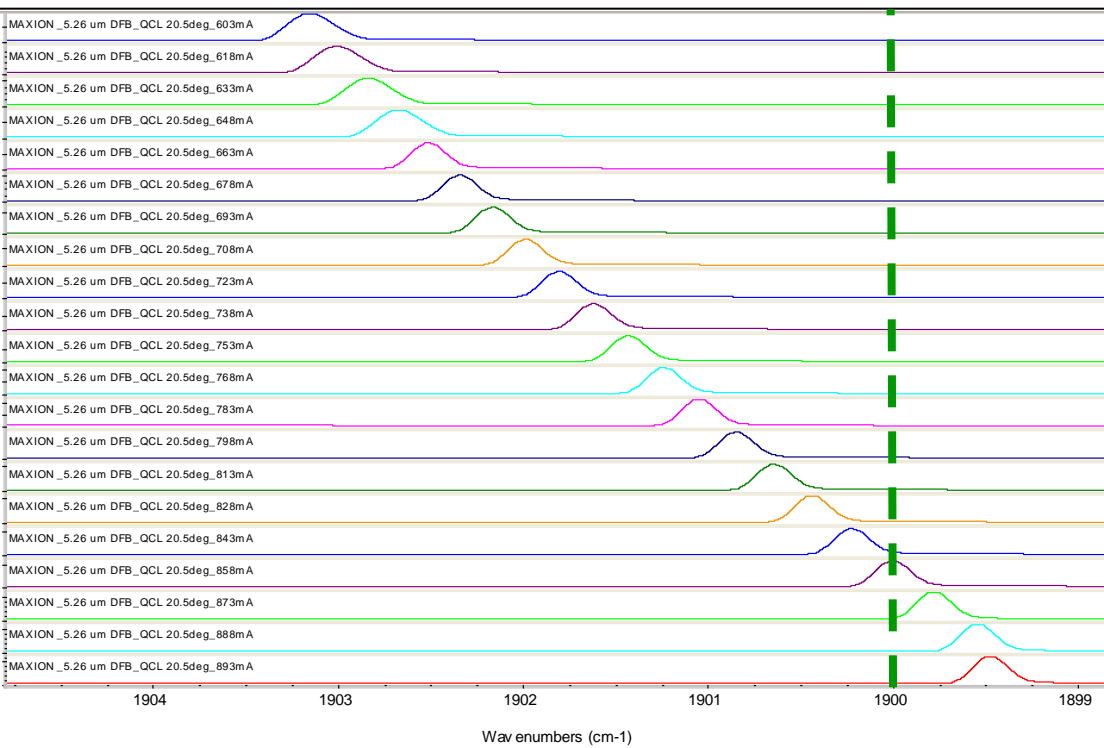
Motivation for Nitric Oxide Detection

- Atmospheric Chemistry
- Environmental pollutant gas monitoring
 - NO_x monitoring from automobile exhaust and power plant emissions
 - Precursor of smog and acid rain
- Industrial process control
 - Formation of oxynitride gates in CMOS Devices
- NO in medicine and biology
 - Important signaling molecule in physiological processes in humans and mammals (1998 Nobel Prize in Physiology/Medicine)
 - Treatment of asthma, COPD, acute lung rejection
- Photofragmentation of nitro-based explosives

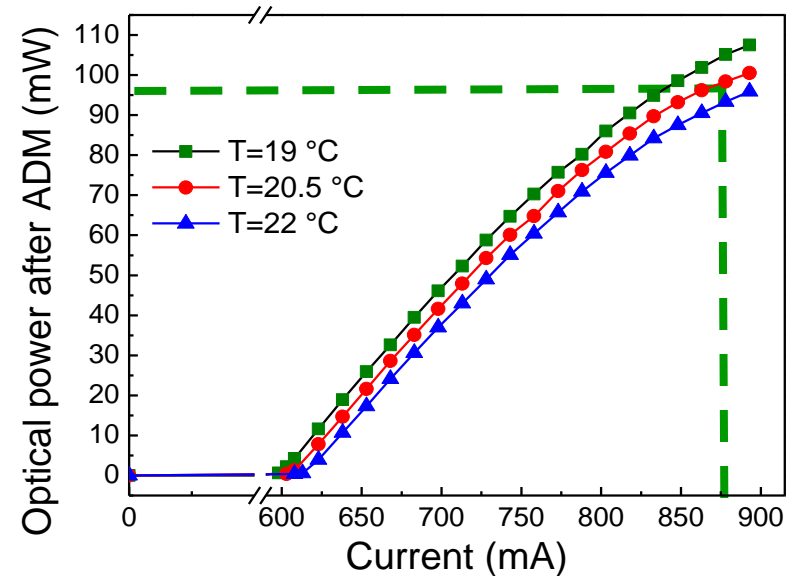
Molecular Absorption Spectra within two Mid-IR Atmospheric Windows and NO absorption @ 5.26 μm



Performance of a 5.26 μm CW HHL TEC DFB-QCL

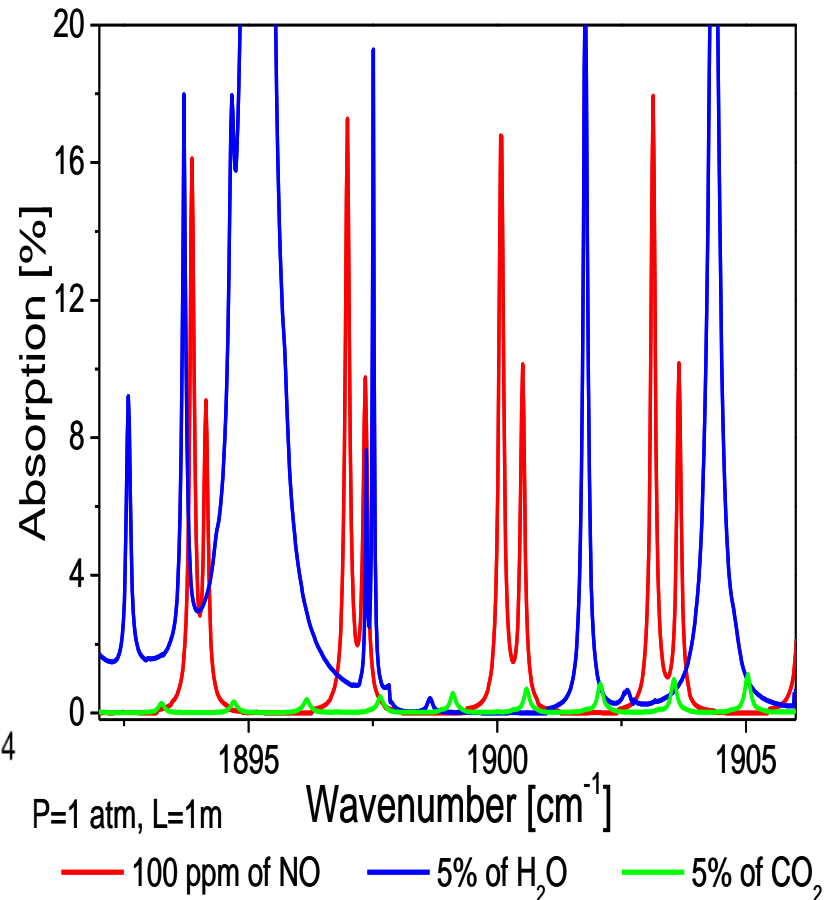
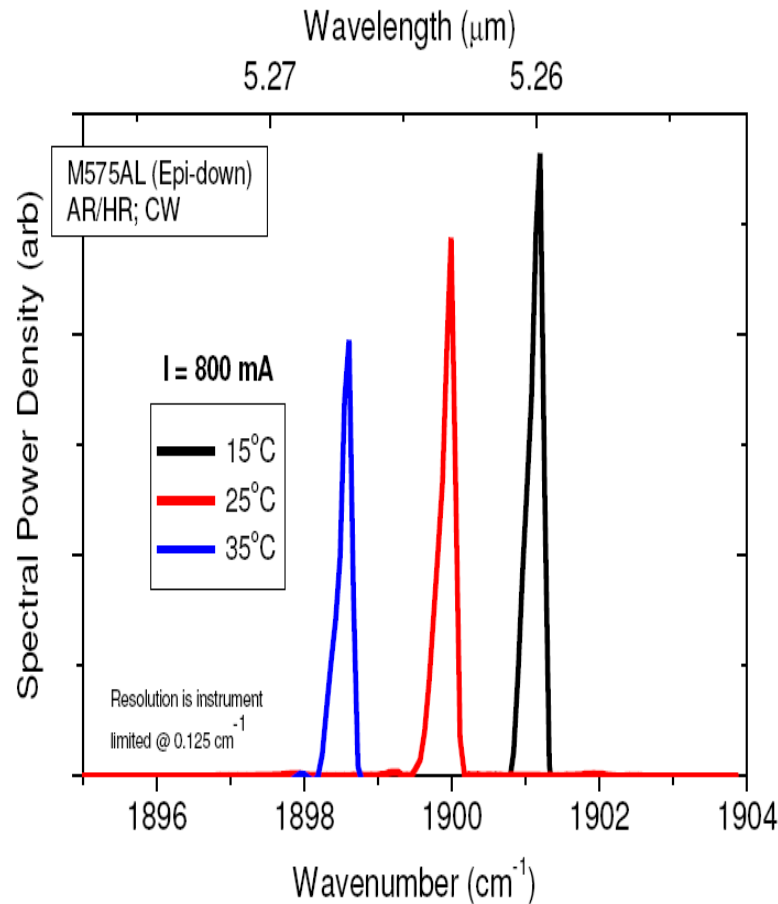


Single frequency QCL radiation recorded with FTIR for different laser current values at a QCL temperature of 20.5°C.



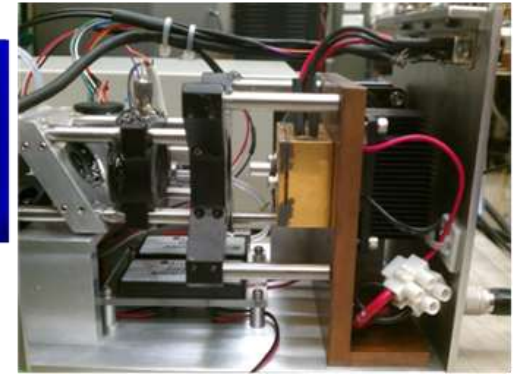
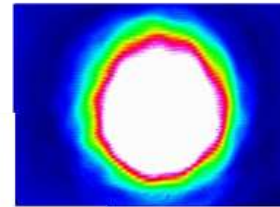
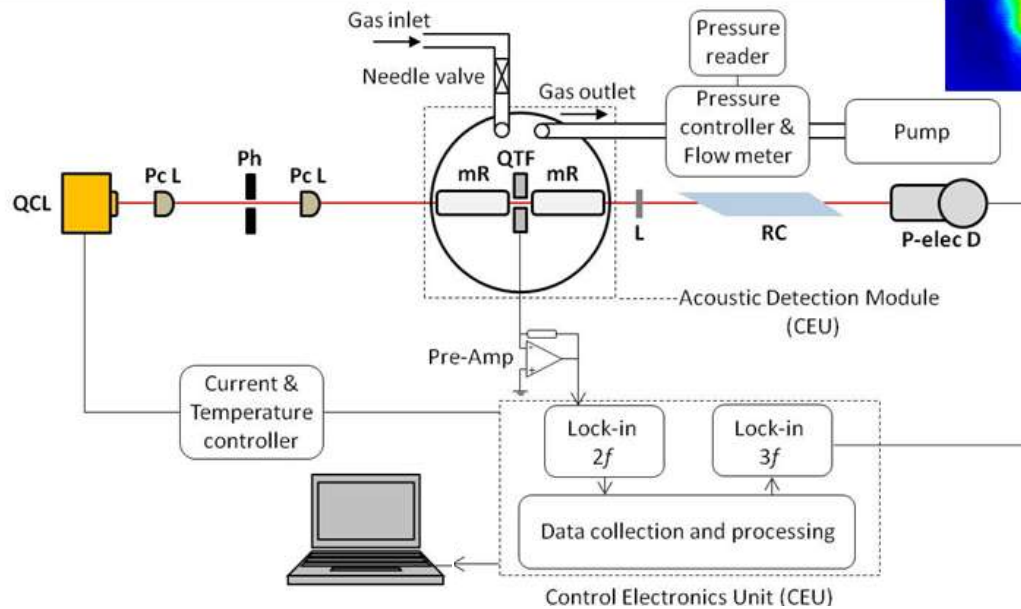
CW DFB-QCL optical power and current tuning at three different temperatures.

Emission spectra of a 1900cm^{-1} TEC CW DFB QCL and HITRAN Simulated spectra



Output power: 117 mW @ 25 C

CW TEC DFB QCL based QEPAS NO Gas Sensor



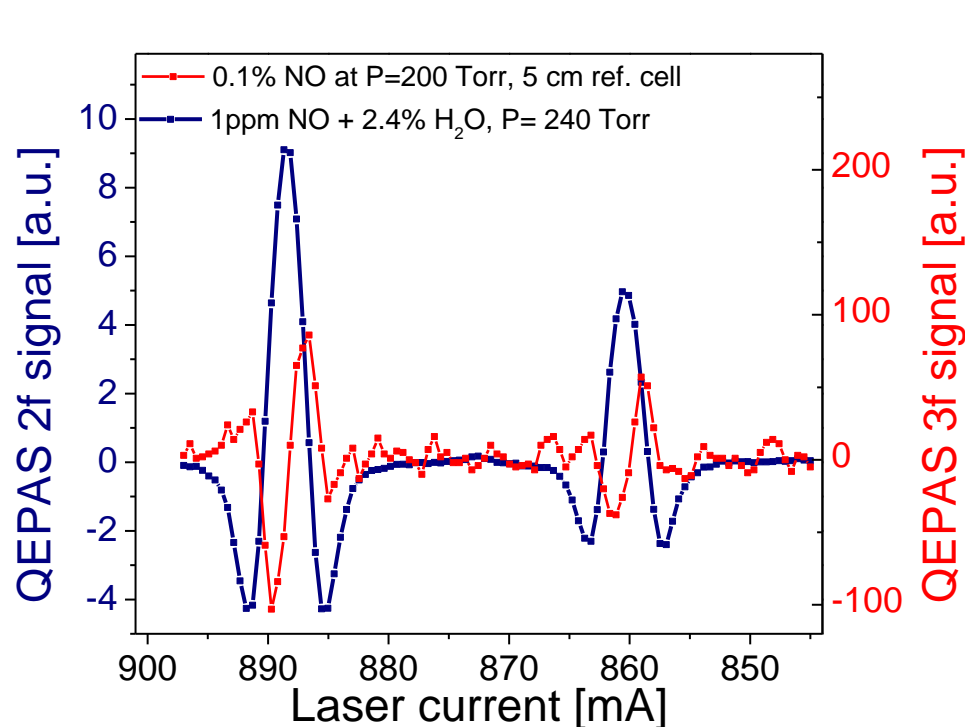
CW HHL TEC DFB-QCL package and IR camera image of the laser beam at 630 mA and 20.5 deg C through tubes after ADM

Schematic of a DFB-QCL based Gas Sensor.
PcL – plano-convex lens, Ph – pinhole,
QTF – quartz tuning fork, mR – microresonator,
RC- reference cell, P-elec D – pyro electric detector

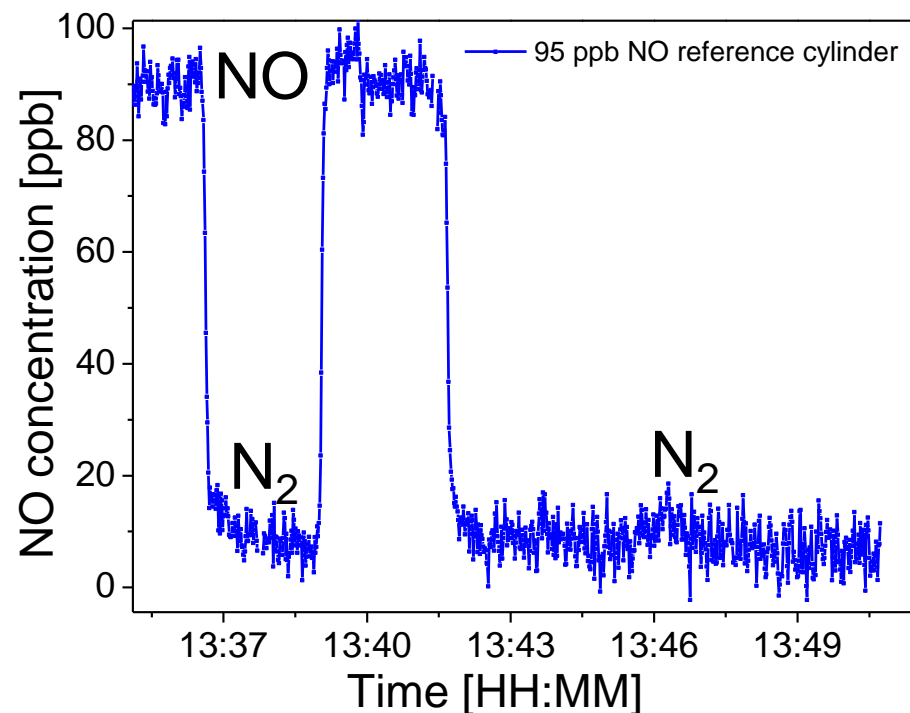


Compact Prototype NO Sensor
(September 2012)

Performance of CW DFB-QCL based WMS QEPAS NO Sensor Platform



2f QEPAS signal (navy) and reference 3f signal (red) when DFB-QCL was tuned across **1900.08 cm⁻¹** NO line.



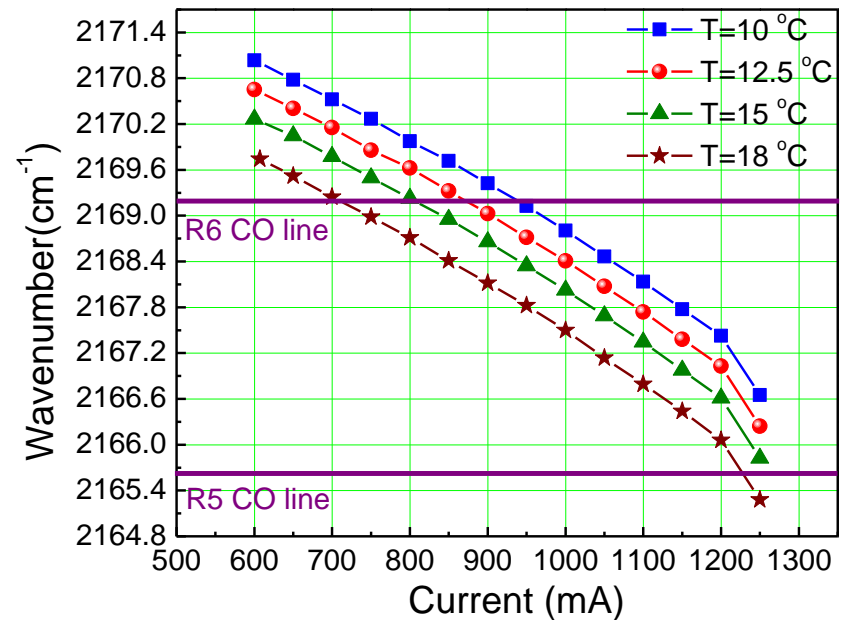
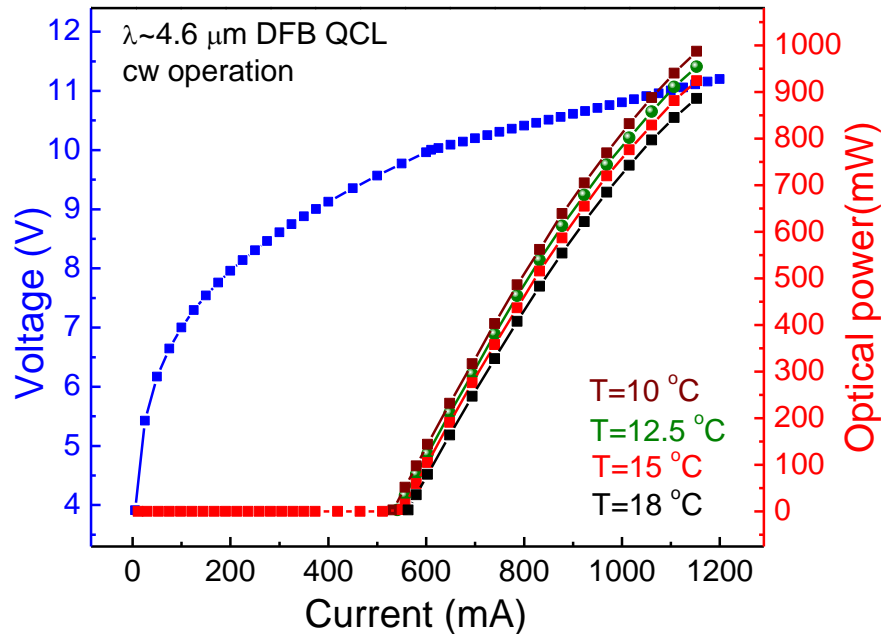
2f QEPAS signal amplitude for 95 ppb NO when DFB-QCL was locked to the **1900.08 cm⁻¹** line.

Minimum detectable NO concentration is:
~ 3 ppbv (1 σ ; 1 s time resolution)

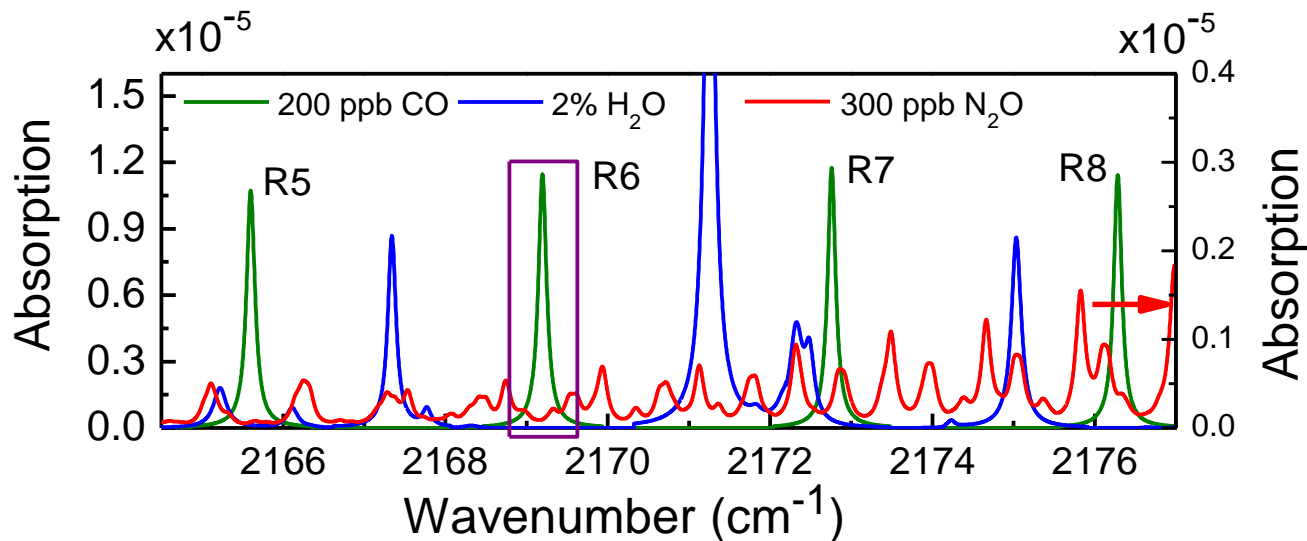
Motivation for Carbon Monoxide Detection

- Atmospheric Chemistry
 - Incomplete combustion of natural gas, fossil fuel and other carbon containing fuels.
 - Major global pollutant. Impact on atmospheric chemistry through its reaction with hydroxyl (OH) for troposphere ozone formation and changing the level of greenhouse gases (e.g. CH₄).
- Public Health
 - Extremely dangerous to human life even at a low concentrations. Therefore CO must be carefully monitored at low concentration levels.
- CO in medicine and biology
 - Hypertension, neurodegenerations, heart failure and inflammation have been linked to abnormality in CO metabolism.

Performance of a NWU 4.61 μm high power CW TEC DFB QCL

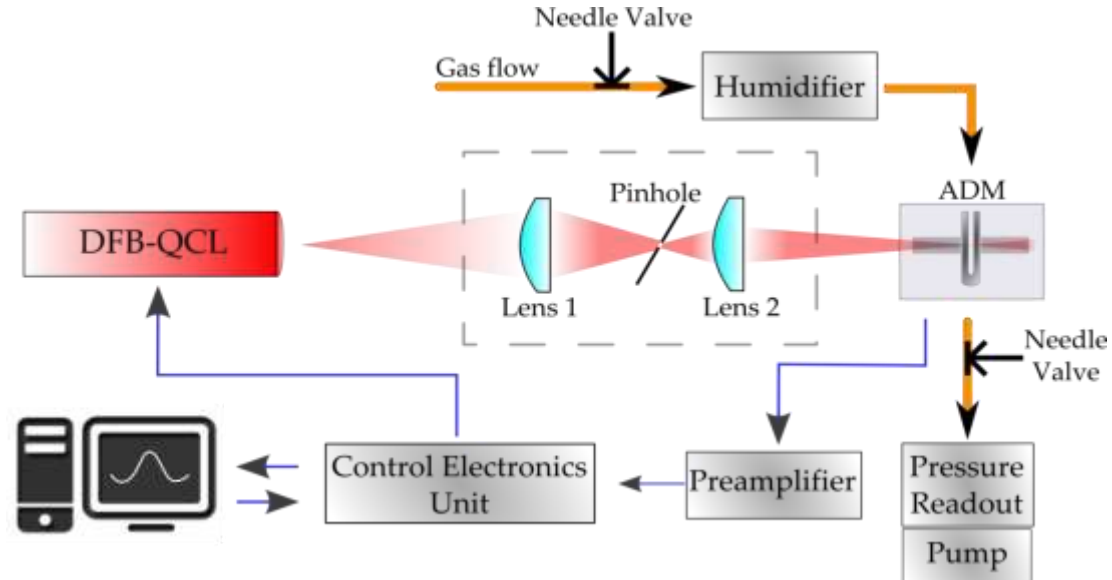


CW DFB-QCL optical power and current tuning at a four different QCL temperatures.



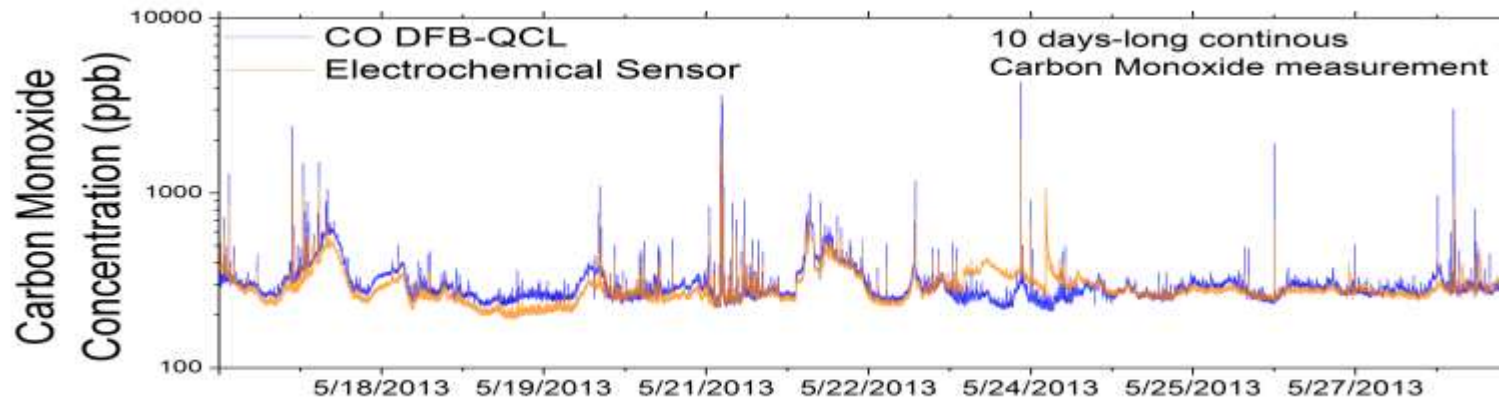
Estimated max wall-plug efficiency (WPE) is $\sim 7\%$ at 1.25A QCL drive-current.

Performance of a NWU 4.61 μm high power CW TEC DFB QCL

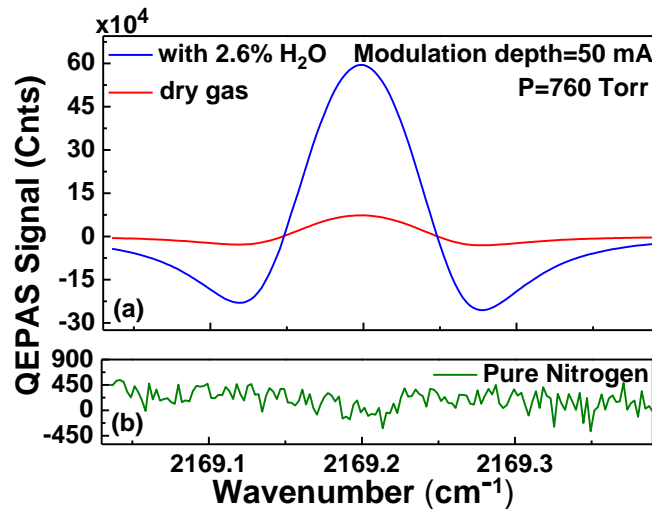


- CO sensor is enclosed in a 6" x 14" x 8" case
- Each 2f scan is completed in $\sim 5\text{s}$, when operating QCL in a frequency scanning mode
- QCL operating temperature is set to 10°C
- Sensor operates at a pressure of **225 Torr**, which is optimal in terms of signal-to-noise ratio

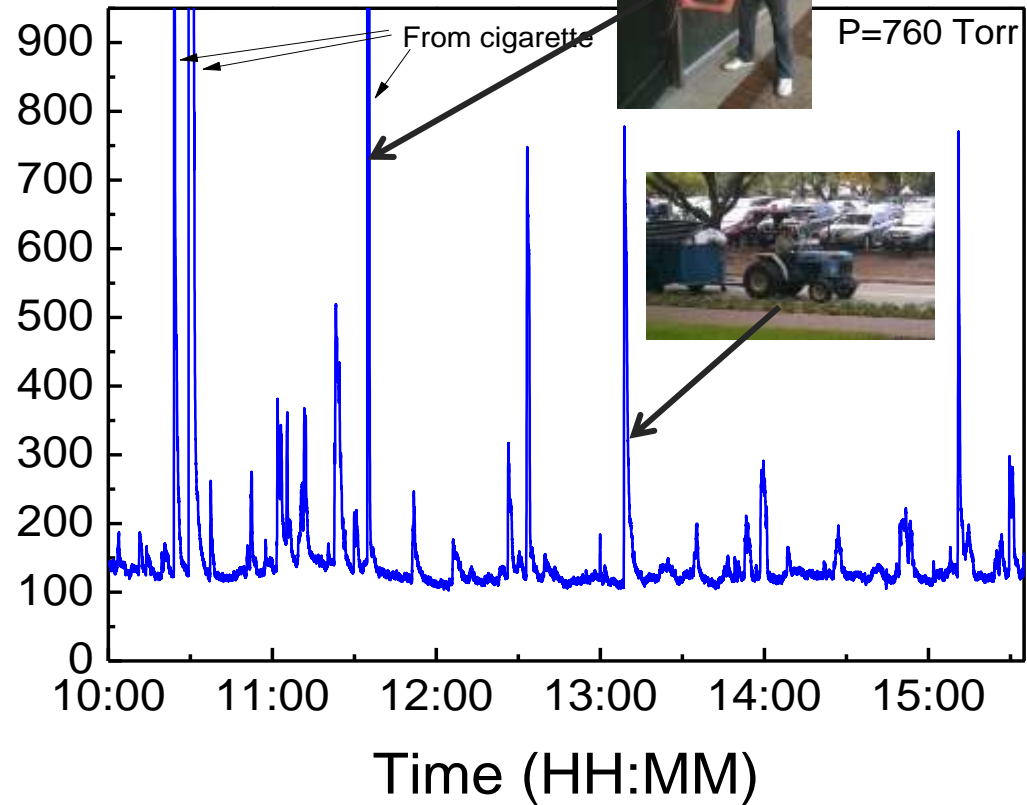
10 days-long continuous measurements were performed to determine CO concentration levels on Rice University



CW DFB-QCL based CO QEPAS Sensor Results

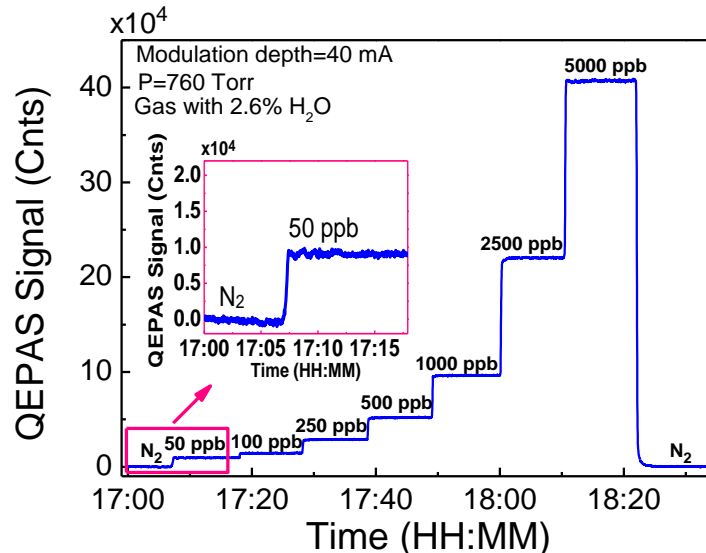


CO concentration (ppb)



Atmospheric CO concentration levels on Rice University campus, Houston, TX

Minimum detectable CO concentration is:
~ 2.5 ppbv (1 σ ; 1 s time resolution)



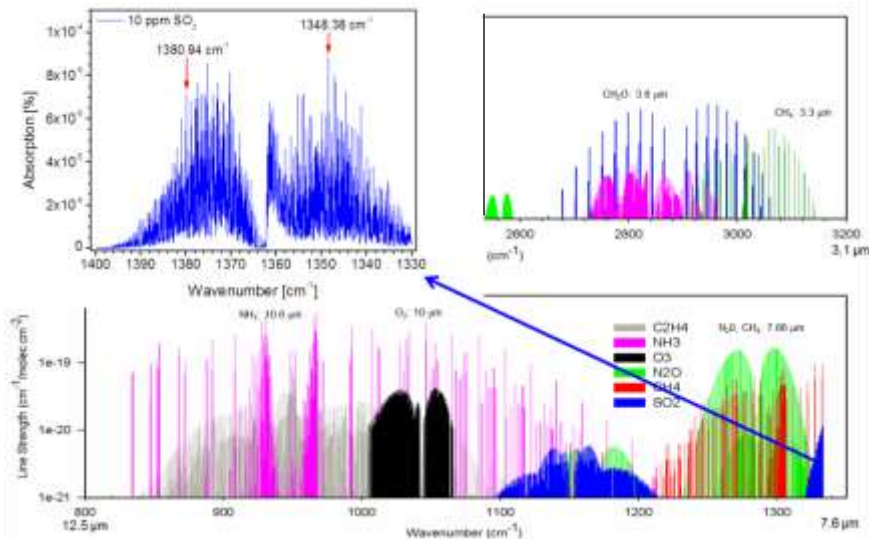
Dilution of a 5 ppm CO reference gas mixture when the CW DFB-QCL is locked to the 2169.2 cm⁻¹ R6 CO line.

CW DFB-QCL based SO₂ QEPAS Results

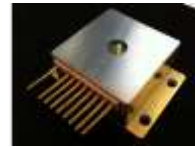
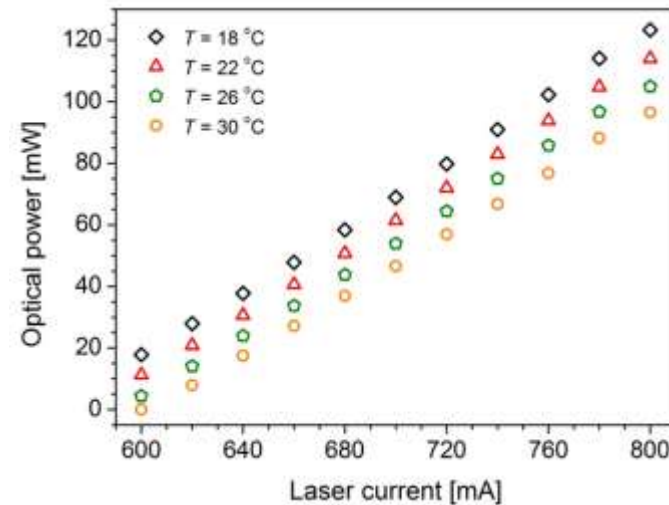
Basic Facts & Motivation of Sulfur Dioxide Detection

- Prominent air pollutant
- Emitted from coal fired power plants (~73%) and other industrial facilities (~20%)
- In atmosphere SO₂ converts to sulfuric acid primary contributor to acid rain
- SO₂ reacts to form sulfate aerosols
- Primary SO₂ exposure for 1 hour is 75 ppb
- SO₂ exposure affects lungs and causes breathing difficulties
- Currently, reported annual average atmospheric SO₂ concentrations range from ~ 1 - 6 ppb
- Major Sources are motor vehicles exhaust, fuel combustion, and fires.

SO₂ Absorption Line Selection

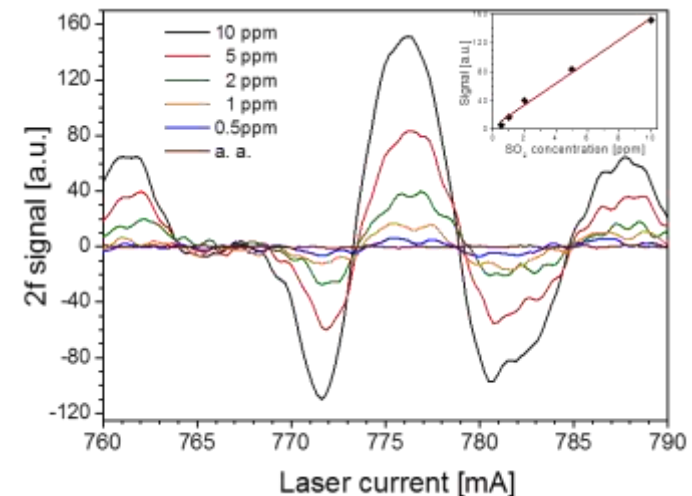


Molecular Absorption Spectra within two Mid-IR Atmospheric Windows



HHL CW DFB QCL

7.24 μm CW DFB-QCL optical power and current tuning at three different operating temperatures.

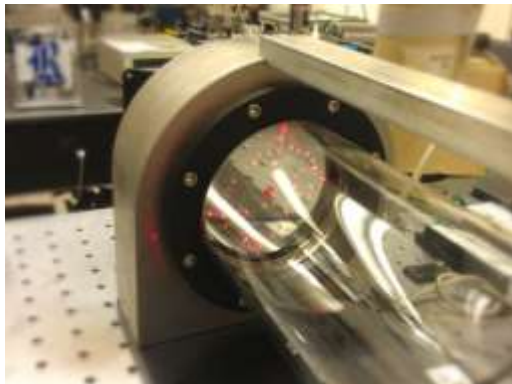
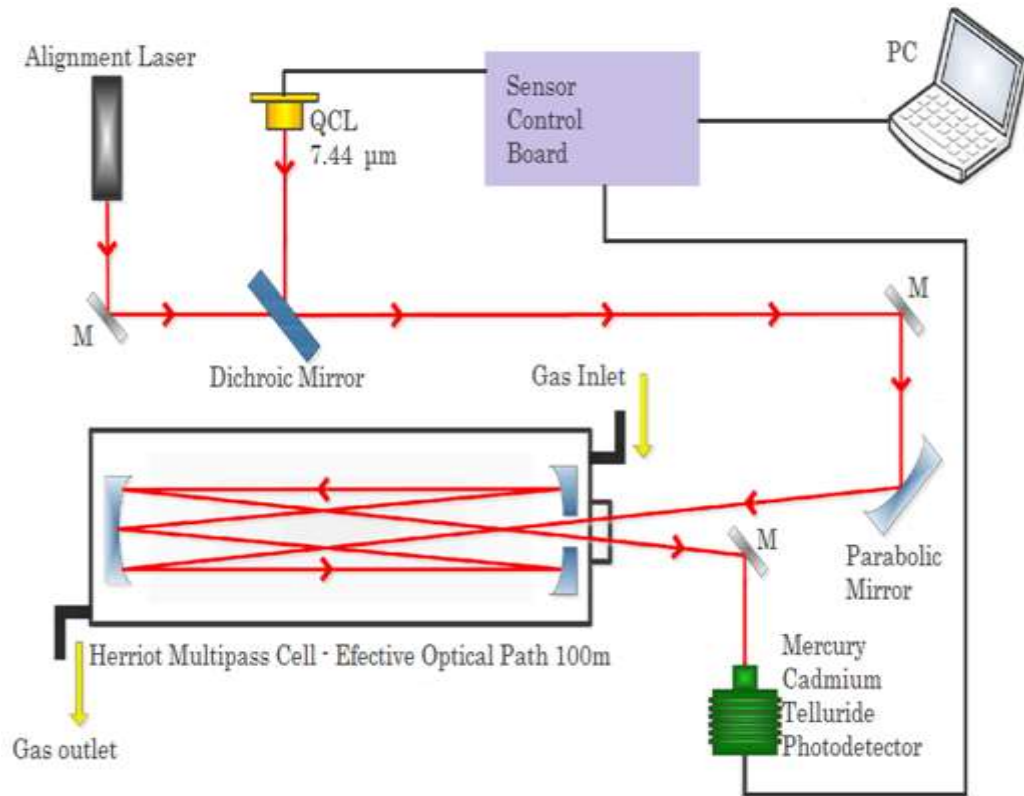


2f WMS QEPAS signals for different SO₂ concentrations when laser was tuned across 1380.9 cm⁻¹ line.

Minimum detectable SO₂ concentration is:

~ 100 ppbv (1σ; 1 s time resolution)

Comparison of SO₂ sensor performance using different techniques



Red beam spot pattern for a 100-m path length Herriott multipass (192 passes)

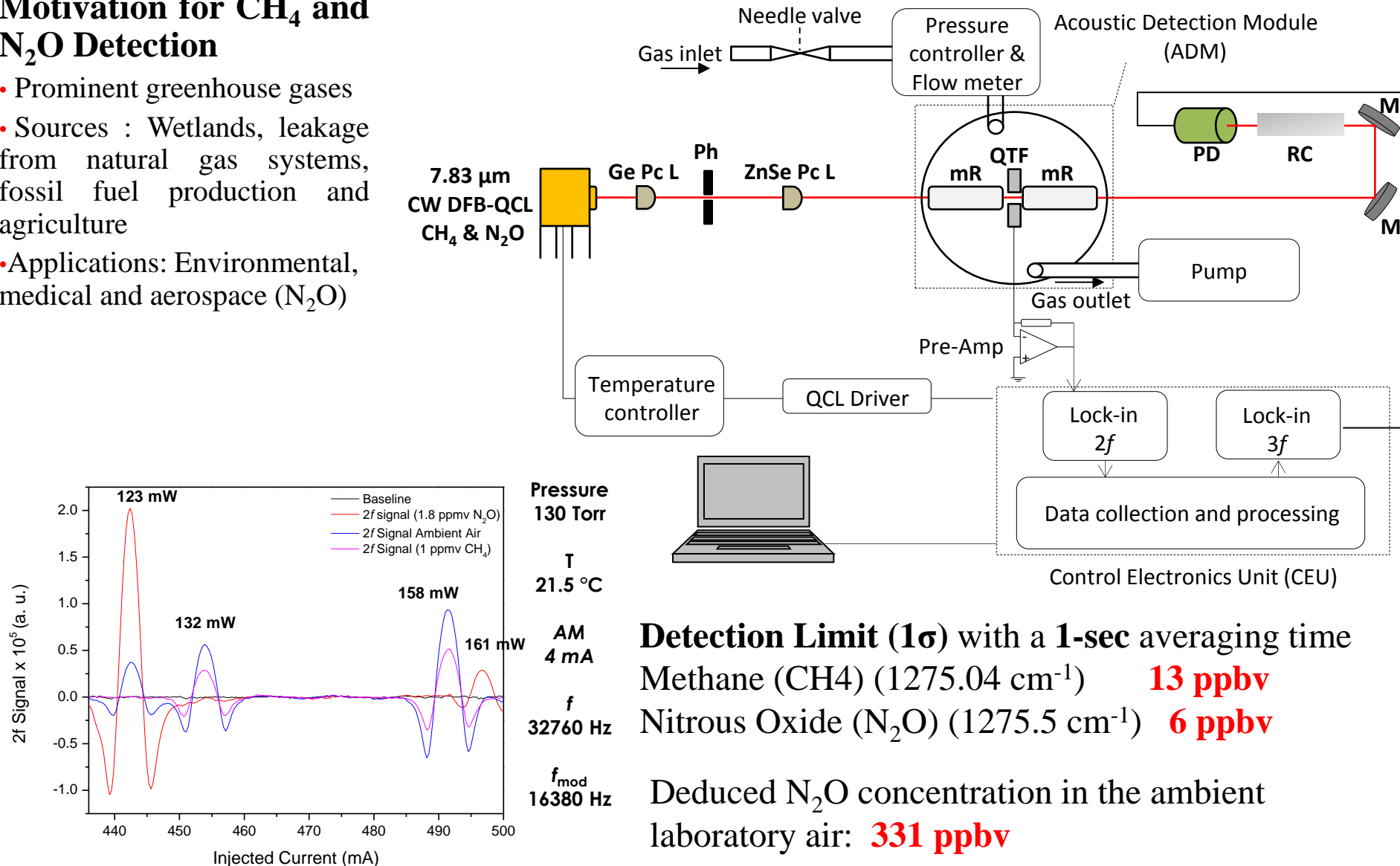
- Direct absorption open path 2m ~ 1 ppmv[1]
- QEPAS with 180mW laser power ~100 ppbv[2]
- Aerodyne WMS MPC system based on a 76m MPC ~0.5 ppb (projected with HITRAN simulation) [3]
- WMS MPC sensor system ~45 ppbv [4]

1. Wilson T. *et al.*
2. Wacławek J. *et al.*
3. Rodrigo J. *et al.*
4. Tarka J. *et al.*

QEPAS based CH₄ and N₂O Gas Sensor

Motivation for CH₄ and N₂O Detection

- Prominent greenhouse gases
- Sources : Wetlands, leakage from natural gas systems, fossil fuel production and agriculture
- Applications: Environmental, medical and aerospace (N₂O)



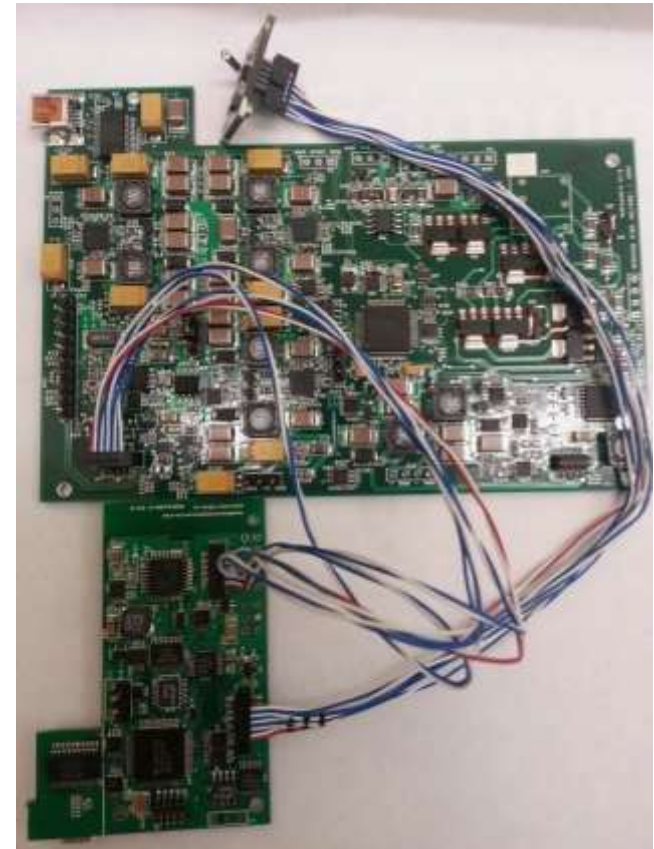
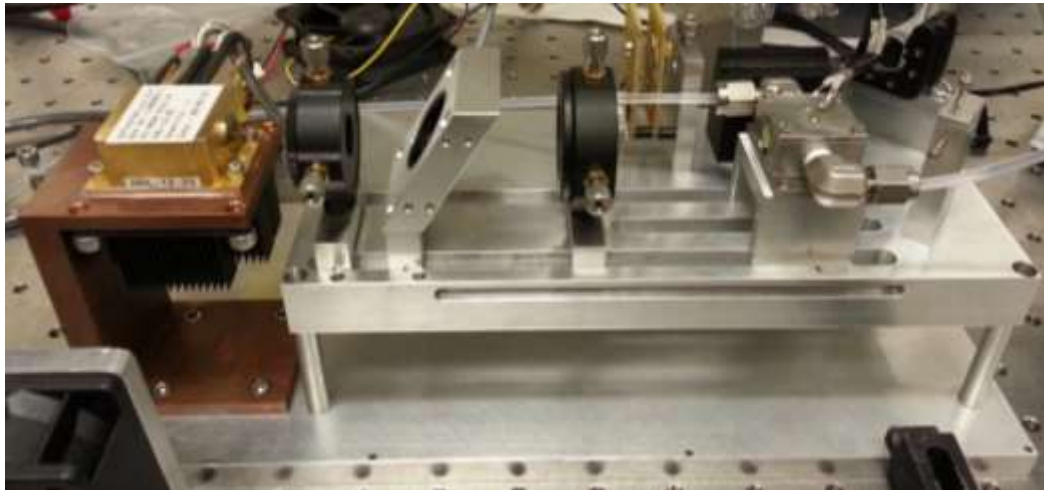
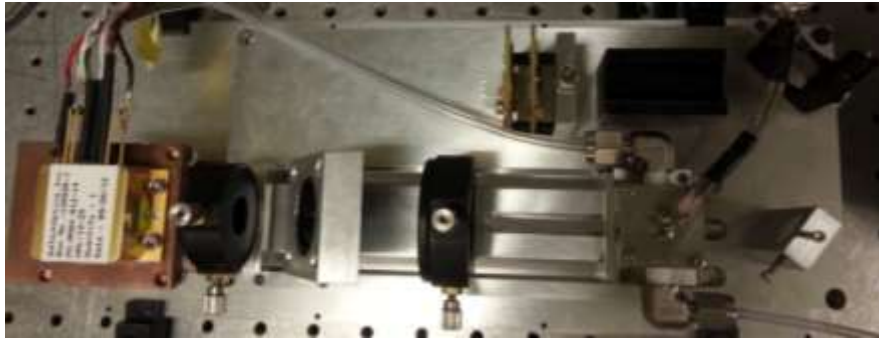
Detection Limit (1σ) with a 1-sec averaging time

Methane (CH₄) (1275.04 cm⁻¹) **13 ppbv**

Nitrous Oxide (N₂O) (1275.5 cm⁻¹) **6 ppbv**

Deduced N₂O concentration in the ambient laboratory air: **331 ppbv**

QEPAS based CH₄ and N₂O Gas Sensor



QEPAS Sensor Control Board

QCL Current and TEC Driver, Performing wavelength modulation, Data acquisition, Applying continuous saw-tooth current ramping at 8 Hz, Testing QTF and low noise pre amplifier

QCL based QEPAS Performance for 10 Trace Gas Species (June 19 2013)

Molecule (carrier gas)	Frequency cm ⁻¹	Pressure Torr	NNEA cm ⁻¹ W/Hz ^{1/2}	QCL Power mW	NEC (τ=1s) ppbV
CH₂O (N₂:75% RH)*	2804.90	75	8.7×10 ⁻⁹	7.2	120
CO (N₂+ 2.2% H₂O)*	2176.28	100	1.57×10 ⁻⁸	71	2
CO (propylene)	2196.66	50	7.4×10 ⁻⁸	6.5	140
N₂O (air+5%SF₆)	2195.63	50	1.5×10 ⁻⁸	19	7
N₂O (N₂+2.37%H₂O)	2201.75	200	2.9×10 ⁻⁸	70	2.5
C₂H₅OH (N₂)**	1934.2	770	2.2×10 ⁻⁷	10	9×10 ⁴
NO (N₂+H₂O)	1900.07	250	7.5×10 ⁻⁹	100	3.6
SO₂ (N₂+2.4%H₂O)	1380.94	100	2.0×10 ⁻⁸	40	100
N₂O (air)	1275.49	230	5.3×10 ⁻⁸	100	30
CH₄ (air)	1275.39	230	1.7×10 ⁻⁷	100	118
C₂HF₅ (N₂)***	1208.62	770	7.8×10 ⁻⁹	6.6	9
NH₃ (N₂)*	1046.39	110	1.6×10 ⁻⁸	20	6
SF₆***	943.73	75	2.7×10 ⁻¹⁰	40	5×10 ⁻²

* - Improved microresonator

** - Improved microresonator and double optical pass through ADM

*** - With amplitude modulation and metal microresonator

NNEA – normalized noise equivalent absorption coefficient.

NEC – noise equivalent concentration for available laser power and τ=1s time constant, 18 dB/oct filter slope.

For comparison: conventional PAS 2.2 (2.6)×10⁻⁹ cm⁻¹W/√Hz (1,800; 10,300 Hz) for NH₃*, ()**

* M. E. Webber et al, Appl. Opt. 42, 2119-2126 (2003); ** J. S. Pilgrim et al, SAE Intl. ICES 2007-01-3152; *** - V. Spagnolo, et al. University and Politecnico of Bari, Italy



Merits of QEPAS based Trace Gas Detection

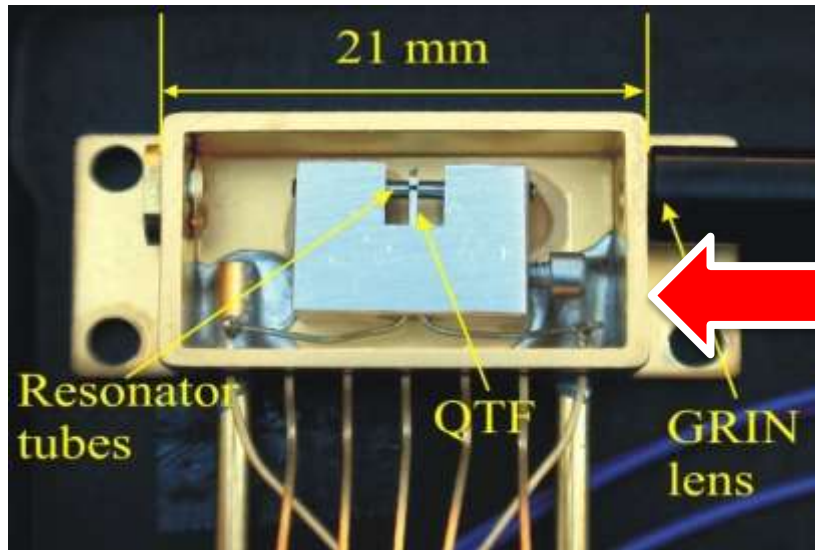
- Very small sensing module and sample volume (a few mm³ to ~2cm²)
- Extremely low dissipative losses
- Optical detector is not required
- Wide dynamic range
- Frequency and spatial selectivity of acoustic signals
- Rugged transducer – quartz monocrystal; can operate in a wide range of pressures and temperatures
- Immune to environmental acoustic noise, sensitivity is limited by the fundamental thermal TF noise: $k_B T$ energy in the TF symmetric mode
- Absence of low-frequency noise: SNR scales as \sqrt{t} , up to $t=3$ hours as experimentally verified

QEPAS: some challenges

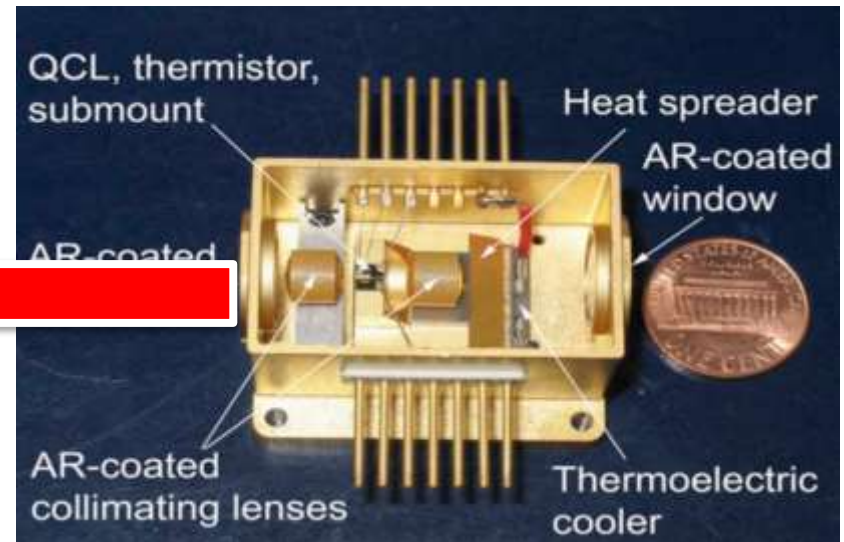
- Cost of Spectrophone assembly
- Sensitivity scales with laser power
- Effect of H₂O
- Responsivity depends on the speed of sound and molecular energy transfer processes
- Cross sensitivity issues



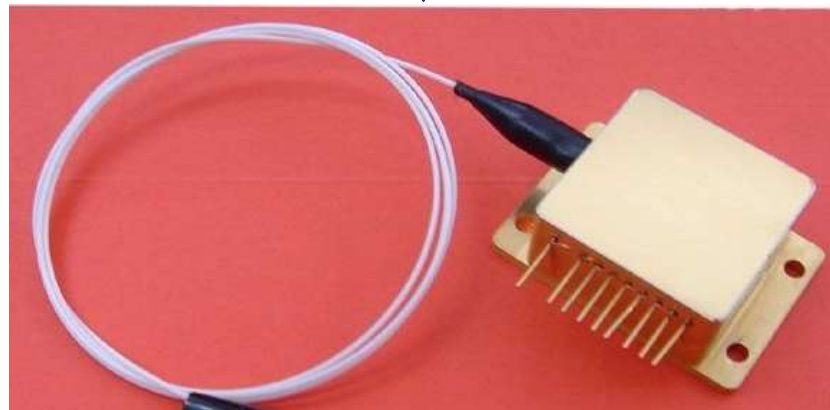
Potential Integration of a CW DFB- QCL and QEPAS Absorption Detection Module



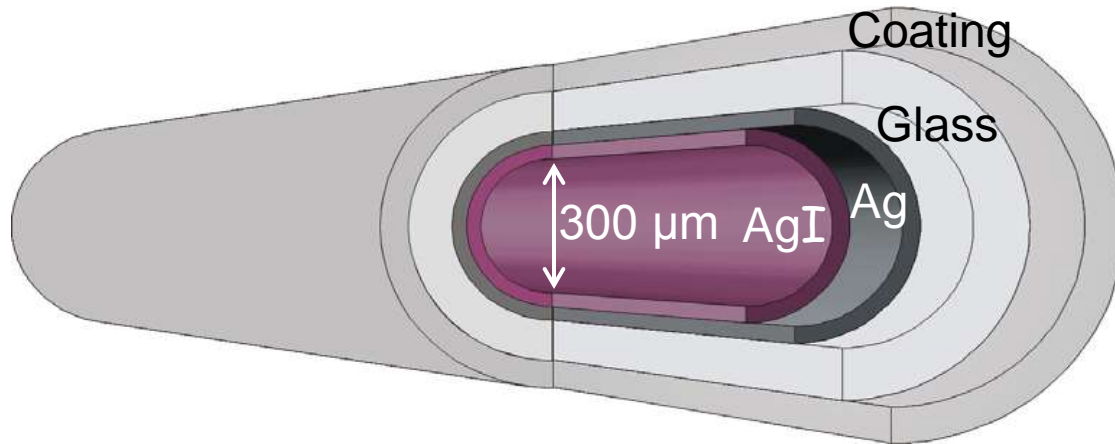
2012 QEPAS ADM



HHL package fiber coupled DFB-QCL

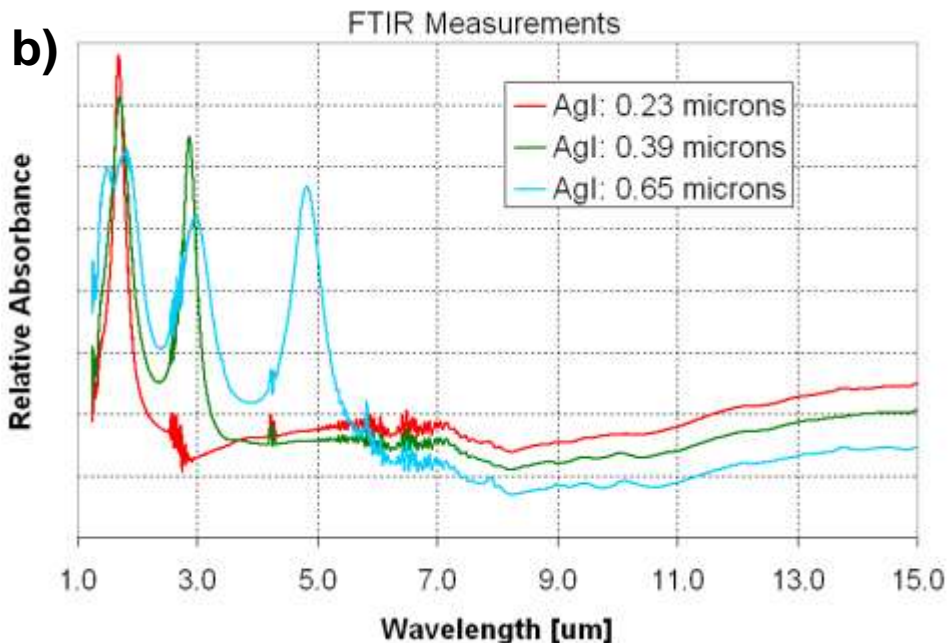


Hollow core waveguide



Hollow Core Glass Waveguides:

- Excellent Infrared transmission out to 20 μm
- Proven single mode delivery for bore size $\sim 30\lambda$
- No end reflections
- High damage threshold
- Very Robust
- 20+ years of experience at Rutgers

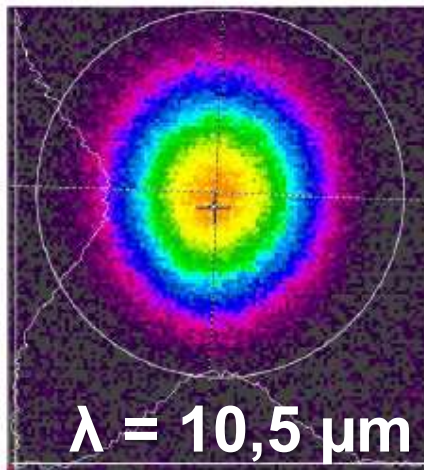


Bending loss is the primary concern

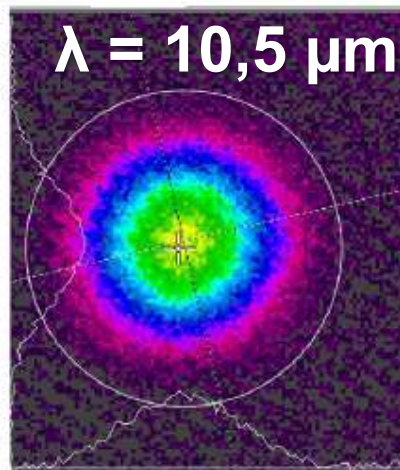
QCL-fiber beam profile and losses



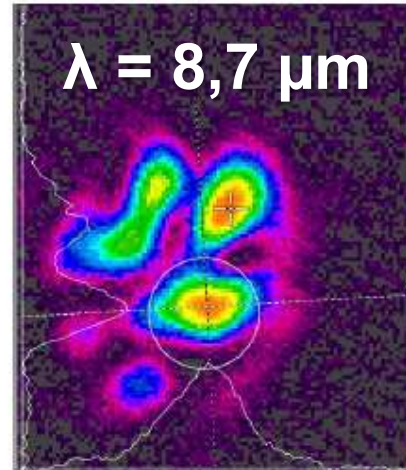
**HWG Fiber with
300 μm bore size
allows single mode
beam delivery @
10,5 μm**



(a) No fiber



(b) Single-mode fiber



(c) Multi-mode fiber

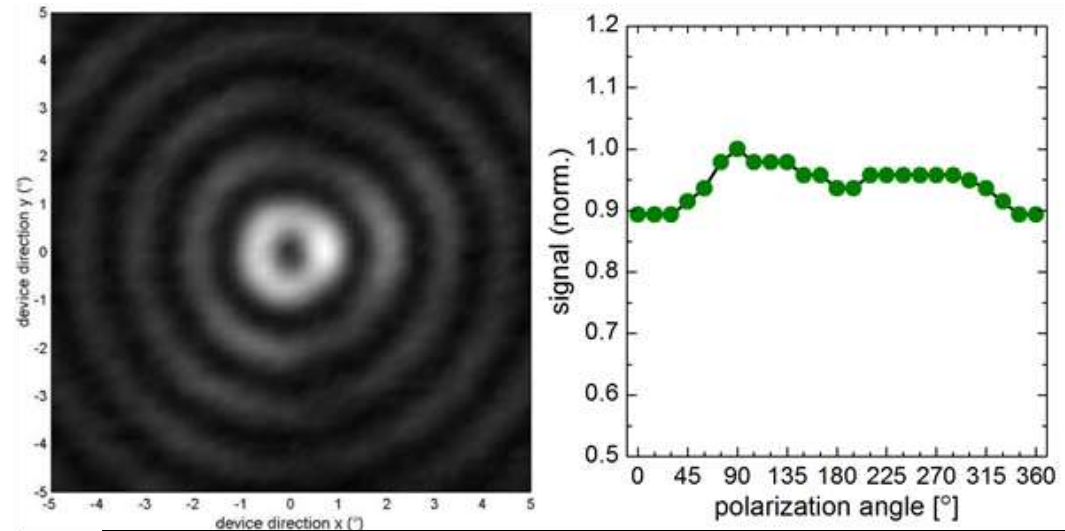
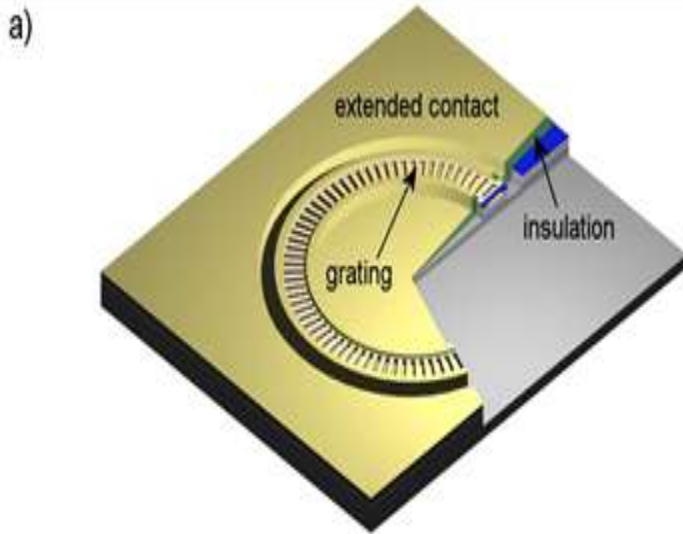
Bore Size	300 μm
Straight Losses	1 dB/m
Bending Losses	0,1 dB/m

Beam Profiling measurement setup and sample beam profiles

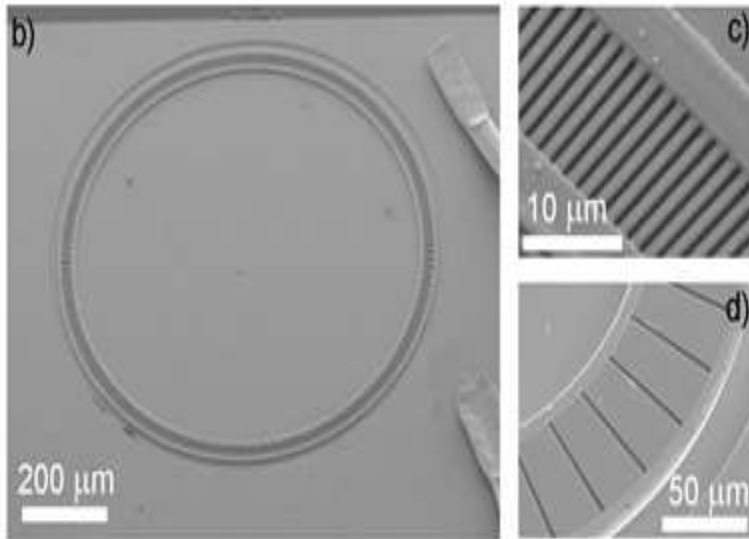
Future Directions and Outlook

- New target analytes such as carbonyl sulfide (OCS), formaldehyde (CH_2O), nitrous acid (HNO_2), hydrogen peroxide (H_2O_2), ethylene (C_2H_4), ozone (O_3), nitrate (NO_3), propane (C_3H_8), and benzene (C_6H_6)
- Ultra-compact, low cost, robust sensors (e.g. C_2H_6 , NO , CO)
- Monitoring of broadband absorbers: acetone ($\text{C}_3\text{H}_6\text{O}$), acetone peroxide (TATP), UF_6
- Optical power build-up cavity designs
- Development of trace gas sensor networks

Mid- IR and THz Ring Cavity Surface Emitting QCLs

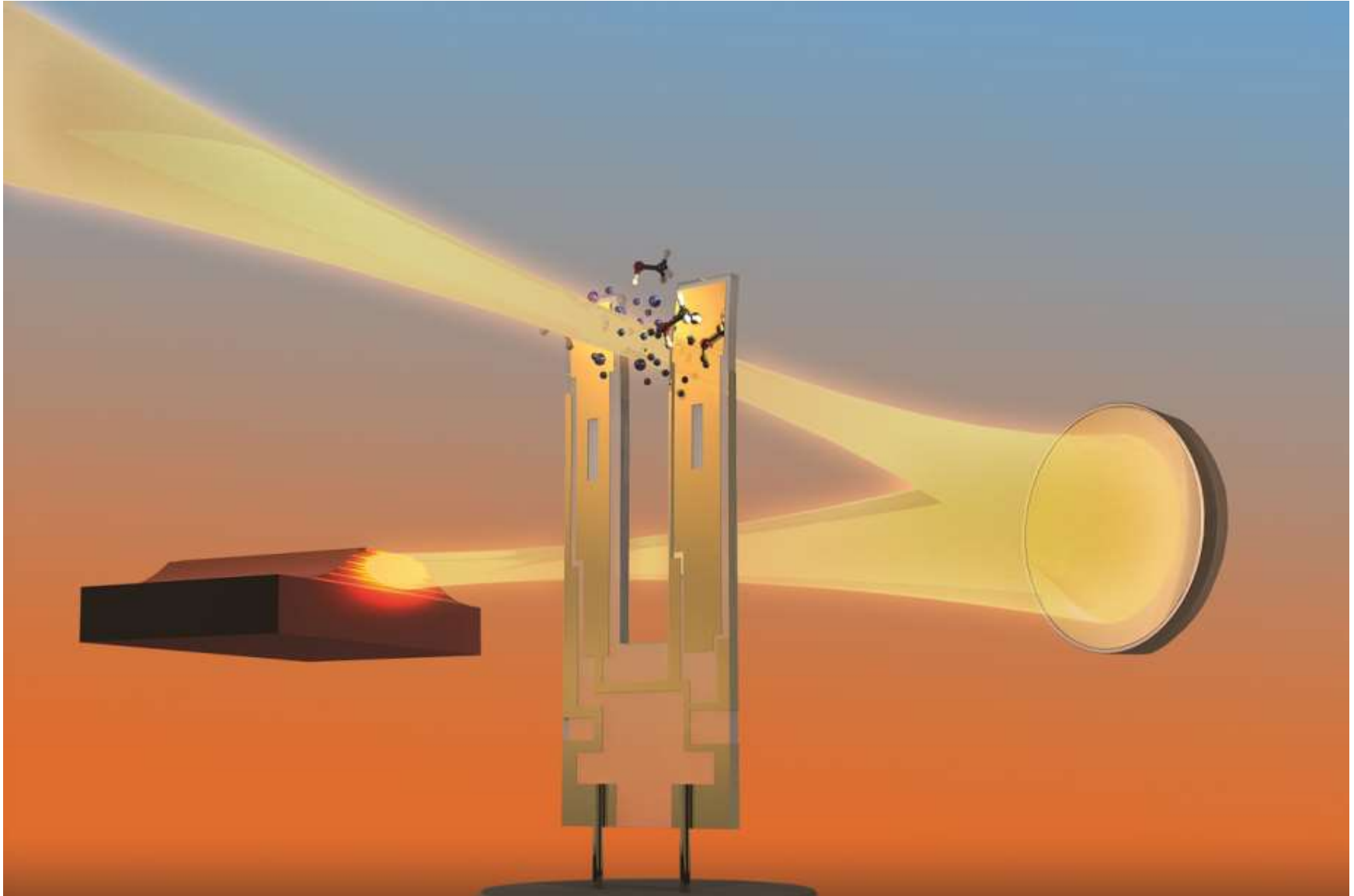


(a) Two-dimensional far-field plot emanating from a MIR RCSE-QCL and recorded with a micro-bolometer camera in a distance of 40 mm. (b) Polarization dependent intensity measurement.



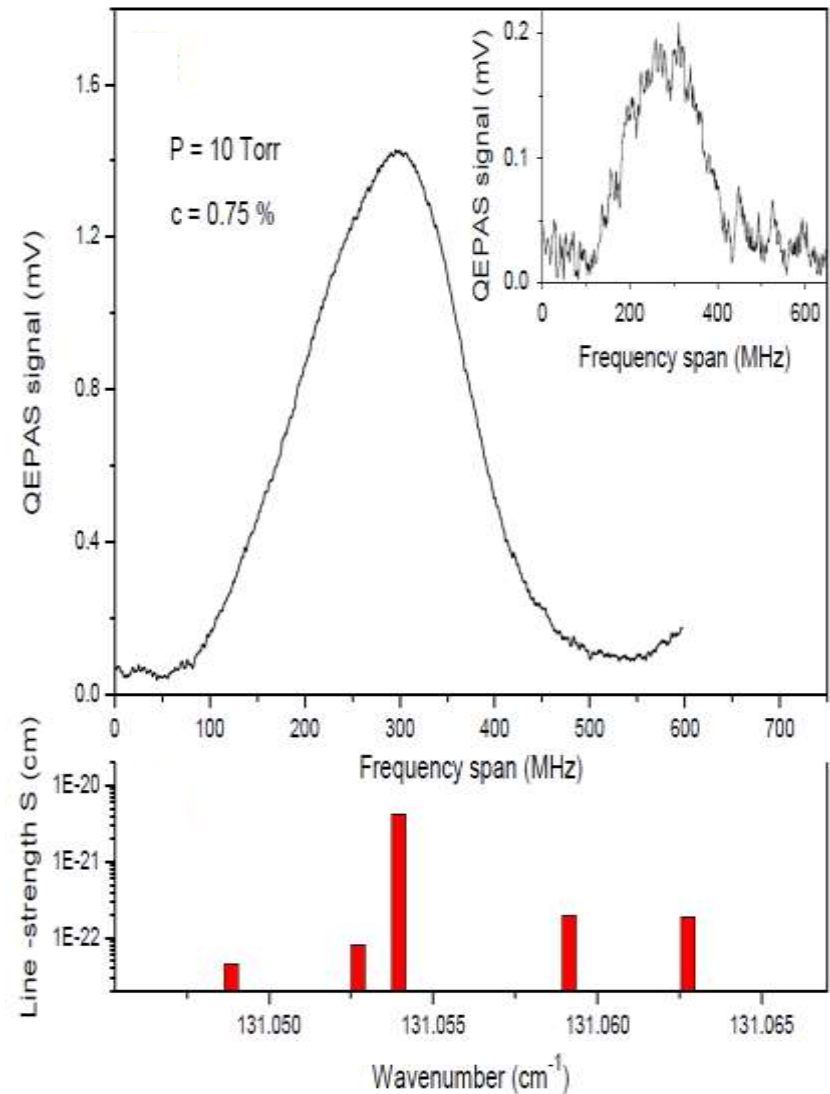
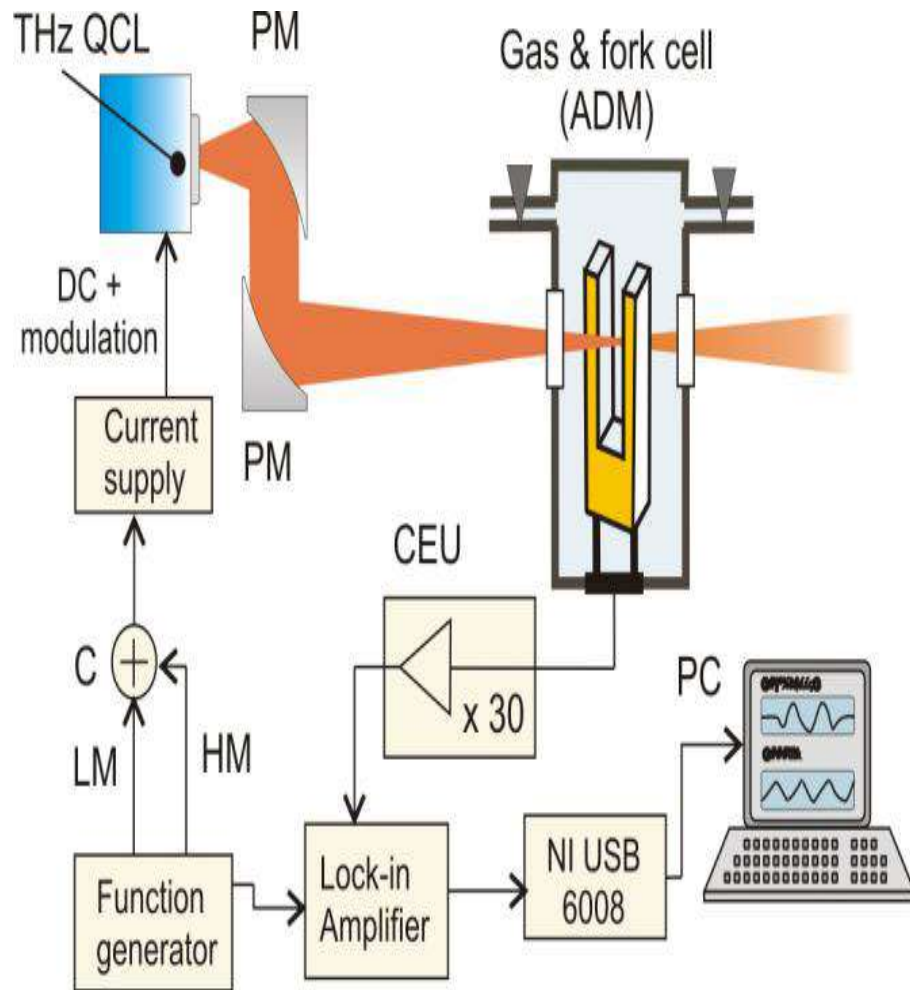
(a) Three-dimensional illustration of a ring cavity surface emitting laser. (b) Scanning electron microscopy image of a processed MIR device. Close-up of a (c) MIR and (d) a THz waveguide section holding second order gratings.

THz based QEPAS

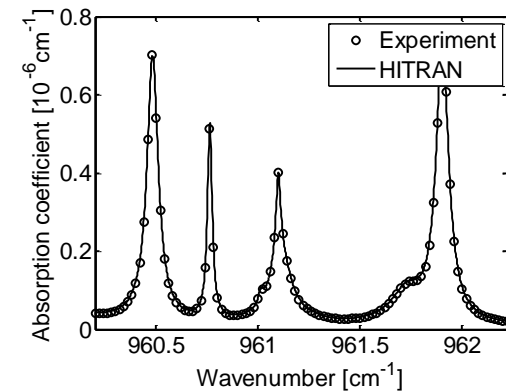
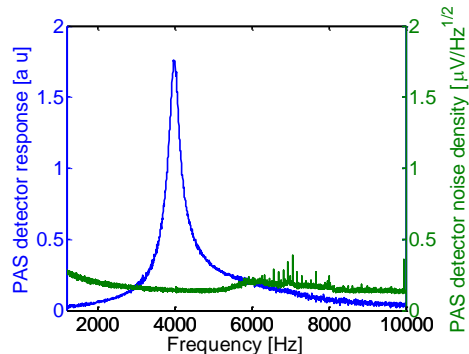
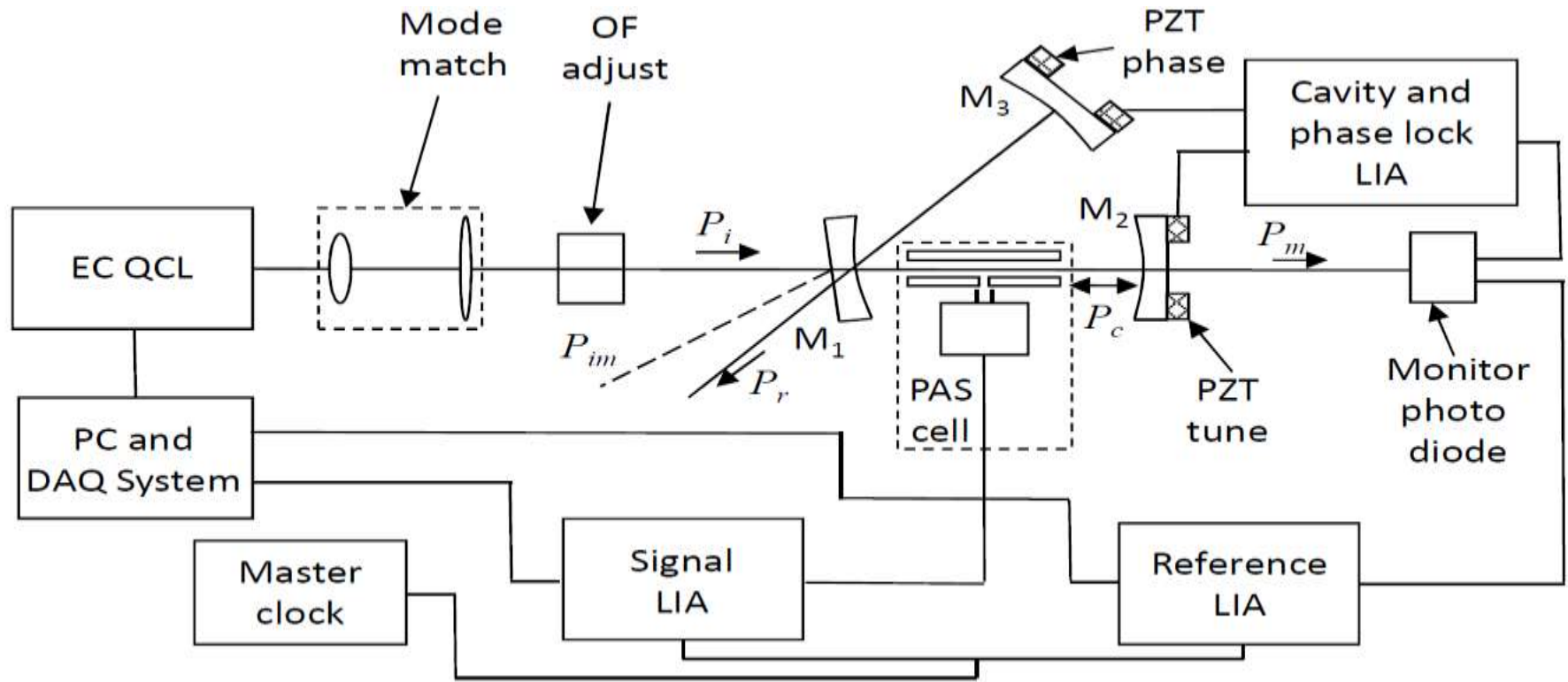


V. Spangolo et. al. , APL(appear), July (2013)

THz based QEPAS



Intracavity NH_3 based QEPAS



Summary

- Laser spectroscopy with a mid-infrared, room temperature, continuous wave, DFB laser diodes and high performance DFB QCL is a promising analytical approach for real time atmospheric measurements and breath analysis.
- Six infrared semiconductor lasers from Nanoplus, Daylight Solutions, Maxion Technologies (PSI), Hamamatsu, Northwestern University and AdtechOptics were used recently (2011-2012) by means of TDLAS, PAS and QEPAS
- Seven target trace gas species were detected with a 1 sec sampling time:
 - C_2H_6 at $\sim 3.36\ \mu\text{m}$ with a detection sensitivity of 130 pptv using TDLAS
 - NH_3 at $\sim 10.4\ \mu\text{m}$ with a detection sensitivity of ~ 1 ppbv (200 sec averaging time);
 - NO at $\sim 5.26\ \mu\text{m}$ with a detection limit of 3 ppbv
 - CO at $\sim 4.61\ \mu\text{m}$ with minimum detection limit of 2.5 ppbv
 - SO_2 at $\sim 7.24\ \mu\text{m}$ with a detection limit of 100 ppbv
 - CH_4 and N_2O at $\sim 7.28\ \mu\text{m}$ currently in progress with detection limits of 20 and 7 ppbv, respectively.
- New target analytes such as OCS, CH_2O , HONO, H_2O_2 , C_2H_4 ,
- Monitoring of broadband absorbers such as acetone, C_3H_8 , C_6H_6 and UF_6
- Compact, robust sensitive and selective single frequency, mid-infrared sensor technology that is capable of performing precise, accurate and autonomous concentration measurements of trace gases relevant in environmental, biomedical, industrial monitoring and national security.